



# Physiological and ecological tipping points caused by ocean acidification

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## 10 Abstract.

Ocean acidification is predicted to cause profound shifts in many marine ecosystems by impairing the ability of calcareous taxa to calcify and grow, and by influencing the photo-physiology of many others. In both calcifying and non-calcifying taxa, ocean acidification could further impair the ability of marine life to regulate internal pH, and thus metabolic function and/or behaviour. Identifying tipping points at which these effects will occur for different taxa due to the direct impacts of ocean acidification on organism physiology is difficult and they have not adequately been determined for most taxa, nor for ecosystems at higher levels. This is due to the presence of both resistant and sensitive species within most taxa. However, calcifying taxa such as coralline algae, corals, molluscs, and sea urchins appear to be most sensitive to ocean acidification. Conversely, non-calcareous seaweeds, seagrasses, diatoms, cephalopods, and fish tend to be more resistant, or even benefit from the direct effects of ocean acidification. While physiological tipping points of the effects of ocean acidification either do not exist or are not well defined, their direct effects on organism physiology will have flow on indirect effects. These indirect effects will cause ecologically tipping points in the future through changes in competition, herbivory and predation. Evidence for indirect effects and ecological change is mostly taken from benthic ecosystems in warm temperate–tropical locations *in situ* that have elevated CO<sub>2</sub>. Species abundances at these locations indicate a shift away from calcifying taxa and towards non-calcareous at high CO<sub>2</sub> concentrations. For example, lower abundance of corals and coralline algae, and higher covers of non-calcareous macroalgae, often turfing species, at elevated CO<sub>2</sub>. However, there are some locations where only minor changes, or no detectable change occurs. Where ecological tipping points do occur, it is usually at locations with naturally elevated pCO<sub>2</sub> concentrations of 500 μatm or more, which also corresponds to just under that concentrations where the direct physiological impacts of ocean acidification are detectable on the most sensitive taxa in laboratory research (coralline algae and corals). Collectively, the available data support the concern that ocean acidification will most likely cause ecological change in the near future in most benthic marine ecosystems, with tipping points in some ecosystems at as low as 500 μatm pCO<sub>2</sub>. However, much more further research is required to more adequately quantify and model the extent of these impacts in order to accurately project future marine ecosystem tipping points under ocean acidification.



## 1 Introduction

Ocean acidification is the process of increasing absorption of atmospheric CO<sub>2</sub> by the surface seawaters, leading to a decrease in pH and shift in the speciation of dissolved inorganic carbon (DIC). As a result, future seawater concentrations of HCO<sub>3</sub><sup>-</sup>, H<sup>+</sup>, and CO<sub>2</sub> will be higher, while CO<sub>3</sub><sup>2-</sup> will be lower. These changes in seawater carbonate chemistry will have complex biological consequences, as all four parameters mentioned previously hold physiological significance for various marine taxa (Hurd et al., 2019). The direct effects of ocean acidification manifest mostly through three mechanisms: 1) negatively impacting calcification (and recruitment of calcareous organisms); 2) altering photo-physiology; and 3) affecting acid-base regulation (i.e., internal pH regulation) or the energy expended in this process. These effects are usually subtle and have relatively minor direct negative effects when compared to stressors like marine heatwaves. However, the cumulative impacts of these physiological changes on individual species will result in more significant ecological effects. Comprehensive meta-analyses on the impacts of ocean acidification on biological processes and ecological outcomes support these statements (Leung et al., 2022; Kroeker et al., 2013b). However, these analyses rarely assess changes in processes at specific time points or at *p*CO<sub>2</sub> values corresponding to such time points. Instead, they usually determine whether standardised responses are different from zero. Thus, in this study, we evaluate when key responses in different taxa would differ from zero and provide *p*CO<sub>2</sub> values of when this typically occurs, if it is possible to make such statements. We also provide background information describing the key physiological and ecological impacts of ocean acidification.

