



# Natural marine bromoform emissions in the fully coupled ocean atmosphere-model NorESM2

3

4 Dennis Booge<sup>1,2</sup>, Jerry F. Tjiputra<sup>3</sup>, Dirk J. L. Olivié<sup>4</sup>, Birgit Quack<sup>1</sup> and Kirstin Krüger<sup>2</sup>

5 <sup>1</sup>GEOMAR Helmholtz Centre for Ocean Research Kiel, Kiel, 24105, Germany

6 <sup>2</sup>Department of Geosciences, Section of Meteorology and Oceanography, University of Oslo, Oslo, 0371, Norway

7 <sup>3</sup>NORCE Norwegian Research Centre and Bjerknes Centre for Climate Research, Bergen, 5007, Norway

8 <sup>4</sup>Norwegian Meteorological Institute, Oslo, 0313, Norway

9 Correspondence to: Dennis Booge (dbooge@geomar.de) and Kirstin Krüger (kkrueger@geo.uio.no)

10 Abstract. Oceanic bromoform (CHBr<sub>3</sub>) is an important precursor of atmospheric bromine. Although highly relevant for 11 the future halogen burden and ozone layer in the stratosphere, the global CHBr3 production in the ocean and its emissions 12 are still poorly constrained in observations and are mostly neglected in climate models. Here, we newly implement marine 13 CHBr3 in the state-of-the-art Norwegian Earth System Model (NorESM2) with fully coupled ocean-sea-ice-atmosphere 14 biogeochemistry interactions. Our results are validated with oceanic and atmospheric observations from the HalOcAt 15 (Halocarbons in the Ocean and Atmosphere) data base. The simulated mean oceanic concentrations (6.61±3.43 pmol L <sup>1</sup>) are in good agreement with observations in open ocean regions (5.02±4.50 pmol L<sup>-1</sup>), while the mean atmospheric 16 17 mixing ratios (0.76±0.39 ppt) are lower than observed but within the range of uncertainty (1.45±1.11 ppt). The NorESM2 18 ocean emissions of CHBr<sub>3</sub> (214 Gg yr<sup>-1</sup>) are in the range of or higher than previously published estimates from bottomup approaches but lower than estimates from top-down approaches. Annual mean emissions are mostly positive (sea-to-19 20 air), driven by oceanic concentrations, sea surface temperature and wind speed, dependent on season and location. During 21 low-productivity winter seasons, model results imply some oceanic regions in high latitudes as sinks of atmospheric CHBr3, because of its elevated atmospheric mixing ratios. We further demonstrate that key drivers for the oceanic and 22 23 atmospheric CHBr<sub>3</sub> variability are spatially heterogeneous. In the tropical West Pacific, which is a hot spot for oceanic 24 bromine delivery to the stratosphere, wind speed is the main driver for CHBr3 emissions on annual basis. In the North 25 Atlantic as well as in the Southern Ocean region the atmospheric and oceanic CHBr<sub>3</sub> variabilities are interacting during 26 most of the seasons except for the winter months where sea surface temperature is the main driver. Our study provides 27 improved process understanding of the biogeochemical cycling of CHBr<sub>3</sub> and more reliable natural emission estimates especially on seasonal and spatial scales compared to previously published model estimates. 28

#### 29 1 Introduction

30 Bromoform (CHBr<sub>3</sub>) from the ocean is the most important organic compound for atmospheric bromine with an 31 atmospheric lifetime of ~2-4 weeks (Carpenter and Liss, 2000; Quack and Wallace, 2003; Salawitch, 2006; Papanastasiou 32 et al., 2014). As a reactive halogenated compound, it belongs to the very short-lived substances (VSLS) with lifetimes of less than 6 months in the atmosphere (Law et al., 2007). In the tropics, VSLSs are rapidly lifted to the stratosphere by 33 34 tropical deep convection (Sala et al., 2014; Navarro et al., 2015; Fuhlbrügge et al., 2016), where they contribute up to 35 ~25% to stratospheric bromine (Dorf et al., 2006 and following work). Bromine is ~60 times more efficient in depleting 36 lower stratospheric ozone than chlorine and significantly contributes to ozone depletion in the lower stratosphere (Daniel 37 et al., 1999; Sinnhuber et al., 2009; Montzka et al., 2011; Villamayor et al., 2023) with potential impacts on the radiation 38 budget of the atmosphere from -0.02 W m<sup>-2</sup> to -0.13 W m<sup>-2</sup> (Hossaini et al., 2015; Saiz-Lopez et al., 2023). 39 The oceanic air-sea gas exchange of CHBr<sub>3</sub> is parameterized based on the concentration gradient between surface water

40 and air and is related to wind speed and sea surface temperature via the transfer velocity (e.g. Nightingale et al., 2000).





41 Due to sparse measurements, marine CHBr<sub>3</sub> emission estimates are subject to large uncertainties (Laube et al., 2023). 42 CHBr3 emission inventories from "bottom-up" approaches (e.g. Quack and Wallace, 2003; Butler et al., 2007; Ziska et al., 2013; Lennartz et al., 2015; Stemmler et al., 2015; Fiehn et al., 2018) are based on in-situ oceanic data, whereas "top-43 down" approaches (e.g. Warwick et al., 2006; Liang et al., 2010; Ordóñez et al., 2012) use in situ atmospheric mixing 44 45 ratio measurements. Resulting CHBr<sub>3</sub> emissions span a large range between 150 and 820 Gg Br yr<sup>-1</sup> (Laube et al., 2023). 46 The different methods cover e.g., statistical extrapolation of measurement-based data (Ziska et al., 2013; and update in 47 Fiehn et al., 2018), scaling of emissions to chlorophyll-a satellite observations (Ordóñez et al., 2012), modelling 48 atmospheric CHBr<sub>3</sub> with a modular flux in a chemistry climate model (Lennartz et al., 2015), and a data-oriented machine-49 learning algorithm (Wang et al., 2019). These studies use limited spatial and temporal data coverage, underrepresenting 50 seasonal and interannual variations and spatial heterogeneity by averaging concentrations. 51 Oceanic CHBr3 is mainly linked to primary production through natural processes such as marine organisms like 52 macroalgae and phytoplankton (Gschwend et al., 1985; Carpenter and Liss, 2000; Quack et al., 2004). Elevated surface 53 water concentrations are observed in coastal and shelf waters especially including the eastern boundary upwelling systems 54 (EBUS) (Quack and Wallace, 2003). Laboratory culture studies of phytoplankton production rates by Tokarczyk and 55 Moore (1994) and Moore et al. (1996) reported CHBr<sub>3</sub> increase during the exponential growth phase of phytoplankton. Those specific growth rates and the corresponding temporal changes in CHBr3 concentrations were first applied in a 56 57 physical biogeochemical water column model for the tropical Atlantic (Hense and Quack, 2009), and later implemented 58 in the global biogeochemical HAMburg Ocean Carbon Cycle model (HAMOCC; Stemmler et al., 2015). Stemmler et al. 59 (2015) explicitly implemented sources and sinks of marine CHBr3 in the three-dimensional ocean biogeochemistry model 60 HAMOCC. However, they are not fully coupled with the atmosphere, and resulting emissions rely on fixed, prescribed, 61 extrapolated, observed atmospheric data of Ziska et al. (2013). Since the atmospheric concentrations are regulated by the 62 oceanic emissions, accurate estimates of atmospheric and oceanic CHBr3 variability require such coupling, which can be 63 achieved using an Earth System Model (ESM). 64 Here, we present the first global model simulation of CHBr<sub>3</sub> in the fully coupled Norwegian ESM (NorESM2), where 65 CHBr<sub>3</sub> production is prognostically related to the primary production in the ocean taking natural biological processes into account. We present results from a historical experiment focusing on the period 1990 to 2014 and compare them with 66

67 HalOcAt observations (https://halocat.geomar.de). Furthermore, we evaluate oceanic CHBr<sub>3</sub> excess and deficit regions

- and use multilinear regression analysis to identify drivers of oceanic and atmospheric CHBr<sub>3</sub>, as wells as CHBr<sub>3</sub> emission
- 69 variations on regional and temporal scales.

## 70 2 Model and Methods

71 We use the latest version of NorESM2 (NorESM2-LM; Seland et al., 2020; Tjiputra et al., 2020), which has participated 72 in the Coupled Model Intercomparison Project phase 6 (CMIP6) and contributed to the latest assessment report of the 73 IPCC-AR6 (Masson-Delmotte et al., 2023). The NorESM2 is a fully coupled ESM and is partly based on the Community 74 ESM Version 2 (Danabasoglu et al., 2020), which is developed by the National Center for Atmospheric Research (NCAR) 75 in the United States. NorESM2 is an updated version of its original version NorESM1 (Bentsen et al., 2013; Tjiputra et 76 al., 2013). It consists of a modified version of the Community Atmosphere Model version 6 (CAM6-Nor), the isopycnic 77 coordinate Bergen Layered Ocean Model (BLOM), the ocean biogeochemistry model isopycnic coordinate HAMOCC 78 (iHAMOCC), the sea ice model (Community Ice CodE version 5.1.2; CICE5.1.2), the Community Land Model version 79 5 (CLM5), and the river runoff model (MOdel for Scale Adaptive River Transport; MOSART). Both BLOM and iHAMOCC apply a tripolar grid with a horizontal resolution of ~1° and 53 vertical isopycnic layers, while CAM6-Nor 80 81 and CLM5 share a common horizontal resolution of  $\sim 2^{\circ}$  and 32 hybrid-pressure layers (lowest atmospheric layer





82 thickness: ~120 m) and a model top at 3.6 hPa (~40 km altitude). Here, we briefly highlight key features of iHAMOCC 83 as well as the CHBr3 implementation (Section 2.1). The iHAMOCC ocean biogeochemical module is based on the original work of Maier-Reimer (2012), has gone through several improvements and was later adapted to an isopycnic coordinate 84 ocean model (Assmann et al., 2010; Tjiputra et al., 2010). The model prognostically simulates inorganic carbon chemistry 85 following the standard Ocean Model Intercomparison Project (OMIP) protocol. It includes a Nutrient Phytoplankton 86 87 Zooplankton Detritus (NPZD) type ecosystem module, where the phytoplankton growth rate is constrained by multinutrient limitation as well as ambient light and temperature. Particulate organic matters produced in the euphotic zone is 88 89 exported to the interior with a sinking velocity that increases linearly with depth before it is remineralized back to 90 inorganic carbon. The NorESM2 is able to simulate the observed large-scale pattern of surface primary productivity as 91 well as the regional seasonal cycle (Tjiputra et al., 2020).