## 2 Direct impacts on key physiological processes

### 2.1 Calcification

Calcium carbonate minerals precipitate or dissolve relative to their saturation state ( $\Omega$ ), which is determined by [Ca<sup>2+</sup>] multiplied by [CO<sub>3</sub><sup>2-</sup>] and divided by *K*<sub>s</sub>. *K*<sub>s</sub>, being dependent on seawater temperature, salinity, and pressure (Morse and Mackenzie, 1990). Consequently, under ocean acidification, declines in [CO<sub>3</sub><sup>2-</sup>] and  $\Omega$  lead to reduced mineral precipitation in seawater. In theory, minerals like calcite have higher  $\Omega$  values than aragonite, while aragonite has a higher  $\Omega$  than high Mg calcite (Andersson et al., 2008; Bathurst, 1972; Feely et al., 2004). Initial experimental work demonstrated that simulated ocean acidification reduces calcification in most taxa (Ries et al., 2009). However, there are also instances of resistant species or entire taxa (Kroeker et al., 2013b; Leung et al., 2022). Thus, the initial assumption was largely that ocean acidification reduces calcification by decreasing  $\Omega$ . Indeed, taxa that precipitated lower  $\Omega$  minerals tended to be more strongly impacted by ocean acidification (Kroeker et al., 2013b). CO<sub>3</sub><sup>2-</sup> was more highly correlated than other components of seawater carbonate chemistry in experiments where the independent role of pH, HCO<sub>3</sub><sup>-</sup> and/or CO<sub>3</sub><sup>2-</sup> were tested, and teased apart independently (Comeau et al., 2013a; Schneider and Erez, 2006). However, calcification generally occurs internally (Ries, 2011). In organisms such as corals, it takes place several tissue layers deep (Allemand et al., 2004), while in coralline algae, it occurs



65 within the cell wall, one to several cells deep (Cornwall et al., 2017a). This internal location is referred to as the calcifying fluid (Mcculloch et al., 2012a; Mcculloch et al., 2012b). Notably, there are no known biological transporters of  $\text{CO}_3^{2-}$ , and it remains membrane impermeable in all organisms. Therefore, it is more likely that increases in external  $[\text{H}^+]$  drive the observed declines in calcification rates observed in corals, a process offset by the provision of greater DIC under ocean acidification (Jokiel, 2013).

To calcify, marine organisms must create conditions that favour the precipitation of calcium carbonate at the site of calcification. Most calcifying organisms actively adjust their internal chemistry to initiate the inorganic precipitation of calcium carbonate by increasing the pH and ensuring a consistent supply of essential ions required for calcification ( $\text{CO}_3^{2-}$ ,  $\text{HCO}_3^-$ ,  $\text{Ca}^{2+}$ ,  $\text{Mg}^{2+}$ ) (Mcculloch et al., 2017; Decarlo et al., 2018). This is done by actively removing protons from the site of calcification via specific cellular transporter. The inorganic carbon utilized for calcification originates from a combination of metabolic  $\text{CO}_2$ ,  $\text{HCO}_3^-$  transported to the site of calcification via cellular transporters, and ions transported via paracellular pathways (e.g., Venn et al., 2020). Additionally, marine organisms have developed specific proteins that aid in maintaining elevated calcium carbonate precipitation (Marin et al., 2007; Drake et al., 2013). While the general principles of calcification are similar across many taxa, the response to ocean acidification is highly taxa and species-specific. Here we will focus on three of the most studied calcifying taxa: corals, molluscs, and calcifying macroalgae. These three taxa are also identified as the most at risk from the effects of ocean acidification.

Corals have received significant attention in studies related to ocean acidification. However, results regarding a potential tipping point in their response to ocean acidification are inconclusive, and the shape of the relationship between calcification and pH (or  $\text{pCO}_2$ ) is still a subject of debate. Early research suggested a linear response of corals growth to pH (Doney et al., 2009; Anthony et al., 2008), but some studies challenged this finding, showing non-linear relationships between calcification and pH (or  $\Omega$ ) (e.g., Ries et al., 2009). It is however important to note that most of these studies had limited treatment levels, making it difficult to assess the shape of the ocean acidification to calcification relationship and detect tipping points. More recent studies, using a wider range of treatments (Comeau et al., 2013b, 2014) and species (Okazaki et al., 2017), experimentally demonstrated that the response of coral calcification to  $\text{pCO}_2$  is generally linear. Linear relationships have been considered the best representation of the coral response of corals to ocean acidification in recent meta-analyses (Cornwall et al., 2021; Kornder et al., 2018). Therefore, there is no clear tipping point in the response of corals to ocean acidification, as their response is generally linear, highly species-specific, and influenced by other factors such as light, temperature, and feeding. At the reef level, studies indicate a linear decline in calcification with the saturation state, but the magnitude of this decrease (i.e., the slope) varies among reefs and likely depends on community composition. Some studies project a shift to net



dissolution when  $\Omega$  decreases below 2, a level that could be reached by the end of the century (Andersson and Gledhill, 2013; Albright et al., 2018).

Molluscs, especially commercial ones, have also been extensively studied in relation to their response to ocean acidification due to their economic importance. Previous reviews on the effects of ocean acidification on molluscs, though conducted several years ago, revealed that the response of adults and juveniles varies significantly among species and even within species (Gazeau et al., 2013; Parker et al., 2013). Similar to corals, the response of molluscs to ocean acidification is strongly influence by their environmental history and interactions with other environmental factors (Falkenberg et al., 2019; Thomsen and Melzner, 2010). Different mollusc species precipitate various forms of calcium carbonate, such as calcite, aragonite, or high Mg-Calcite, which are known to respond differently to ocean acidification. Moreover, there is evidence suggesting that molluscs could undergo transgenerational acclimation to ocean acidification, potentially mitigating the negative effects of decreasing pH (Parker et al., 2021). As a result, determining a tipping point at which molluscs calcification and growth might collapse proves to be extremely challenging. Instead, it is more likely that some species will experience a steady decrease in calcification and shell strength, while others may thrive in lower pH conditions. The complexity of these responses makes it difficult to predict the overall impact of OA on molluscs with certainty.

Coralline algal are extremely sensitive to the effects of ocean acidification, with their calcification rates, cover in the field at natural CO<sub>2</sub> sites, recruitment in laboratory and in situ work, and internal pH at the site of calcification all negatively impacted by ocean acidification (Cornwall et al., 2022). Calcification rates of coralline algae are negatively affected 75% of the time, with in most instance a linear decrease of calcification with decreasing pH. However, like in molluscs, greater resistance to ocean acidification has been demonstrated in coralline algae grown for multiple generations (6 generations) under conditions simulating the effects of ocean acidification, versus those from controls (Cornwall et al., 2020; Moore et al., 2021). Additionally, some species display natural tolerance to ocean acidification (Cornwall et al., 2018; Cornwall et al., 2017a). Although the physiological mechanisms via which this tolerance is achieved is relatively unexplored, there is possible links between greater upregulation of pH within the site of calcification and maintenance of calcification under ocean acidification. Irrespective of these tolerance species, it is likely coralline algae will be among the first taxa to clearly show negative impacts of ocean acidification, as on average they are universally sensitive. This will cause large flow on effects at ecological levels (see section below on this), as they play important roles in maintaining coral reef growth and cementation (Cornwall et al., 2023), and act as crucial settlement substrates for invertebrate larvae such as corals, abalone and sea urchins (Roberts, 2001; Fabricius et al., 2017).

Declines in calcification within most taxa due to ocean acidification is expected, but determining a specific tipping point for when this decline will occur is a complex task. The complexity arises from the diverse approaches used in past research, the



different atmospheric CO<sub>2</sub> concentrations projected under various climate change scenarios, and the lack of alignment between meta-analyses and these atmospheric CO<sub>2</sub> concentration scenarios. We summarise the past attempts at meta-analyses, or pseudo-meta-analyses in Table 1. Notably, there is a significant overlap between the majority of past meta-analyses (Hendriks et al., 2010; Kroeker et al., 2013b; Kroeker et al., 2010; Harvey et al., 2013; Kornder et al., 2018) and more recent meta-analyses (Leung et al., 2022; Cornwall et al., 2021; Cornwall et al., 2022). Leung et al. offer a more detailed description of the responses of various calcifying taxa to ocean acidification compared to most prior meta-analyses. However, they group all relevant studies necessary to identify tipping points into a single category (~pH 7.9 to pH 7.6). Additionally, they group physiologically different calcifying algae that are not grouped in some other recent analyses on coralline algae [and corals], which attempt to determine temporal differences in effect sizes (Cornwall et al., 2021; Cornwall et al., 2022). Furthermore, Wittmann and Pörtner (2013), although not a true meta-analysis, provide detailed and interesting breakdowns of the impacts of different pH range values on calcification rates of various animal taxa. To estimate calcification and growth rates of most taxa, we consider them largely interchangeable since the ability to create new skeletal material is essential for inorganic growth and size increases. Therefore, for our estimates, we draw on all recent relevant research, including the work of Wittmann and Pörtner.