### 92 2.1 Bromoform module in NorESM2

#### 93 2.1.1 Oceanic bromoform

The marine CHBr<sub>3</sub> processes implemented in iHAMOCC comprise of advection (adv), production ( $\beta$ ), air-sea gas exchange (F), and three sink terms of: photolysis (UV), hydrolysis (H) and halogen substitution (S), as shown in Eq. (1). The production and photolysis occur in the euphotic layer (top 100 m depth) of the model, whereas the air-sea gas exchange is computed in the top-most layer of the ocean (upper 10 m). Advection and other sink terms are calculated throughout the water column. The change over time of the oceanic CHBr<sub>3</sub> concentration is modelled as:

$$\frac{[\text{CHBr}_3]}{dt} = adv(\text{CHBr}_3) + \beta - F - UV - H - S.$$
(1)

99

100 The parameterizations for the different processes are largely based on Stemmler et al. (2015). CHBr<sub>3</sub> is produced during

101 biological production as follows:

$$\beta = \beta_0 * \left( \frac{f_1 * Si(OH)_4}{K_{phy}^{Si(OH)_4} + Si(OH)_4} + \frac{f_2 * K_{phy}^{Si(OH)_4}}{K_{phy}^{Si(OH)_4} + Si(OH)_4} \right),$$
(2)

102

where diatom and non-diatom contributing factors,  $f_1$  and  $f_2$ , are set equally to 1. In contrast to Stemmler et al. (2015), the bulk CHBr<sub>3</sub> production ratio ( $\beta_0$ ) is modified and set to 2.4x10<sup>-6</sup> nmol CHBr<sub>3</sub> (mmol N)<sup>-1</sup>, based on Kurihara et al. (2012)

105 and Roy (2010).

106 The air-sea gas exchange is calculated as follows:

$$F = k_w * (C_w - \frac{C_a}{H_{bromo}}), \tag{3}$$

107

where  $C_w$  and  $C_a$  are CHBr<sub>3</sub> concentrations in the surface ocean and CHBr<sub>3</sub> mixing ratios in the atmosphere, respectively. Positive emissions are defined as outgassing to the atmosphere. The temperature-dependent dimensionless Henry's law solubility constant ( $H_{bromo}$ ) is defined in Moore et al. (1995):

$$H_{bromo} = e^{13.16 - \frac{4973}{SST}},\tag{4}$$

111

112 with SST the sea-surface temperature in Kelvin.  $k_w$  represents the gas transfer velocity calculated following Nightingale

113 et al. (2000) using the 10 m surface wind speed (*u*):

$$k_w = (0.222u^2 + 0.33u) * (\frac{660}{Sc_{bromo}})^{0.5}.$$
(5)





- 114
- 115 The Schmidt number ( $Sc_{bromo}$ ) for CHBr<sub>3</sub> is defined in Quack and Wallace (2003) using the sea surface temperature *SST* 116 in °C:

$$Sc_{bromo} = 4662.8 - 319.45 * SST + 9.9012 * SST^2 - 0.1159 * SST^3.$$
(6)

117 118

The loss term due to photolysis is computed as follows:

$$UV = I_{UV} * \frac{I_0}{I_{ref}} * e^{(-a_W * z)} * [CHBr_3],$$
(7)

119

- 120 where the decay time scale  $(I_{UV})^{-1}$  is set to 30 days (Carpenter and Liss, 2000).  $I_0$  and  $I_{ref}$  are the prognostic incoming UV
- 121 radiation (i.e., 30% of shortwave radiation) and annual average irradiance at the surface layer, respectively. z is the depth
- 122 and  $a_w$  is the attenuation coefficient of UV radiation, set to 0.33 m<sup>-1</sup>.
- 123 The loss term related to hydrolysis is estimated following Stemmler et al. (2015):

$$H = A_1 * e^{\left(-\frac{E_A}{RT}\right)} * [OH^-] * [CHBr_3],$$
(8)

124 125

126 127

with $A_1$ , $E_A$ , and $R$ set to $1.23 \times 10^{17}$ L mol <sup>-1</sup>	min <sup>-1</sup> , 107 300 J mol <sup>-1</sup> , and 8.314 J K <sup>-1</sup> mol <sup>-1</sup> , respectively (Washington, 1995).
T is the seawater temperature in Kelvin.	

128 Degradation due to halogen substitution (Eq. 5-6 of Stemmler et al., 2015):

$$S = L_{ref} * e^{\left(A_2 * \left(\frac{1}{T_{ref}} - \frac{1}{T}\right)\right)} * [CHBr_3], \tag{9}$$

129

130 with  $L_{ref}$  and  $A_2$  set to 7.33x10<sup>-10</sup> s<sup>-1</sup> and 12507.13 K, respectively, and  $T_{ref} = 298$  K.

## 131 2.1.2 Atmospheric CHBr<sub>3</sub>

132 CHBr<sub>3</sub> is implemented as a 3-dimensional tracer in the atmospheric model and is transported by the large-scale

atmospheric circulation and sub-grid scale processes (shallow and deep convection, and boundary layer turbulence). It is
 removed in the atmosphere by photolysis:

$$CHBr_3 + hv \rightarrow 3 Br, \qquad (10)$$

135

136 and by reaction with the OH radical:

$$CHBr_3 + OH \rightarrow 3 Br.$$
(11)

137

138 The reaction rate k [cm<sup>3</sup> molecules<sup>-1</sup> s<sup>-1</sup>] for the removal of CHBr<sub>3</sub> by OH in

$$\frac{d[\text{CHBr}_3]}{dt} = -k * [\text{CHBr}_3] * [\text{OH}]$$
(12)

139

140 is defined as follows:

$$k = 9.0 * 10^{-13} \exp\left(-\frac{360}{T}\right),\tag{13}$$

141





142 with *T* the ambient temperature in Kelvin, and  $[CHBr_3]$  and [OH] in molecules cm<sup>-3</sup>. The loss rate of CHBr<sub>3</sub> by photolysis 143 can be expressed by

$$\frac{d[CHBr_3]}{dt} = -I \ [CHBr_3], \tag{14}$$

144

- 145 where *I* [s<sup>-1</sup>] depends on the intensity of solar radiation and photo-physical properties of CHBr<sub>3</sub>. The OH concentration is
- 146 a monthly-varying climatology obtained from a Whole Atmosphere Community Climate Model (WACCM) historical
- 147 simulation with full tropospheric and stratospheric chemistry (Gettelman et al., 2019).
- $148 \qquad CHBr_3 \text{ in the atmosphere has no other sinks than reaction with OH (annual mean CHBr_3 lifetime: ~46 days) and photolysis$
- 149 (CHBr<sub>3</sub> lifetime: ~23 days) and is not affected by dry or wet deposition.

## 150 2.2 Model setup

151 A historical transient model run from 1850-2014, based on the CMIP6 protocol, was performed following a 500-year 152 preindustrial spin-up. The coupling of CHBr3 between the ocean and the atmosphere is carried out with an hourly time 153 frequency exchanging the air-sea gas transfer. For analysis of the model climatology as well as for analysis of the model 154 validation with observations and further analysis of the driving CHBr3 factors, daily model output data was averaged over 155 a period of 25 years (1990-2014) resulting in one mean value for each day of the year. The standard deviation of each 156 day reflects the variability within this time period. The 1990 to 2014 interval was chosen as most of the observations for 157 the model validation are within that time period, as compiled in the HalOcAt database (https://halocat.geomar.de, last access: 13.10.2023). 158

## 159 2.3 Observations: HalOcAt database

The HalOcAt database, compiled by Ziska et al. (2013), updated by Fiehn et al. (2018) and by this study, is an observationbased database for global oceanic and atmospheric data of short-lived halogenated compounds, such as CHBr<sub>3</sub>. To date, there are 9369 oceanic and 65179 atmospheric CHBr<sub>3</sub> measurements listed in 68 oceanic and 156 atmospheric datasets (campaigns), respectively. The following criteria were applied to the observations in order to be used for model validation:

- Sampling locations with an ocean bottom depth less than 200 m or closer than 100 km to land were excluded.
- Sampling depth of oceanic CHBr<sub>3</sub> measurements had to be within the first 10 m of the water column in order to
   be comparable to the CHBr<sub>3</sub> output of the upper surface ocean model layer (10 m depth).
- 167 Maximum sampling height of atmospheric CHBr<sub>3</sub> measurements was set to 30 m altitude.
- Wherever applicable, individual measurements throughout one day were averaged to result in a daily averaged
   surface ocean concentration or atmospheric mixing ratio in order to consider the same temporal resolution as the
   daily model output. The coordinates of the respective averaged data points throughout a day were also equally
- 171 averaged. These locations were used to compare the observation with the closest grid point of the model output.





172



Figure 1: Locations of oceanic (red, n=666) and atmospheric (blue, n=697) daily mean CHBr<sub>3</sub> observations from the HalOcAt database used to compare to daily mean NorESM2 model output.

173 After screening the HalOcAt data base with the above-mentioned criteria, the individual oceanic and atmospheric datasets

174 (including the remaining datapoints) were tested for outliers. The mean from each dataset was calculated and the group

175 of all average values was tested for outliers. An outlier was defined as an element with more than three standard deviations

176 from the mean. According to the outlier test for oceanic and atmospheric datasets the corresponding dataset was removed

177 and not used for further validation of the model.

178 By addressing the mentioned criteria and datasets, we were able to validate the model with 666 daily mean oceanic (5154

179 individual) and 697 daily mean atmospheric (8411 individual) CHBr<sub>3</sub> observations from the HalOcAt database covering

180 both hemispheres (northern hemisphere (NH): 61%, southern hemisphere (SH): 39%), from the tropics (0-20°N/°S; 36%)

to the polar regions (60-90°N/S; 18%) with most observations in or above the Atlantic Ocean (44%) (Figure 1).

## 182 2.4 Bromoform excess/deficit calculation

The CHBr<sub>3</sub> excess/deficit (balance) rate ( $k_{bal}$ , Eq. (15), pmol m<sup>-2</sup> h<sup>-1</sup>) is the difference between the CHBr<sub>3</sub> production rate and the sum of different CHBr<sub>3</sub> loss rates, with all rates integrated over the upper 100 m depth):

$$k_{bal} = \sum production \ rate - \sum loss \ rate = k_{\beta} - (k_{UV} + k_F).$$
(15)

185

The production term is described as the biological oceanic CHBr<sub>3</sub> production rate ( $k_{\beta}$ , Eq. (2)) and the loss term includes the two fastest loss processes, i.e., photolysis due to UV radiation ( $k_{UV}$ , Eq. (7)) and the loss to the atmosphere via air-sea gas exchange ( $k_F$ , Eq. (3)). We define a positive  $k_{bal}$  as CHBr<sub>3</sub> excess rate and a negative  $k_{bal}$  as CHBr<sub>3</sub> deficit rate. The loss terms related to hydrolysis and to halogen substitution are not included as they are several orders of magnitude smaller than  $k_{UV}$  and  $k_F$ , in the surface ocean.

## 191 **2.5 Bromoform driving factor calculation**

Different parameters impact the variations of oceanic and atmospheric CHBr<sub>3</sub> values and influence the air-sea gas exchange. These impacts can vary in magnitude and frequency dependent on local and seasonal conditions. Daily mean average model output values from 1990-2014 were used to calculate annual and seasonally resolved (DJF, MAM, JJA, SON) driving factors for oceanic CHBr<sub>3</sub> concentrations (*Bromo<sub>oce</sub>*), atmospheric CHBr<sub>3</sub> mixing ratios (*Bromo<sub>air</sub>*) and CHBr<sub>3</sub> emissions (*Bromo<sub>flux</sub>*) in three different specific areas (North Atlantic, tropical West Pacific, Southern Ocean),





which are presented in Section 3.4. Driving factors for each area, parameter and season were derived using multilinearregression (MLR) analyses.

In order to compare parameters with different magnitudes, input data of each parameter was standardized prior to MLR analysis by centering to have a mean of 0 and scaled to have a standard deviation of 1. Input data for each parameter consisted of daily mean averages over the specific area, providing 365 values as basis for annually and  $\sim$  90 values for seasonally resolved MLR. Equations for MLR calculations were as follows with coefficients *a*, *b*, *c*, *d*, *e*, *f*, CHBr<sub>3</sub> production (*Bromo<sub>prod</sub>*), *Bromo<sub>oce</sub>*, *Bromo<sub>air</sub>* and *Bromo<sub>flux</sub>*, as well as the 10 m surface wind speed (*wind*) and sea surface temperature (*SST*):

$$Bromo_{oce} = a * SST + b * wind + c * Bromo_{prod} + d * Bromo_{flux} + e * Bromo_{air} + f$$
(16)

205

$$Bromo_{flux} = a * SST + b * wind + c * Bromo_{prod} + d * Bromo_{oce} + e * Bromo_{air} + f$$
(17)

206

$$Bromo_{air} = a * SST + b * wind + c * Bromo_{prod} + d * Bromo_{flux} + e * Bromo_{oce} + f$$
(18)

207

Other oceanic CHBr<sub>3</sub> loss processes (e.g. photolysis) were neglected in these calculations as the loss due to gas exchange is ~70 times higher than the loss due to photolysis (data not shown). If the highest resulting coefficient for each season and MLR is significantly higher than all other coefficients, the corresponding parameter is presented as the main driver for either  $Bromo_{oce}$ ,  $Bromo_{air}$  or  $Bromo_{flux}$ . If the highest resulting coefficient is not significantly different from the second or third highest coefficient, more than one coefficient and corresponding parameters are presented as main drivers. Table 1 lists the annual mean coefficients, Table S1 lists the seasonally resolved main drivers.

#### 214 **3** Results and Discussion

#### 215 3.1 Model climatology

216 The annual and seasonal CHBr<sub>3</sub> oceanic concentrations, atmospheric mixing ratios and emissions reveal significant 217 spatial variations (Figure 2). The annual global average surface CHBr<sub>3</sub> concentrations are 5.04 pmol L<sup>-1</sup> (DJF: 218 5.36 pmol L<sup>-1</sup>, JJA: 4.86 pmol L<sup>-1</sup>) with highest annual mean concentrations of 28.37 pmol L<sup>-1</sup> in the upwelling region off 219 the coast of Peru and lowest annual mean concentrations of 1.37 pmol L<sup>-1</sup> in the Gulf of Boothia (71°N, 91°W) north of 220 Canada. The areas with the lowest oceanic CHBr3 concentrations are the central parts of the North and the South Pacific 221 Gyres. Concentrations of surface ocean CHBr<sub>3</sub> in the entire NH (JJA: 5.9 pmol  $L^{-1}$  > DJF: 4.3 pmol  $L^{-1}$ ) and SH (DJF: 2.2.2 6.1 pmol  $L^{-1}$  > JJA: 4.1 pmol  $L^{-1}$ ) are generally higher during the respective summer than during the respective winter 223 season. These distinct differences of oceanic CHBr<sub>3</sub> concentrations are also due to the higher biological production in 224 summer (NH: 335 pmol  $m^{-2} h^{-1}$ ; SH: 371 pmol  $m^{-2} h^{-1}$ ) than in winter (NH: 235 pmol  $m^{-2} h^{-1}$ ; SH: 173 pmol  $m^{-2} h^{-1}$ ) as 225 shown in Fig. S3. The direct link of CHBr3 to the biological production applies to the low oceanic CHBr3 concentrations 226 in the North and South Pacific Gyres and to the high oceanic concentrations in the areas of the EBUS. 227 Variations in annual mean atmospheric CHBr3 mixing ratios mainly follow the surface ocean concentrations with highest 228 mixing ratios in the tropics, especially in the EBUS. Global annual average mixing ratios over the ocean are 0.67 ppt

(DJF: 0.70 ppt, JJA: 0.69 ppt) with highest annual mean mixing ratios of 2.21 ppt in the south-eastern Pacific upwelling region off the coast of Peru and lowest annual mean mixing ratios of 0.13 ppt over the Persian Gulf. On a global average, the variability of atmospheric mixing ratios is lower than the variability of CHBr<sub>3</sub> concentrations in the surface ocean (Figure 2). During austral winter (JJA), mostly dark and cold conditions increase the lifetime of atmospheric CHBr<sub>3</sub>,

233 which leads to a uniform mixing ratio (0.67±0.05 ppt) over the entire Southern Ocean. Similar to oceanic CHBr<sub>3</sub>





concentrations, central parts of the North and South Pacific Gyre have low atmospheric CHBr<sub>3</sub> mixing ratios ( $0.46\pm0.05$  ppt). During austral summer (DJF) atmospheric mixing ratios increase further as strong biological activity increases surface ocean concentrations, which enhance the oceanic emissions.

237 Generally, supersaturation of CHBr<sub>3</sub> in the world's ocean leads to emissions from the ocean to the atmosphere (defined as positive). Global annual mean emissions are 268 pmol m<sup>-2</sup> h<sup>-1</sup> (DJF: 294 pmol m<sup>-2</sup> h<sup>-1</sup>, JJA: 253 pmol m<sup>-2</sup> h<sup>-1</sup>) with 238 239 highest annual mean emissions of 953 pmol  $m^{-2} h^{-1}$  in the upwelling region off the coast of Peru. In the tropical regions, 240 annual mean emissions of 427 pmol m<sup>-2</sup> h<sup>-1</sup> between 10°N and 10°S, add to atmospheric entrainment of oceanic CHBr<sub>3</sub> up into the stratosphere (Fiehn et al., 2018; Tegtmeier et al., 2020). Lowest annual mean emissions of -1 pmol m<sup>-2</sup> h<sup>-1</sup> are 241 242 modelled under ice free conditions in the Gulf of Boothia (71°N, 91°W) north of Canada (white regions in Figure 2) with 243 very low oceanic CHBr<sub>3</sub> production and low seawater temperatures. However, the atmospheric mixing ratios are 244 comparably high under these conditions. These conditions favour negative emissions, which, according to the results of 245 our fully coupled ESM, can be seen in the Arctic and Antarctic during winter season, confirming the results by Stemmler 246 et al. (2015) and Ziska et al. (2013) although with a lower amount.

247 Generally, the modelled CHBr<sub>3</sub> emissions are high, where the ocean concentration is high and the elevated emissions lead 248 to elevated atmospheric mixing ratios. However, due to oceanic transport processes, locations of high oceanic CHBr<sub>3</sub> 249 emissions do not always coincide with locations of high oceanic CHBr3 production (compare Figure 2 and Fig. S3). In 250 the northern part of the Bay of Bengal (>18°N) e.g., ocean concentrations during DJF are very high (average: 251 21.64 pmol  $L^{-1}$ ), while the emissions are not as high compared to other ocean regions, due to low wind speeds. This leads 252 to a lower atmospheric mixing ratio than expected from the oceanic concentrations and shows that oceanic CHBr<sub>3</sub> 253 concentrations and emissions as well as atmospheric mixing ratios show regionally different interdependencies, which is 254 addressed in detail in Section 3.4.

255



Figure 2: Simulated annual (left), DJF (middle) and JJA (right) mean oceanic surface CHBr<sub>3</sub> concentrations (upper panel), atmospheric mixing ratios (middle panel) and CHBr<sub>3</sub> emissions (lower panel) for the period 1990-2014.

#### 256 **3.2 Model validation with observations**

The annual median surface oceanic CHBr<sub>3</sub> concentration (Figure 3a) from the 666 daily mean observations is 3.65 pmol L<sup>-1</sup> with a 25<sup>th</sup> and 75<sup>th</sup> percentile of 2.19 and 6.16 pmol L<sup>-1</sup>, respectively (min: 0.05 pmol L<sup>-1</sup>, max: 28.21 pmol L<sup>-1</sup>, mean: 5.02 pmol L<sup>-1</sup>). The global annual median surface oceanic CHBr<sub>3</sub> concentration from the model using only locations

corresponding with an existing observation is 6.00 pmol L<sup>-1</sup> with a 25<sup>th</sup> and 75<sup>th</sup> percentile of 4.23 and 8.10 pmol L<sup>-1</sup>,





respectively (min: 1.39 pmol  $L^{-1}$ , max: 24.25 pmol  $L^{-1}$ , mean: 6.61 pmol  $L^{-1}$ ). These results indicate that the model values are in the range with observed concentrations of oceanic CHBr<sub>3</sub>. While the median concentration of the model is higher than the median of the observations, all validated model data points fall within the full range of the observations. The model data cover a grid of ~100 km resolution, which leads to a smoothing of the values, whereas observational data is local daily mean point data.