Evidence strongly suggests that the calcification rates of coccolithophores, pteropods, coralline algae, *Halimeda* spp., corals, bivalves, and gastropods are likely to experience significant decreases due to ocean acidification by the end of the century under RCP8.5 emissions scenarios or similar conditions (e.g., ~pH 7.63 and 936  $\mu\text{atm CO}_2$ ) (Leung et al., 2022; Cornwall et al., 2022). However, the evidence remains unclear for foraminifera, calcifying sponges, bryozoans, crustaceans, polychaetes, and echinoderms, as there is mixed evidence presently available (Leung et al., 2022). Specifically, coralline algal calcification rates significantly decline from present-day rates within ocean acidification scenarios that simulate RCP4.5 2050 and above (e.g., 538  $\mu\text{atm}$ ), whereas corals experience similar departures around 2050 RCP8.5 (e.g., 572  $\mu\text{atm}$ ) (Cornwall et al., 2021). Though not equivalent, the point at which more than 50% of studies find negative impacts on all measures of organism fitness is around 780 to 840  $\mu\text{atm}$  for molluscs and echinoderms respectively, and above 2000  $\mu\text{atm}$  for crustaceans (Wittmann and Pörtner, 2013). Although not all of these measurements directly involve calcification, this highlights the difficulty in determining physiological tipping points for changes in calcification rates for specific taxa, given the available and disparate research.

## 2.2. Photo-physiology

Most macroalgae and all phytoplankton species have the ability to directly take up HCO<sub>3</sub><sup>-</sup> from seawater during photosynthesis, converting it into complex sugars (Raven et al., 2011; Raven et al., 2002a). Similarly, all marine phototrophs can passively take up CO<sub>2</sub> through diffusion for the same process (Raven et al., 2012), with some of them relying solely on this diffusive



uptake. In corals, symbiotic dinoflagellates use DIC within the coral tissues through the same process (Raven et al., 2020). The active uptake of  $\text{HCO}_3^-$  using pumps, symports, antiports is referred to as a  $\text{CO}_2$  concentrating mechanism (CCM) (Raven et al., 2011). The evolution of CCM was believed to be a response to the low concentrations and slow diffusivity of  $\text{CO}_2$  in seawater compared to air (Raven et al., 2002b; Raven et al., 2008), where ~90% of seawater DIC is in the form of  $\text{HCO}_3^-$  and only about 1% is in the form of  $\text{CO}_2$ . Therefore, using  $\text{HCO}_3^-$  was considered to theoretically overcome DIC limitation. However, this reasoning is simplistic. This is noted in the various works by Raven and colleagues (e.g., Raven et al., 2005; Raven and Beardall, 2014, 2016), and we summarise here. There are various types of CCMs that differ in their efficiency in natural seawater. Although seawater pH typically is around ~8.05, it can be higher in regions with high photosynthetic uptake of DIC and high water retention (Rivest et al., 2017). In these habitats,  $\text{CO}_2$  concentrations would be extremely small and species utilising diffusive  $\text{CO}_2$  uptake would be at a competitive disadvantage over those with CCMs (Hepburn et al., 2011). In additions to true CCMs, there are various associated external and internal carbonic anhydrase enzymes that collectively enhance the diffusive  $\text{CO}_2$  uptake externally, or convert internal  $\text{CO}_2$  into  $\text{HCO}_3^-$ , thereby regulating internal  $\text{CO}_2$  levels and maintaining internal pH. The efficiency and production of these enzymes can vary between species. Moreover, both the creation of CCMs and carbonic anhydrase enzymes require energy and nutrients. As a result, possessing a CCM does not guarantee that a species is not DIC limited in the habitats it currently occupies. Increasing  $\text{CO}_2$  (and possibly  $\text{HCO}_3^-$ ) could potentially alleviate DIC limitation in marine phototrophs that are constrained by DIC availability. However, quantifying the exact benefits that these organisms would receive is challenging because there has been no comprehensive effort to study this by subjecting organisms with well-known CCMs and carbonic anhydrases to seawater conditions resembling ocean acidification. Nevertheless, it has been observed that several DIC-limited species do show an increase in their photosynthetic rates and/or growth rates in response to elevated DIC or ocean acidification (Kroeker et al., 2010; Harvey et al., 2013).