266 The median CHBr<sub>3</sub> atmospheric mixing ratio (Figure 3b) from the 697 daily mean observations is 1.17 ppt with a 25<sup>th</sup> 267 and 75th percentile of 0.78 and 1.71 ppt, respectively (min: 0.03 ppt, max: 9.80 ppt). The global median atmospheric mixing ratio of CHBr<sub>3</sub> from the model at locations with observations is 0.69 ppt with a 25<sup>th</sup> and 75<sup>th</sup> percentile of 0.49 268 269 and 0.90 ppt, respectively (min: 0.22 ppt, max: 2.70 ppt). This comparison shows that the observed atmospheric mixing 270 ratios of CHBr<sub>3</sub> are in the same magnitude but generally higher than those from the model output. While our model 271 experiment focuses on natural CHBr3 production by phytoplankton, other sources as coastal macroalgae (Carpenter and 272 Liss, 2000) and anthropogenic sources, such as power plant cooling (Maas et al., 2021) or desalination plants (Agus et 273 al., 2009), may explain parts of the higher global annual median observational data of 41%. Jia et al. (2023) calculated

oceanic CHBr<sub>3</sub> [pmol L<sup>-1</sup>] atmospheric CHBr<sub>3</sub> [ppt] 30 10 HalOcAt NorESM HalOcAt NorESM 25 20 15 3.5 10 3 2.5 2 1.5 ŧ 5 0.5 0 annua DJF JJA annual DJF JJA annua annual n=666 n=697 20 [pmol L<sup>-1</sup>] D.JF DJF n=109 atm [ppt] n=119 20 ocean CHBr. CHBr. JJA JJA n=190 HalOcAt NorESM HalOcAt NorESM n=146 -60 -40 -60 Latitude [°N] Latitude [°N]

Figure 3: Boxplot comparison of NorESM2 model results with HalOcAt observations for oceanic (a) and atmospheric CHBr<sub>3</sub> (b), c: zonal mean oceanic CHBr<sub>3</sub> comparison annually and in DJF and JJA, d: zonal mean atmospheric CHBr<sub>3</sub> comparison annually and in DJF and JJA, d: zonal mean atmospheric CHBr<sub>3</sub> comparison annually and in DJF and JJA, d: zonal bin averages. Boxplots (left) have a break in the y axis to increase readability of the figure. The line inside the box represents the median value, the circle the mean value, the boxes show the first to third quartile, and the whiskers illustrate the highest and lowest values that are not outliers. The plus signs represent outliers.

- an increase of global CHBr<sub>3</sub> emissions of 31.5% when including anthropogenic emissions, which partly explains also the
- 275 lower observed atmospheric mixing ratios in the model compared to the observations.
- 276 Figure 3 also shows a more detailed comparison between observations and model data in 5° meridional binned averages
- 277 (shaded areas) for oceanic (Figure 3c) and atmospheric (Figure 3d) CHBr<sub>3</sub> on annual basis as well as in JJA and DJF.
- 278 The modelled data compare well with observations of oceanic CHBr<sub>3</sub> (Figure 3c) on annual basis over the 5° latitudinal





bins. In the HalOcAt database, there are no oceanic and atmospheric observations available north of  $50^{\circ}$ N and south of 30°S during boreal (DJF) and austral winter (JJA), respectively, which highlights the need of model data to entirely describe spatially and temporally resolved CHBr<sub>3</sub> (see also Fiehn et al., 2018). During DJF, the model overestimates the measured concentrations between 20°N and 5°S. During JJA, averaged model concentrations in the NH (10°N – 60°N) are slightly higher than the averaged observations. These discrepancies could indicate a missing process understanding, revealing lower oceanic production or additional loss processes.

285 With all data available, the 5° latitudinal averaged atmospheric CHBr<sub>3</sub> observations have a large spread in the tropics 286 resulting in a high standard deviation (Figure 3d). The model results in this region are uniform with a much lower standard 287 deviation. During boreal winter (DJF) atmospheric CHBr3 observations and model results show a good agreement, with an exception at 40-50°S. In this latitude range, observational atmospheric CHBr<sub>3</sub> mixing ratios (>3 ppt, Figure 3d) were 288 289 recorded between 24° and 60° W in the South Atlantic in 2007 (Gebhardt, 2008). Gebhardt (2008) reports enhanced 290 biological production in the Argentinian shelf-break zone (55°-60° W) with elevated chlorophyll-a concentration up to 291 4.5  $\mu$ g L<sup>-1</sup>. These values suggest also a high production of CHBr<sub>3</sub> and subsequent high emissions to the atmosphere. The 292 prevailing westerly winds, transported the CHBr3 enriched air masses eastward to the remote South Atlantic region in 293 2007, while in the model lower biological production entails lower atmospheric mixing ratios compared to the 294 observations. During boreal summer (JJA) very good agreement between atmospheric observations and model results is 295 obtained between 10°N and 60°N. North of 60°N, the model underestimates the measured atmospheric mixing ratios in 296 the polar region. Local meteorological and biological conditions (e.g. high wind speed, distinct phytoplankton blooms) 297 are averaged by the model to a resolution of ~100 km. Averaging data over time or space leads to lower values (e.g. gas emissions, Bates and Merlivat, 2001), which explains lower modelled atmospheric mixing ratios compared to the 298 299 observations. These local and short term temporal variations contribute to the discrepancy in atmospheric values at global 300 scale as well as potential anthropogenic CHBr<sub>3</sub> emissions (Jia et al., 2023). Furthermore, discrepancies between model 301 results and observations also point to missing process understanding, which helps to improve our understanding of the 302 biogeochemical cycling of CHBr3.

#### 303 **3.3 Excess and deficit regions of oceanic bromoform**

304 In most of the world's surface oceans CHBr<sub>3</sub> production and loss rates are balanced on an annual average with a k<sub>bal</sub> close 305 to zero (e.g. North and South Pacific, top panel of Figure 4). The equator region experiences a strong excess rate (positive 306  $k_{bal}$ ) on annual average with values up to 300 pmol m<sup>-2</sup> h<sup>-1</sup> showing higher CHBr<sub>3</sub> production than loss of CHBr<sub>3</sub> in the 307 upper ocean, caused by strong primary production (Fig. S3) in the equatorial upwelling. Surface currents transport the 308 CHBr<sub>3</sub> enriched surface water masses away from the equator, while experiencing loss of CHBr<sub>3</sub> to the atmosphere. 309 Therefore, adjacent marine areas north and south of the equator experience a deficit rate (negative  $k_{bal}$ ) of CHBr<sub>3</sub> (blue 310 areas, Figure 4), as no production balances the loss. The seasonality of  $k_{bal}$  is pronounced in the extratropics (bottom 311 panels of Figure 4). In these regions, a CHBr3 excess rate is observed mainly during summer and a CHBr3 deficit rate 312 mainly during winter in the respective hemispheres. A high  $k_{\beta}$  (elevated biological production) and a low  $k_F$  (weak 313 emissions to the atmosphere) caused by lower winds during summer, lead to a higher CHBr3 surface ocean concentration 314 in summer compared to winter time (Figure 2). During winter in both hemispheres, lower biological activity (low  $k_{\beta}$ ) and elevated wind speed (high  $k_F$ ) decrease the CHBr<sub>3</sub> production and increase the emissions to the atmosphere, which leads 315 to a CHBr<sub>3</sub> deficit rate. These results reveal seasonal as well as spatial differences in parameters (driving factors), which 316 317 influence CHBr3 concentrations in the world's ocean.







60°W 0° 60°E 120°E 180° 120°W 60°W 0° 60°E 120°E 180° 120°W

Figure 4: Mean CHBr<sub>3</sub> excess/deficit rates on annual (top) and seasonal (DJF: bottom left; JJA: bottom right) basis. Three rectangles in the top figure illustrate locations of case studies. NA: North Atlantic; TWP: Tropical West Pacific; SO: Southern Ocean.

- 318 In the following subsection, we selected three different case study areas, indicated in Figure 4, in order to contrast the
- 319 driving factors of the variations of oceanic and atmospheric CHBr<sub>3</sub> on regional and temporal scales:
- 320 North Atlantic, with an annual mean CHBr<sub>3</sub> deficit rate ( $k_{bal} = -33$  pmol m<sup>-2</sup> h<sup>-1</sup>)
- 321 Tropical West Pacific, with an annual mean CHBr<sub>3</sub> excess rate  $(k_{bal} = +32 \text{ pmol m}^{-2} \text{ h}^{-1})$
- 322 Southern Ocean, with negative emissions during the respective winter season  $(k_{bal} = +15 \text{ pmol m}^{-2} \text{ h}^{-1})$

## 323 **3.4** Driving factors of bromoform on regional and temporal scales

- 324 This section investigates the seasonal changes of oceanic and atmospheric CHBr3 and other parameters in three
- $325 \qquad \text{contrasting regions. Daily means of oceanic CHBr_3 concentrations, production, emissions, balance (as defined in Eq. \\$
- 326 (15)), atmospheric mixing ratios as well as SST and wind speed, in the North Atlantic, tropical West Pacific and Southern
- 327 Ocean over an entire year reveal large differences between the regions (Figure 5). With MLR analysis, the main driving
- 328 factors of oceanic and atmospheric CHBr<sub>3</sub> variability and its emissions in each region and season are investigated.





329



Figure 5: Seasonal changes of oceanic and atmospheric CHBr<sub>3</sub> (upper panel, a-c), CHBr<sub>3</sub> production, emissions and balance (middle panel, d-f), SST and wind speed (bottom panel, g-i), in the North Atlantic (left), tropical west Pacific (middle) and Southern Ocean (right). Shaded area is one standard deviation of the average value in the corresponding area. Note: y-limits for SST are not similar in between the three regions in order to increase readability of the figure.

330

#### 331 North Atlantic

The North Atlantic region (50°N - 60°N, 15°W - 35°W) is characterized by a strong seasonal cycle of both oceanic 332 333 CHBr3 concentrations and atmospheric mixing ratios (Figure 5a). The magnitude of the cycle is strongest among the three 334 investigated regions in this study (compare with Figure 5b,c). Oceanic CHBr3 concentrations are on average 3.64 pmol L-335 <sup>1</sup> with a minimum concentration of 1.87 pmol L<sup>-1</sup> during end of March and a maximum concentration of 6.93 pmol L<sup>-1</sup> 336 during July. Atmospheric mixing ratios show a similar seasonal cycle, shifted by one month, with average values of 337 0.60 ppt, a minimum mixing ratio of 0.30 ppt during April and a maximum mixing ratio of 1.12 ppt during August. Figure 338 5d shows, that the CHBr<sub>3</sub> emissions (199 $\pm$ 91 pmol m<sup>-2</sup> h<sup>-1</sup>) follow the pattern of both oceanic and atmospheric values. The seasonal cycle of CHBr<sub>3</sub> production (171±191 pmol m<sup>-2</sup> h<sup>-1</sup>) is similar to the cycle of CHBr<sub>3</sub> concentration, while the 339 340 sharp peak in May/June when the spring phytoplankton bloom evolves in the North Atlantic, is not reflected in the oceanic 341 concentrations. The strong seasonality in CHBr<sub>3</sub> production leads to a CHBr<sub>3</sub> excess rate during summer (JJA: 342 103 pmol m<sup>-2</sup> h<sup>-1</sup>) and a CHBr<sub>3</sub> deficit rate in winter time (DJF: -114 pmol m<sup>-2</sup> h<sup>-1</sup>), respectively (Figure 5d).

343 The MLR analysis indicates, that on an annual basis, variations in atmospheric mixing ratios are mainly associated with CHBr<sub>3</sub> ocean concentrations (Table 1, R<sup>2</sup>=0.89, p-value<0.05) and vice versa (Figure 6a,d). A higher surface water CHBr<sub>3</sub> 344 345 concentration increases the emissions to the atmosphere resulting in increasing atmospheric mixing ratios. According to 346 the MLR analysis on a seasonal basis, oceanic CHBr<sub>3</sub> concentrations are mainly driven by the oceanic production during 347 MAM (R<sup>2</sup>=0.93, p-value<0.05) and SON (R<sup>2</sup>=0.99, p-value<0.05) (Table S1), which increases from March to June 348 sharply to 680 pmol m<sup>-2</sup> h<sup>-1</sup> before gradually decreasing in SON (Figure 5d). Annually, atmospheric mixing ratios are 349 mainly driven by the oceanic concentration (Figure 6d, Table 1, R<sup>2</sup>=0.89, p-value<0.05). This is also true on a seasonal 350 basis except for the winter (DJF) season, where SST is the main (indirect) factor influencing the atmospheric mixing ratio 351 variations (Table S1, R<sup>2</sup>=0.95, p-value<0.05), as lower SSTs increase the solubility of CHBr<sub>3</sub> and hardly any oceanic 352 CHBr<sub>3</sub> production occurs. Thus, the emissions of CHBr<sub>3</sub> decrease, even during comparably high wind speeds, which 353 leads to a decreasing atmospheric mixing ratio. CHBr<sub>3</sub> emissions are mainly driven by oceanic concentrations on an





354 annual basis (Figure 6g, Table 1, R<sup>2</sup>=0.81, p-value<0.05). The MLR results further indicate that the driving factors are 355 highly variable, when looking on a seasonal basis. During spring (MAM), wind speed, SST, and CHBr3 production are almost equally driving the emissions (Table S1). During this season, CHBr<sub>3</sub> emissions are pretty constant at 356 357 130±29 pmol m<sup>-2</sup> h<sup>-1</sup> (Figure 5d). CHBr<sub>3</sub> production and SSTs slightly increase in spring, and an increase of the emissions 358 to the atmosphere is expected. However, emissions stay constant as surface wind speed decreases and lower the emissions. 359 This regional and seasonal pattern explicitly illustrates the interaction of different conditions influencing the CHBr<sub>3</sub> 360 emissions. During summer (JJA), low winds and a high oceanic CHBr<sub>3</sub> concentration equally influence the increasing 361 emissions (Table S1). In contrast to spring, higher SSTs (lower solubility) are only of minor importance during JJA. In autumn and winter, decreasing emissions are mainly driven by decreasing SSTs and, in DJF, additionally by high 362 atmospheric mixing ratios (in comparison to oceanic concentrations), which additionally dampens the emissions (Table 363 364 S1).

## 365 Tropical West Pacific

Figure 4 shows that the equatorial regions of the Atlantic and Pacific Oceans generally have positive kbal and therefore 366 367 are a source of oceanic CHBr<sub>3</sub>, which is transported to other oceanic regions. CHBr<sub>3</sub> ocean concentrations in the tropical 368 West Pacific (4°S - 4°N, 150°E - 180°E) show a reduced seasonal cycle in comparison to the above discussed North Atlantic region (Figure 5b). Oceanic concentrations are on average 9.11 pmol L<sup>-1</sup> which is significantly higher than 369 370 average concentrations in the North Atlantic while the seasonal amplitude (min: 7.42 pmol L<sup>-1</sup> in March; max: 371 10.58 pmol L<sup>-1</sup> in October) is less pronounced. CHBr<sub>3</sub> production (536±42 pmol m<sup>-2</sup> h<sup>-1</sup>), CHBr<sub>3</sub> emissions 372 (492±84 pmol m<sup>-2</sup> h<sup>-1</sup>) and atmospheric mixing ratios (1.07±0.08 ppt) show hardly any seasonality (Figure 5b, e). The 373 same is true for SST (29.50±0.28 °C) and wind speed (4.71±0.76 m s<sup>-1</sup>) (Figure 5h). The CHBr<sub>3</sub> balance is positive 374 throughout the whole year except for DJF (Figure 5e). During this period high wind speed leads to higher emission than 375 production rates and induces low oceanic concentrations which results in a CHBr3 deficit. However, this deficit does not 376 compensate for the CHBr<sub>3</sub> excess during the rest of the year leading to an overall positive  $k_{bal}$  of 32 pmol m<sup>-2</sup> h<sup>-1</sup>.

377 MLR analysis shows that the wind speed is the main factor influencing the variations of oceanic CHBr<sub>3</sub> concentrations

 $378 \qquad (R^2=0.51, p-value<0.05), CHBr_3 \text{ atmospheric mixing ratios } (R^2=0.74, p-value<0.05), and CHBr_3 \text{ emissions } (R^2=0.73, p-value<0.05), and (R^2=0.05), and (R$ 

 $\label{eq:stars} 379 \qquad value < 0.05) \ on \ an \ annual \ basis \ (Figure \ 6b, e, h, \ Table \ 1) \ in \ the \ equatorial \ region. \ During \ JJA \ and \ SON \ CHBr_3 \ production$ 

drives the CHBr<sub>3</sub> concentrations (Table S1, JJA: R<sup>2</sup>=0.43, p-value<0.05, SON: R<sup>2</sup>=0.51, p-value<0.05) which increases

381 from 477 pmol m<sup>-2</sup> h<sup>-1</sup> in July to 618 pmol m<sup>-2</sup> h<sup>-1</sup> by the end of September. This results in an increase of oceanic CHBr<sub>3</sub>

382 concentrations as all other parameters stay constant during that period.

#### 383 Southern Ocean

384 The selected Southern Ocean region  $(71^{\circ}\text{S} - 66^{\circ}\text{S}, 40^{\circ}\text{W} - 20^{\circ}\text{W})$  experiences water temperatures, which are negative almost any time around the year (Figure 5i). Average temperature is -1.08°C with minimum temperatures of -1.71°C in 385 386 September (late winter) and maximum temperatures of +0.19°C in January/February (late summer). Wind speed in this 387 region is nearly constant throughout the year (7.33 m s<sup>-1</sup>) with lower average wind speed of 5.76 m s<sup>-1</sup> only during austral 388 summer (DJF, Figure 5h). Oceanic CHBr3 concentrations in the Southern Ocean region are on average higher 389 (5.38 pmol L<sup>-1</sup>) than in the North Atlantic region with maximum concentrations (7.74 pmol L<sup>-1</sup>) in January and lowest concentrations (4.04 pmol L<sup>-1</sup>) end of December. This sharp increase of CHBr<sub>3</sub> concentrations within two months 390 391 demonstrates the limited biological activity period, visible in the CHBr<sub>3</sub> production rate (Figure 5f). Due to decreasing 392 SSTs as well as a decreased day length, CHBr3 production rates are almost zero from May to October and sharply increase 393 up to 306 pmol m<sup>-2</sup> h<sup>-1</sup> in January. Atmospheric mixing ratios are highest (max: 0.85 ppt) from January to beginning of 394 April but decline very slowly (Figure 5c) under low light levels, until they reach their minimum of 0.31 ppt in November.





- Constant high atmospheric mixing ratios, due to light limitations in combination with very low SST, and decreasing oceanic CHBr<sub>3</sub> concentrations, after the short summer bloom in DJF, influence the switch from positive emissions to negative emissions between April and July (Figure 5f). CHBr<sub>3</sub> is in excess during times of CHBr<sub>3</sub> production (DJF) and is almost balanced during the autumn and wintertime (April – September, Figure 5f). On annual basis, CHBr<sub>3</sub> is almost balanced in this region with a slight excess of 15 pmol m<sup>-2</sup> h<sup>-1</sup>.
- 400



Figure 6: Main drivers of oceanic CHBr<sub>3</sub> concentrations (a, b, c), atmospheric mixing ratios (d, e, f) and CHBr<sub>3</sub> emissions (g, h, i) in the North Atlantic (a, d, g), tropical West Pacific (b, e, h) and Southern Ocean (c, f, i). Different colours denote different seasons of the year. Each data point represents a daily mean average over the specific case study area.

401 Overall, the MLR analysis confirms that atmospheric mixing ratios are the main factor influencing the variations of 402 oceanic concentrations on annual basis (Figure 6c, Table 1, R<sup>2</sup>=0.56, p-value<0.05). CHBr<sub>3</sub> production is the driving 403 factor for the ocean concentration during autumn (MAM, Table S1). During this time CHBr<sub>3</sub> production decreases and 404 so does the ocean concentration Figure 5c,f). The atmospheric mixing ratio is mainly driven by oceanic concentrations at 405 times with high oceanic CHBr<sub>3</sub> concentrations (DJF: R<sup>2</sup>=0.88, p-value<0.05) and by SST during cold winter times (JJA: 406 R<sup>2</sup>=0.95, p-value<0.05) (Table S1). In winter during low light levels, atmospheric CHBr3 reactions are reduced, which 407 increases the lifetime of atmospheric CHBr<sub>3</sub>. Additionally, low SSTs increase the solubility of oceanic CHBr<sub>3</sub>. These two 408 conditions favour the overall dampening of the CHBr<sub>3</sub> sea-to-air emissions during winter (JJA). During the summer in 409 DJF CHBr<sub>3</sub> emissions are mainly driven by SST (Table S1, R<sup>2</sup>=0.60, p-value<0.05), as the solubility of CHBr<sub>3</sub> in the ocean significantly decreases due to the increasing sea surface temperatures. After this short summer period, temperatures 410 411 decline in autumn (MAM) and increase the solubility of oceanic CHBr<sub>3</sub>, which results in decreased emissions (Figure 412 5f,i). During winter (JJA) and spring (SON), surface temperatures and oceanic CHBr3 concentrations stay low and 413 therefore, increasing emissions are mainly driven by decreasing atmospheric mixing ratios.

414

These results demonstrate the benefits of simulating  $CHBr_3$  in a fully coupled ESM configuration to calculate driving factors for different parameters on temporal and spatial basis. Studying the influence of one or more parameters on the variability of other parameters in the model is not realistic when using prescribed oceanic concentrations or atmospheric mixing ratios. Investigating the  $CHBr_3$  cycling in different locations and different time scales helps to understand their interconnection and to better integrate their results in today's as well as in a future climate.