Increasing  $\text{CO}_2$  concentrations could also lead to species with a CCM using additional  $\text{CO}_2$  when the external  $\text{CO}_2$  concentration is higher, thus downregulating their CCM (Hepburn et al., 2011). The potential benefits of this downregulation might be minimal, but it does appear as though both marine and freshwater macroalgae at elevated  $\text{CO}_2$  concentrations either downregulate their CCM or switch the ratio of  $\text{HCO}_3^-$  to  $\text{CO}_2$  uptake (Cornwall et al., 2017b; Maberly et al., 2014). Another possible outcome is that higher external  $[\text{CO}_2]$  levels could reduce the leakage of internal  $\text{CO}_2$  back into seawater, saving energy and possibly nutrients. However, the specific impacts of these processes under ocean acidification are challenging to predict and likely depend on local irradiance availability and nutrient concentrations relative to the specific species' requirements.

Photosynthetic efficiency ( $F_v/F_m$ ) is observed to decline in many taxa (e.g., corals, seaweeds) under ocean acidification (Leung et al., 2022; Cornwall et al., 2022), but the significance of these decreases as stress indicators is debatable. For instance, declines in  $F_v/F_m$  of approximately 0.4 or more are typically associated with acute heat stress (e.g., Schoepf et al., 2019),



185 whereas declines caused by simulated ocean acidification are usually on the order of 0.02. Therefore, it is possible that these declines are in response to non-stress related increases in cellular CO<sub>2</sub> or decreased pH, affecting the functioning of the photosystem II. Further research is needed to understand the physiological and molecular mechanisms that underpin these responses in various phototrophic organisms and to determine if they indeed represent significant changes that would impact future photosynthetic species in a future high CO<sub>2</sub> ocean.

190 Meta-analyses highlight that ocean acidification will impact the photo-physiology of marine species. However, due to the complexity of the underlying processes (as highlighted above), there is considerable variability within and between taxa, and limited data on photosynthetic responses. Consequently, it becomes even more challenging to pinpoint a tipping point at which changes in photosynthetic rates would occur. On average, ocean acidification seems to have a positive impact on coccolithophore photosynthetic rates, but there is no average detectable effect in foraminifera, symbiont bearing sponges, corals, and calcifying algae (Leung et al., 2022). However, both corals and coralline algae show declines in photosynthetic efficiency, and there is evidence of both strong positive and negative impacts on their photosynthetic rates (Leung et al., 2022; Cornwall et al., 2022). The underlying physiological mechanisms for these negative observations require further investigation. Older meta-analyses also indicate positive effects of ocean acidification on seagrass, diatom, and non-calcareous macroalgal growth and photosynthesis, though with varied responses (see Table 1) (Harvey et al., 2013; Kroeker et al., 2013b).

### 2.3 Internal pH regulation

200 Ocean acidification also presents significant changes for the regulation of internal pH in different tissues or fluids of many calcifying taxa. This is evident in corals (Mcculloch et al., 2012b; Venn et al., 2013), coralline algae (Cornwall et al., 2017a), *Halimeda* spp. (Comeau et al., 2019), echinoderms (Stumpp et al., 2012), and crustaceans (Carter et al., 2013) for example. However, some of these species show an increased ability to upregulate internal pH at sites of calcification, allowing them to maintain calcification under simulated ocean acidification conditions. As a result, relying solely on meta-analyses to predict when specific species will be impacted by ocean acidification may have limitations. That being said, meta-analyses do indicate significant impacts of OA on pH regulation in certain echinoderms and corals (Leung et al., 2022), as well as in coralline algae (Leung et al., 2022; Cornwall et al., 2022). However, the extent to which OA affects pH within different components of various taxa remains unexplored, and this could potentially explain some of the variability in responses even among non-calcifying taxa, such as many seaweeds. This is a vital controller of organism metabolism and fitness, but further research on this topic is required before tipping points could be identified.



### 210 3. Changes at naturally high CO<sub>2</sub> locations

Research on the potential effects of ocean acidification on marine organisms highlight the connection between increasing atmospheric and oceanic CO<sub>2</sub> levels, changes in carbonate chemistry, and their impacts on marine organisms. As highlighted previously, numerous experimental studies have documented significant impacts of future ocean acidification on diverse aspects of individual species' physiology, life history, and ecology, as well as on populations. However, when attempting to  
215 assess the impact of ocean acidification at higher levels of biological organisation, such as community structure, food web dynamics, and ecosystem function, uncertainty arises. Particular difficulties arise when attempting to predict the outcomes of interactions among numerous species. Nevertheless, all these individual responses contribute to ecosystem change, and so ocean acidification is anticipated to lead towards changes in the structure, composition, and functioning of marine ecosystems.