420





421

422

Table 1: Annual coefficients of predictors for each MLR in the different case studies. Bold coefficients are the highest value
 within a MLR analysis of one parameter and region and act as indicator for the driving factors of the predicted parameter
 (Eq. (16)-(18)).

	Predictor parameter	North Atlantic	Tropical West Pacific	Southern Ocean
Ocean concentration	Wind speed	-0.02	-0.96	-0.10
	SST	<0.01	<0.01	<0.01
	Atm. mixing ratio	0.68	0.19	0.60
	CHBr <sub>3</sub> production	0.39	0.13	0.53
	CHBr <sub>3</sub> emissions	<0.01	<0.01	<0.01
Atmospheric mixing ratio	Wind speed	0.29	0.94	0.55
	SST	0.32	<0.01	< 0.01
	Ocean concentration	0.93	0.12	1.07
	CHBr <sub>3</sub> production	<0.01	0.02	<0.01
	CHBr <sub>3</sub> emissions	<0.01	<0.01	<0.01
CHBr3 emissions	Wind speed	0.20	1.27	0.21
	SST	0.67	0.50	0.53
	Ocean concentration	0.83	0.16	1.00
	CHBr <sub>3</sub> production	<0.01	0.10	<0.01
	Atm. mixing ratio	-0.32	-0.02	-1.22

426 427

## 428 **3.5** Global bromoform emission inventories

429 A comparison of our modelled versus published global CHBr3 emissions are presented in Figure 7. Global annual CHBr3 emissions from top-down approaches are 449 Gg yr<sup>-1</sup>, 528 Gg yr<sup>-1</sup> and 592 Gg yr<sup>-1</sup> based on calculations from Liang et 430 al. (2010), Ordóñez et al. (2012) and Warwick et al. (2006), respectively. These inventories are about two to eight times 431 432 higher than calculated annual emissions from bottom-up approaches, which are in the range of 76 Gg yr<sup>-1</sup> (Stemmler et 433 al., 2015) to 238 Gg yr<sup>-1</sup> (Lennartz et al., 2015). Our results (214 Gg yr<sup>-1</sup>) are similar to emission estimates published by Ziska et al. (2013) of 215 Gg yr<sup>-1</sup> but significantly higher than the 76 Gg yr<sup>-1</sup> estimate by Stemmler et al. (2015), which 434 435 is based on the oceanic CHBr3 observations from HalOcAt. 436 As we apply a 2.38 higher CHBr<sub>3</sub> production rate in the ocean as Stemmler et al. (2015), we simulate a production rate

of 0.88 Gmol yr<sup>-1</sup> compared to 0.37 Gmol yr<sup>-1</sup> by Stemmler et al. (2015). Our emissions (214 Gg yr<sup>-1</sup>) are 2.82 times
higher (Figure 7, global values) compared to the emission estimate (76 Gg yr<sup>-1</sup>) from Stemmler et al. (2015). Our model

adaption is based on the higher bulk CHBr<sub>3</sub> production ratio ( $\beta_0$ ) according to Kurihara et al. (2012) and Roy (2010) (see

440 Section 2.1.1). This production rate is at the higher end of published values. Therefore, the resulting CHBr<sub>3</sub> production

441 can be seen as an upper limit.





442 Comparing bottom-up and top-down approaches, the annual CHBr<sub>3</sub> emissions account for ~47% (105 Gg yr<sup>-1</sup>) and ~66% 443 (351 Gg yr<sup>-1</sup>), respectively, from the tropics ( $20^{\circ}$ S –  $20^{\circ}$ N, Figure 7), which account for ~37% of global oceanic surface, 444 underlining the tropics as the most important source region of CHBr<sub>3</sub> of the earth.

Emissions in the middle latitudes (20 to 50°N/S) of the NH and SH show a similar distinction between top-down and bottom-up approaches. However, the annual CHBr<sub>3</sub> emissions are only half that of the tropics. Natural open ocean emission estimates from our study are proportional to the surface area between NH and SH in the middle latitudes. This relationship is reversed for the top-down approach estimates. Top-down emission estimates are higher in the NH compared to the SH although the oceanic surface area is lower in the NH (17%) compared to the SH (26%). This indicates

- 450 the strong influence of coastal emissions on observational atmospheric mixing ratios used in top-down approaches.
- 451 In the high latitudes (50-90°N/S), emissions of bottom-up approaches are in the same range (SH) and even higher (NH)



compared to top-down approaches (Figure 7). In the northern polar region (8% of global oceanic surface area), CHBr3 emissions from our study account for 3% (6 Gg yr<sup>-1</sup>) of global emissions and are significantly lower than the other two bottom-up approaches from Lennartz et al. (2015) (11%, 27 Gg yr<sup>-1</sup>) and Ziska et al. (2013) (21%, 45 Gg yr<sup>-1</sup>), which appears mainly due to the resolved seasonality within our study. According to the HalOcAt database, no measurements are recorded from November to February and from May to September north of 50°N in the NH and south of 50°S in the SH, respectively. Therefore, the prescribed atmospheric values in Ziska et al. (2013) but also

Figure 7: Comparison of global and latitudinally binned annual CHBr<sub>3</sub> emissions from different studies. Grey and blue bars denote top-down and bottom-up approaches, respectively.

467 in Stemmler et al. (2015) are biased to the ice-free summer months, with higher atmospheric mixing ratios, thus artificially 468 dampening the emissions from the ocean to the atmosphere during winter seasons. Due to the influence of the annually 469 fixed prescribed atmospheric mixing ratios in Stemmler et al. (2015), negative emissions are more pronounced between 470  $50^{\circ}$ N/S and  $70^{\circ}$ N/S up to -100 pmol m<sup>-2</sup> h<sup>-1</sup> at ~  $60^{\circ}$ N/S. Our lesser negative emissions in the coupled ESM approach 471 appear more realistic as they are not based on summer biased prescribed values.

472

473 Our global CHBr3 emission inventory indicates distinct differences to the top-down approaches reflecting only 40%-50% 474 of global emissions calculated by Liang et al. (2010), Ordóñez et al. (2012) and Warwick et al. (2006). Atmospheric 475 CHBr<sub>3</sub> values in the top-down approaches are higher than the calculated atmospheric mixing ratios from our fully coupled 476 model analysis. They include elevated coastal (Scenario A and C, Liang et al., 2010), including anthropogenic (Ordóñez 477 et al., 2012) sources, which may partly explain the discrepancy. 478 An additional explanation for the overall higher atmospheric mixing ratios of CHBr3 from observations could be that 479 observations from coastal areas (100 km within the coastline) were excluded from this study and are not represented in 480 the model, as they are difficult to quantify (e.g. tide-dependent CHBr3 emissions of macroalgae) with a horizontal model

 $\label{eq:second} 481 \qquad \text{resolution of $1^\circ$. However, coastal emissions lead to higher atmospheric mixing ratios of CHBr_3 (Fuhlbrügge et al., 2013; and 2013) and 2013 are second at the second s$ 

482 Fuhlbrügge et al., 2016; Hepach et al., 2016), which can be transported to remote open ocean regions, while these higher

483 observational values are not included in the model results (Figure 3).





484 Another explanation for the underestimation of the modelled atmospheric mixing ratios compared to observations is the 485 use of air-sea gas exchange parameterizations, whose uncertainty is estimated to be 25% (Wanninkhof, 2007) and may 486 be underestimated up to 75% (Yang et al., 2022) at low wind speeds.

#### 487 **4** Conclusions and Outlook

488 Our study is the first one to derive oceanic and atmospheric CHBr3 concentrations, as well as emissions, from a fully 489 coupled ESM simulation. The model prognostically simulates oceanic CHBr3 production by phytoplankton and includes oceanic CHBr<sub>3</sub> loss due to air-sea gas exchange, photolysis, hydrolysis and halogen substitution. Atmospheric loss of 490 491 CHBr<sub>3</sub> is described by photolysis and the reaction with OH. We validate the model results with more than 5,100 oceanic 492 and 8,400 atmospheric observations from the HalOcAt database. The simulated global mean CHBr3 emission rate (214 493 Gg yr<sup>-1</sup>) is in the range of previously published bottom-up approaches (76-238 Gg yr<sup>-1</sup>), but significantly lower than topdown approaches (449-592 Gg yr<sup>1</sup>). The model allows to realistically resolve seasonal and spatial variations and to 494 495 identify different drivers of oceanic and atmospheric CHBr3 variability on regional and seasonal scales. Our results 496 indicate that only during high productive seasons a consequently high CHBr<sub>3</sub> production drives high oceanic CHBr<sub>3</sub> 497 concentrations. During low productive seasons, relatively high atmospheric mixing ratios suppress the gas exchange and 498 consequently influence variations in oceanic CHBr3 concentrations. In tropical regions (e.g. tropical West Pacific) with a 499 small seasonal cycle, but high oceanic concentrations and atmospheric mixing ratios, wind speed is the main factor driving 500 the variability of oceanic and atmospheric CHBr<sub>3</sub> and its emissions. The results clearly indicate the benefit of a fully 501 coupled ocean-atmosphere-biogeochemistry ESM. In earlier modelling studies, prescribed, fixed atmospheric or oceanic 502 values were applied, which bias the seasonal impact of different factors on oceanic and atmospheric CHBr3 and 503 subsequently induce additional uncertainties to the magnitude of CHBr3 emissions. 504 Our fully coupled ocean atmosphere approach resolves natural biogenic oceanic and atmospheric CHBr<sub>3</sub> including their 505 emissions at relatively high temporal and spatial model resolution. Validation with observational data shows good

agreement for large scale spatial patterns and we attribute the remaining model-data differences to missing coastal sources, which are not implemented in the model. Comparison with other published CHBr<sub>3</sub> inventories indicates that approaches without seasonality lack to resolve CHBr<sub>3</sub> emissions especially in high latitudes.

509 Our results demonstrate the potential for applying a fully coupled ESM to elucidate the primary drivers of the observed 510 CHBr<sub>3</sub> concentrations and emissions variability across spatial and temporal scales. Moreover, this model set-up allows 511 to implement additional oceanic derived VSLS in order to further investigate their influence on the atmospheric chemistry. 512 The dissociation of open ocean natural derived CHBr3 from coastal area derived CHBr3 in this study reveal that coastal 513 derived CHBr3 influences open ocean atmospheric mixing ratios. Therefore, implementing natural coastal next to 514 anthropogenic sources and concurrent model resolution increase in these areas will help to further close the gap of 515 published CHBr3 emission estimates between bottom-up and top-down approaches. Long-term future changes in CHBr3 516 dynamics under various scenarios should be investigated with a fully coupled ESM, to study the impact of climate change 517 on CHBr3 dynamics, e.g. in the Arctic, associated with loss of sea-ice and its climate feedback through interaction with 518 ozone chemistry.

519

- 522
- 523
- 524

<sup>520</sup> **Data availability.** Observational data can be downloaded from https://halocat.geomar.de. Model data will be archived 521 and will be made available upon request.





525

526	Author	contributions.
-----	--------	----------------

- 527 DB wrote the manuscript and led the discussion with contributions from all authors. DB analysed the model simulations
- 528 and prepared the graphics. JFT and DJLO implemented the CHBr<sub>3</sub> model code changes in NorESM2 in discussion with
- 529 BQ and all other authors. JFT carried out the model runs. KK led this project and initiated the research idea for this study.
- 530
- 531 **Competing interests.** The authors declare that they have no conflict of interest.
- 532
- 533 Acknowledgements. This work was financed by the Research Council of Norway through the KeyCLIM project
- 534 (295046) within the KLIMAFORSK/POLARFORSK program. Resources for the model simulations and data storage
- 535 were provided by Sigma2 the National Infrastructure for High Performance Computing and Data Storage in Norway.
- 536

#### 537 References

- Agus, E., Voutchkov, N., and Sedlak, D. L.: Disinfection by-products and their potential impact on the quality of water
   produced by desalination systems: A literature review, Desalination, 237, 214-237, 10.1016/j.desal.2007.11.059, 2009.
- Assmann, K. M., Bentsen, M., Segschneider, J., and Heinze, C.: An isopycnic ocean carbon cycle model, Geoscientific
   Model Development, 3, 143-167, DOI 10.5194/gmd-3-143-2010, 2010.
- 542 Bates, N. R. and Merlivat, L.: The influence of short-term wind variability on air-sea CO
- 543 exchange, Geophysical Research Letters, 28, 3281-3284, Doi 10.1029/2001gl012897, 2001.
- 544 Butler, J. H., King, D. B., Lobert, J. M., Montzka, S. A., Yvon-Lewis, S. A., Hall, B. D., Warwick, N. J., Mondeel, D. J.,
- Aydin, M., and Elkins, J. W.: Oceanic distributions and emissions of short-lived halocarbons, Global Biogeochemical
   Cycles, 21, Artn Gb1023
- 547 10.1029/2006gb002732, 2007.
- Carpenter, L. J. and Liss, P. S.: On temperate sources of bromoform and other reactive organic bromine gases, Journal of
   Geophysical Research-Atmospheres, 105, 20539-20547, Doi 10.1029/2000jd900242, 2000.
- 550 Danabasoglu, G., Lamarque, J. F., Bacmeister, J., Bailey, D. A., DuVivier, A. K., Edwards, J., Emmons, L. K., Fasullo, 551 J., Garcia, R., Gettelman, A., Hannay, C., Holland, M. M., Large, W. G., Lauritzen, P. H., Lawrence, D. M., Lenaerts, J. 552 T. M., Lindsay, K., Lipscomb, W. H., Mills, M. J., Neale, R., Oleson, K. W., Otto-Bliesner, B., Phillips, A. S., Sacks, 553 W., Tilmes, S., van Kampenhout, L., Vertenstein, M., Bertini, A., Dennis, J., Deser, C., Fischer, C., Fox-Kemper, B., Kay, J. E., Kinnison, D., Kushner, P. J., Larson, V. E., Long, M. C., Mickelson, S., Moore, J. K., Nienhouse, E., Polvani, 554 555 L., Rasch, P. J., and Strand, W. G.: The Community Earth System Model Version 2 (CESM2), Journal of Advances in 556 Modeling Earth Systems, 12, e2019MS001916, ARTN e2019MS001916 557 10.1029/2019MS001916, 2020.
- Daniel, J. S., Solomon, S., Portmann, R. W., and Garcia, R. R.: Stratospheric ozone destruction: The importance of
   bromine relative to chlorine, Journal of Geophysical Research-Atmospheres, 104, 23871-23880, Doi
   10.1029/1999jd900381, 1999.
- Dorf, M., Butler, J. H., Butz, A., Camy-Peyret, C., Chipperfield, M. P., Kritten, L., Montzka, S. A., Simmes, B., Weidner,
   F., and Pfeilsticker, K.: Long-term observations of stratospheric bromine reveal slow down in growth, Geophysical
- 563 Research Letters, 33, Artn L24803
- 564 10.1029/2006gl027714, 2006.
- Fiehn, A., Quack, B., Stemmler, I., Ziska, F., and Krüger, K.: Importance of seasonally resolved oceanic emissions for
   bromoform delivery from the tropical Indian Ocean and west Pacific to the stratosphere, Atmospheric Chemistry and
   Physics, 18, 11973-11990, 10.5194/acp-18-11973-2018, 2018.
- 568 Fuhlbrügge, S., Krüger, K., Quack, B., Atlas, E., Hepach, H., and Ziska, F.: Impact of the marine atmospheric boundary
- layer conditions on VSLS abundances in the eastern tropical and subtropical North Atlantic Ocean, Atmospheric
   Chemistry and Physics, 13, 6345-6357, 10.5194/acp-13-6345-2013, 2013.





- 571 Fuhlbrügge, S., Quack, B., Tegtmeier, S., Atlas, E., Hepach, H., Shi, Q., Raimund, S., and Krüger, K.: The contribution 572 of oceanic halocarbons to marine and free tropospheric air over the tropical West Pacific, Atmospheric Chemistry and
- 573 Physics, 16, 7569-7585, 10.5194/acp-16-7569-2016, 2016.
- 574 Gebhardt, S.: Biogenic emission of halocarbons, Mainz, Univ., Diss., 2008, 10.25358/openscience-2211, 2008.
- 575 Gettelman, A., Mills, M. J., Kinnison, D. E., Garcia, R. R., Smith, A. K., Marsh, D. R., Tilmes, S., Vitt, F., Bardeen, C.
- G., McInerny, J., Liu, H. L., Solomon, S. C., Polvani, L. M., Emmons, L. K., Lamarque, J. F., Richter, J. H., Glanville,
  A. S., Bacmeister, J. T., Phillips, A. S., Neale, R. B., Simpson, I. R., DuVivier, A. K., Hodzic, A., and Randel, W. J.: The
- Whole Atmosphere Community Climate Model Version 6 (WACCM6), Journal of Geophysical Research-Atmospheres,
- 579 124, 12380-12403, 10.1029/2019jd030943, 2019.
- Gschwend, P. M., Macfarlane, J. K., and Newman, K. A.: Volatile halogenated organic compounds released to seawater
   from temperate marine macroalgae, Science, 227, 1033-1035, 10.1126/science.227.4690.1033, 1985.
- Hense, I. and Quack, B.: Modelling the vertical distribution of bromoform in the upper water column of the tropical
   Atlantic Ocean, Biogeosciences, 6, 535-544, DOI 10.5194/bg-6-535-2009, 2009.
- Hepach, H., Quack, B., Tegtmeier, S., Engel, A., Bracher, A., Fuhlbrügge, S., Galgani, L., Atlas, E. L., Lampel, J., Friess,
  U., and Krüger, K.: Biogenic halocarbons from the Peruvian upwelling region as tropospheric halogen source,
  Atmospheric Chemistry and Physics, 16, 12219-12237, 10.5194/acp-16-12219-2016, 2016.
- Hossaini, R., Chipperfield, M. P., Montzka, S. A., Rap, A., Dhomse, S., and Feng, W.: Efficiency of short-lived halogens
  at influencing climate through depletion of stratospheric ozone, Nature Geoscience, 8, 186-190, 10.1038/Ngeo2363,
  2015.
- Jia, Y., Hahn, J., Quack, B., Jones, E., Brehon, M., and Tegtmeier, S.: Anthropogenic Bromoform at the Extratropical
   Tropopause, Geophysical Research Letters, 50, e2023GL102894, ARTN e2023GL102894
- 592 10.1029/2023GL102894, 2023.
- Kurihara, M., Iseda, M., Ioriya, T., Horimoto, N., Kanda, J., Ishimaru, T., Yamaguchi, Y., and Hashimoto, S.: Brominated
   methane compounds and isoprene in surface seawater of Sagami Bay: Concentrations, fluxes, and relationships with
   phytoplankton assemblages, Marine Chemistry, 134, 71-79, 10.1016/j.marchem.2012.04.001, 2012.
- Laube, J. C., Tegtmeier, S., Fernandez, R. P., Harrison, J., Hu, L., Krummel, P., Mahieu, E., Park, S., and Western, L.:
  Update on Ozone-Depleting Substances (ODSs) and Other Gases of Interest to the Montreal Protocol, 978-9914-733-976, 2023.
- Law, K., Sturges, W., Blake, D., Blake, N., Burkholder, J., Butler, J., Cox, R., Haynes, P., Ko, M., and Kreher, K.:
   Halogenated very short-lived substances, Chapter 2 in: Scientific Assessment of Ozone Depletion: 2006, Global Ozone
   Research and Monitoring Project–Report No. 50, World Meteorological Organization, Geneva, Switzerland, 572, 2007.
- Lennartz, S. T., Krysztofiak, G., Marandino, C. A., Sinnhuber, B. M., Tegtmeier, S., Ziska, F., Hossaini, R., Krüger, K.,
  Montzka, S. A., Atlas, E., Oram, D. E., Keber, T., Bönisch, H., and Quack, B.: Modelling marine emissions and
  atmospheric distributions of halocarbons and dimethyl sulfide: the influence of prescribed water concentration vs.
  prescribed emissions, Atmospheric Chemistry and Physics, 15, 11753-11772, 10.5194/acp-15-11753-2015, 2015.
- Liang, Q., Stolarski, R. S., Kawa, S. R., Nielsen, J. E., Douglass, A. R., Rodriguez, J. M., Blake, D. R., Atlas, E. L., and
   Ott, L. E.: Finding the missing stratospheric Br: a global modeling study of CHBr3 and CH2Br2, Atmospheric Chemistry
   and Physics, 10, 2269-2286, DOI 10.5194/acp-10-2269-2010, 2010.
- Maas, J., Tegtmeier, S., Jia, Y., Quack, B., Durgadoo, J. V., and Biastoch, A.: Simulations of anthropogenic bromoform
   indicate high emissions at the coast of East Asia, Atmospheric Chemistry and Physics, 21, 4103-4121, 10.5194/acp-21 4103-2021, 2021.
- Maier-Reimer, E.: Geochemical cycles in an ocean general circulation model. Preindustrial tracer distributions, Global
   Biogeochemical Cycles, 7, 645-677, 10.1029/93gb01355, 2012.
- 614 Masson-Delmotte, V., Zhai, P., Pirani, A., Connors, S. L., Péan, C., Berger, S., Caud, N., Chen, Y., Goldfarb, L., Gomis,
- M. I., Huang, M., Leitzell, K., Lonnoy, E., Matthews, J. B. R., Maycock, T. K., Waterfield, T., Yelekçi, O., Yu, R., Zhou,
   B., and Ipcc: Annex I: Observational Products, in: Climate Change 2021 The Physical Science Basis, Cambridge
- B., and Ipcc: Annex I: Observational Products, in: Climate Change 2021 The Physical Science Basis, Cambridge
  University Press, Cambridge, United Kingdom and New York, NY, USA, 2061-2086, 10.1017/9781009157896.015,
  2023.