220 Ecosystem changes, although usually gradual, can reach tipping points where they undergo sudden and significant shifts, leading to alterations in the structure and function of biological communities. These regime shifts are of particular concern due to the fact that the newly established habitats are primarily composed of species that have lower ecological, functional, and human value when compared to the habitats they have replaced. Examples of this include the extensive displacement of key foundation species (e.g., coral reefs and kelp forests) by simplified degraded ones like turf algae-dominated systems (e.g.,  
225 Knowlton, 1992; Moy and Christie, 2012; Wernberg et al., 2016). Ecosystem changes associated with climate change are evident today in various ecosystems and biogeographic regions, often associated with temperature. However, these changes are often complex, involving concurrent shifts in multiple environmental factors and processes. On the other hand, the direct attribution of ongoing ecosystem changes, or specific elements, to anthropogenic ocean acidification is far more difficult to establish.

230 Seasonal upwelling regions, such as the California coast, offer valuable insights into the contemporary impacts of altered carbonate chemistry. In these areas, nutrient-rich waters from deeper layers of the ocean, which are also characterised by high CO<sub>2</sub> and low pH, are transported to the surface. This process provides examples of how low pH waters can bring about ecosystem changes by impacting early life stages of economically and ecologically important species (e.g., Pacific Shellfish  
235 species) (Feely et al., 2008). Another contemporary example includes coral reefs in the Equatorial Upwelling System, such as those found in the Galápagos and Cocos Islands, which exhibit low species diversity and limited development of carbonate reef frameworks attributed to the low pH/aragonite saturation levels in the upwelling waters of the region (Glynn, 2001; Manzello, 2010; Manzello et al., 2008). One of the issues in the attribution of ecosystem change to ocean acidification (beyond the confounding effects of other environmental factors) is associated with the difficulty in the large-scale monitoring of  
240 carbonate chemistry. However, a study examining artificial ocean alkalisation on a localised scale yielded important findings (Albright et al., 2016). It showed that increasing alkalinity to pre-industrial levels resulted in an increase in net community calcification at a coral reef on One Tree Island, Great Barrier Reef. This study therefore suggests that contemporary



communities are experiencing a decline in net community calcification compared to pre-industrial conditions, indicating that coral reef growth may potentially already be impaired by ocean acidification. Conversely, the addition of CO<sub>2</sub> to a coral reef at the same location caused declines in net community calcification rates beyond those expected to occur due to the direct impacts on coral (Albright et al., 2018). This could be due to resident coralline algae or carbonate sediments, that are high in abundance in some areas on this reef and more prone to dissolution at rates much higher than their calcification rates. While contemporary examples, such as regions of seasonal upwelling, promote our understanding of the current impacts of altered carbonate chemistry, understanding the potential future impacts of ocean acidification remains crucial.

Research efforts are also currently underway into the long-term effects of ocean acidification on ecosystem-level processes using natural analogues (Figure 1). Natural analogues for ocean acidification predominantly involve a natural, localised change in carbonate chemistry conditions that mimic future ocean acidification conditions which can be compared with an adjacent region under contemporary conditions. These include CO<sub>2</sub> seeps, associated with volcanic CO<sub>2</sub> emissions, and semi-enclosed bays (e.g., semi-enclosed lagoons, and mangrove estuaries), where certain ecologically significant parameters for marine life (carbonate chemistry, temperature, oxygen) closely resemble or even surpass the projected conditions anticipated for the end-of-the-century. Since marine communities and ecosystems are home to a wide range of species, that each interact with each other, these research approaches importantly enable researchers to simultaneously examine the response of a naturally-assembled community as a whole; regardless of whether they are investigating a specific population, functional group or the community as a whole. While not exact replicas of future conditions, these natural laboratories present unique opportunities to investigate the mechanisms by which ecosystems may adapt and respond to the challenges of climate change.