- Montzka, S. A., Reimann, S., Engel, A., Kruger, K., Sturges, W. T., Blake, D., Dorf, M., Fraser, P., Froidevaux, L., and
   Jucks, K.: Ozone-depleting substances (ODSs) and related chemicals (chapter 1), 2011.
- Moore, R. M., Geen, C. E., and Tait, V. K.: Determination of Henry's Law constants for a suite of naturally occurring halogenated methanes in seawater, Chemosphere, 30, 1183-1191, 10.1016/0045-6535(95)00009-w, 1995.
- Moore, R. M., Webb, M., Tokarczyk, R., and Wever, R.: Bromoperoxidase and iodoperoxidase enzymes and production
   of halogenated methanes in marine diatom cultures, Journal of Geophysical Research-Oceans, 101, 20899-20908, Doi
   10.1029/96jc01248, 1996.
- Navarro, M. A., Atlas, E. L., Saiz-Lopez, A., Rodriguez-Lloveras, X., Kinnison, D. E., Lamarque, J. F., Tilmes, S., Filus,
  M., Harris, N. R., Meneguz, E., Ashfold, M. J., Manning, A. J., Cuevas, C. A., Schauffler, S. M., and Donets, V.: Airborne
  measurements of organic bromine compounds in the Pacific tropical tropopause layer, Proc Natl Acad Sci U S A, 112,
  13789-13793, 10.1073/pnas.1511463112, 2015.
- Nightingale, P. D., Malin, G., Law, C. S., Watson, A. J., Liss, P. S., Liddicoat, M. I., Boutin, J., and Upstill-Goddard, R.
  C.: In situ evaluation of air-sea gas exchange parameterizations using novel conservative and volatile tracers, Global
  Discrete huminal Caulty 14, 272–287, Doi:10.1020/1000-b000001.2000
- 632 Biogeochemical Cycles, 14, 373-387, Doi 10.1029/1999gb900091, 2000.
- Ordóñez, C., Lamarque, J. F., Tilmes, S., Kinnison, D. E., Atlas, E. L., Blake, D. R., Sousa Santos, G., Brasseur, G., and
   Saiz-Lopez, A.: Bromine and iodine chemistry in a global chemistry-climate model: description and evaluation of very
   short-lived oceanic sources, Atmospheric Chemistry and Physics, 12, 1423-1447, 10.5194/acp-12-1423-2012, 2012.
- Papanastasiou, D. K., McKeen, S. A., and Burkholder, J. B.: The very short-lived ozone depleting substance CHBr
  (bromoform): revised UV absorption spectrum, atmospheric lifetime and ozone depletion potential, Atmospheric
  Chemistry and Physics, 14, 3017-3025, 10.5194/acp-14-3017-2014, 2014.
- Quack, B. and Wallace, D. W. R.: Air-sea flux of bromoform: Controls, rates, and implications, Global Biogeochemical
   Cycles, 17, 10.1029/2002gb001890, 2003.
- 641 Quack, B., Atlas, E., Petrick, G., Stroud, V., Schauffler, S., and Wallace, D. W. R.: Oceanic bromoform sources for the
- tropical atmosphere, Geophysical Research Letters, 31, Artn L23s05
- 643 10.1029/2004g1020597, 2004.
- Roy, R.: Short-term variability in halocarbons in relation to phytoplankton pigments in coastal waters of the central eastern Arabian Sea, Estuarine Coastal and Shelf Science, 88, 311-321, 10.1016/j.ecss.2010.04.011, 2010.
- Saiz-Lopez, A., Fernandez, R. P., Li, Q., Cuevas, C. A., Fu, X., Kinnison, D. E., Tilmes, S., Mahajan, A. S., Gomez
  Martin, J. C., Iglesias-Suarez, F., Hossaini, R., Plane, J. M. C., Myhre, G., and Lamarque, J. F.: Natural short-lived
  halogens exert an indirect cooling effect on climate, Nature, 618, 967-973, 10.1038/s41586-023-06119-z, 2023.
- Sala, S., Bönisch, H., Keber, T., Oram, D. E., Mills, G., and Engel, A.: Deriving an atmospheric budget of total organic
   bromine using airborne in situ measurements from the western Pacific area during SHIVA, Atmospheric Chemistry and
   Physics, 14, 6903-6923, 10.5194/acp-14-6903-2014, 2014.
- 652 Salawitch, R. J.: Atmospheric chemistry: biogenic bromine, Nature, 439, 275-277, 10.1038/439275a, 2006.
- Sinnhuber, B. M., Sheode, N., Sinnhuber, M., Chipperfield, M. P., and Feng, W.: The contribution of anthropogenic
   bromine emissions to past stratospheric ozone trends: a modelling study, Atmospheric Chemistry and Physics, 9, 2863 2871, DOI 10.5194/acp-9-2863-2009, 2009.
- 556 Stemmler, I., Hense, I., and Quack, B.: Marine sources of bromoform in the global open ocean global patterns and 557 emissions, Biogeosciences, 12, 1967-1981, 10.5194/bg-12-1967-2015, 2015.
- Tegtmeier, S., Atlas, E., Quack, B., Ziska, F., and Krüger, K.: Variability and past long-term changes of brominated very short-lived substances at the tropical tropopause, Atmospheric Chemistry and Physics, 20, 7103-7123, 10.5194/acp-20-7103-2020, 2020.
- Tjiputra, J. F., Assmann, K., Bentsen, M., Bethke, I., Otterå, O. H., Sturm, C., and Heinze, C.: Bergen Earth system model
   (BCM-C): model description and regional climate-carbon cycle feedbacks assessment, Geoscientific Model
   Davalopment 3, 123, 141, DOI 10, 5104/amd 3, 123, 2010, 2010





- Tjiputra, J. F., Schwinger, J., Bentsen, M., Morée, A. L., Gao, S., Bethke, I., Heinze, C., Goris, N., Gupta, A., He, Y. C.,
  Olivié, D., Seland, O., and Schulz, M.: Ocean biogeochemistry in the Norwegian Earth System Model version 2
  (NorESM2), Geoscientific Model Development, 13, 2393-2431, 10.5194/gmd-13-2393-2020, 2020.
- Tokarczyk, R. and Moore, R. M.: Production of Volatile Organohalogens by Phytoplankton Cultures, Geophysical
   Research Letters, 21, 285-288, Doi 10.1029/94gl00009, 1994.
- Villamayor, J., Iglesias-Suarez, F., Cuevas, C. A., Fernandez, R. P., Li, Q. Y., Abalos, M., Hossaini, R., Chipperfield, M.
   P., Kinnison, D. E., Tilmes, S., Lamarque, J. F., and Saiz-Lopez, A.: Very short-lived halogens amplify ozone depletion

trends in the tropical lower stratosphere, Nature Climate Change, 13, 554-+, 10.1038/s41558-023-01671-y, 2023.

Wang, S. Y., Kinnison, D., Montzka, S. A., Apel, E. C., Hornbrook, R. S., Hills, A. J., Blake, D. R., Barletta, B., Meinardi,
S., Sweeney, C., Moore, F., Long, M., Saiz-Lopez, A., Fernandez, R. P., Tilmes, S., Emmons, L. K., and Lamarque, J.
F.: Ocean Biogeochemistry Control on the Marine Emissions of Brominated Very Short-Lived Ozone-Depleting
Substances: A Machine-Learning Approach, Journal of Geophysical Research-Atmospheres, 124, 12319-12339,
10.1029/2019jd031288, 2019.

Wanninkhof, R.: The Impact of Different Gas Exchange Formulations and Wind Speed Products on Global Air-Sea CO2
Fluxes, in: Transport at the Air-Sea Interface, edited by: Garbe, C. S., Handler, R. A., and Jähne, B., Environmental
Science and Engineering, Springer Berlin Heidelberg, Berlin, Heidelberg, 1-23, 10.1007/978-3-540-36906-6\_1, 2007.

- 680 Warwick, N. J., Pyle, J. A., Carver, G. D., Yang, X., Savage, N. H., O'Connor, F. M., and Cox, R. A.: Global modeling
- 681 of biogenic bromocarbons, Journal of Geophysical Research-Atmospheres, 111, Artn D24305
- 682 10.1029/2006jd007264, 2006.
- Washington, J. W.: Hydrolysis Rates of Dissolved Volatile Organic-Compounds Principles, Temperature Effects and
   Literature-Review, Ground Water, 33, 415-424, DOI 10.1111/j.1745-6584.1995.tb00298.x, 1995.
- 685 Yang, M. X., Bell, T. G., Bidlot, J. R., Blomquist, B. W., Butterworth, B. J., Dong, Y. X., Fairall, C. W., Landwehr, S.,
- 686 Marandino, C. A., Miller, S. D., Saltzman, E. S., and Zavarsky, A.: Global Synthesis of Air-Sea CO
- 687 Transfer Velocity Estimates From Ship-Based Eddy Covariance Measurements, Front. Mar. Sci., 9, ARTN 826421
- 688 10.3389/fmars.2022.826421, 2022.

689 Ziska, F., Quack, B., Abrahamsson, K., Archer, S. D., Atlas, E., Bell, T., Butler, J. H., Carpenter, L. J., Jones, C. E.,

- Harris, N. R. P., Hepach, H., Heumann, K. G., Hughes, C., Kuss, J., Krüger, K., Liss, P., Moore, R. M., Orlikowska, A.,
- Raimund, S., Reeves, C. E., Reifenhäuser, W., Robinson, A. D., Schall, C., Tanhua, T., Tegtmeier, S., Turner, S., Wang,
- L., Wallace, D., Williams, J., Yamamoto, H., Yvon-Lewis, S., and Yokouchi, Y.: Global sea-to-air flux climatology for
   bromoform, dibromomethane and methyl iodide, Atmospheric Chemistry and Physics, 13, 8915-8934, 10.5194/acp-13 8915-2013, 2013.
- 695