In early studies utilising CO<sub>2</sub> seeps (Hall-Spencer et al., 2008), the loss of vulnerable calcifying organisms and the potential benefit observed in certain non-calcified primary producers (such as seagrass and algae) were noted. However, it was also observed that some calcifying organisms managed to survive while certain non-calcifying primary producers were being lost. This significant finding emphasised the limitations of solely considering the direct physiological impacts of ocean acidification when predicting community-level responses, highlighting the need for a comprehensive understanding of the ecological interactions within a community (Fabricius et al papers). Ongoing research at a number of different CO<sub>2</sub> seeps has consistently documented shifts in habitat structure and community composition along the natural *p*CO<sub>2</sub> gradients. These shifts often favored simplified systems with reduced biodiversity and less ecological complexity over larger habitat-forming species (Sunday et al., 2016). Examples include transitions from canopy-forming brown algae to turf (Ischia, Italy), hard corals to soft corals (Iwotorishima, Japan), corals and macroalgae to turf (Shikine Island, Japan) (Inoue et al., 2013; Hall-Spencer et al., 2008; Agostini et al., 2018). Many of these shifts occurred at *p*CO<sub>2</sub> conditions projected by the mid-century and suggest that thresholds of tipping points may be reached in the near future. We project that in the future, many ecosystems (but not all) will shift from states dominated by the cover of marine forests or corals to more depauperate states.



280 The changes in the extent and structural complexity of biogenic habitat observed within CO<sub>2</sub> seeps can additionally mediate further biodiversity shifts, with the impact of ocean acidification on habitat-forming species projected to lead to lower associated species diversity (Sunday et al., 2016). Field surveys in a temperate Pacific CO<sub>2</sub> seep found that the ocean acidification-driven transition in habitat resulted in a reduction in the diversity of associated fish species, this loss is largely attributed to the disappearance of species that are closely associated with habitats that have undergone significant loss themselves. Moreover, it led to the selection of fish species that are more adapted to simplified ecosystems dominated by algae (Cattano et al., 2016). Similar declines in reef-associated macroinvertebrate communities, as well as small cryptic invertebrates, have also been observed in tropical CO<sub>2</sub> seeps (Fabricius et al., 2014; Plaisance et al., 2021).

285 Habitats play a crucial role in shaping the diversity of associated organisms, including macroinvertebrates and fishes; however, it is important to note that these organisms themselves also exert control over the habitats they inhabit through their grazing. In acidified conditions, it has been suggested that the top-down control exerted by some grazers may diminish, as evidenced by reductions in the diversity, abundance and size of many marine fauna observed at CO<sub>2</sub> seeps (Garilli et al., 2015; Harvey et al., 2018; Harvey et al., 2016). For example, there is a decrease in the number of sea urchin feeding halos at a Mediterranean CO<sub>2</sub> seeps (Kroeker et al., 2013a). Top-down control in ecological systems functions as a feedback loop, where the abundance and behaviour of grazers play a crucial role in shaping the population dynamics and composition of the habitat. With any loss or reduction of grazers and their top-down control, there is a potential for significant alterations in the overall structure and functioning of an ecosystem, and so diminished top-down control is likely contributing towards the simplification of coastal ecosystems.

300 The early-stage recruitment and trajectory of community development are important mechanisms responsible for altering shallow marine communities exposed to ocean acidification. Ocean acidification has notable effects on the composition of prokaryotic biofilm communities on both natural and artificial substrates deployed in CO<sub>2</sub> seeps (Kerfahi et al., 2014; Taylor et al., 2014). Similarly, studies on eukaryotic biofilm communities in these seep environments have also observed changes in community composition (Allen et al., 2021). Collectively, these findings indicate that the ecological patterns driving biofilm community responses in CO<sub>2</sub> seeps are characterised by a selection for distinct cohorts of organisms, where the conditions at the reference and elevated pCO<sub>2</sub> sites represent distinct niches. When examining the longer-term development of communities, the divergence of community composition over time is often influenced by changes in competitive interactions among habitat-forming organisms (Kroeker et al., 2013a; Kroeker et al., 2011). In general, dominant species tend to outcompete others during the early stages of succession, rapidly outgrowing or overgrowing them, which can result in the community being locked into a depauperate and low-complexity state.



When examining the impact of ocean acidification on ecosystem change, the interplay of physiological thresholds and ecological interactions, such as habitat provisioning, community development, and top-down control, becomes crucial. The absence or loss of certain species indicates that either their physiology and/or their ecological dynamics are unable to cope with the adverse effects of ocean acidification. For example, even if a species has the physiological capacity to tolerate ocean acidification conditions, it may still be competitively excluded during community development, leading to its absence in elevated  $p\text{CO}_2$  conditions. For example, if crucial settlement substrates such as coralline algae are absent, this ocean acidification may then have large indirect negative impacts on corals, irrespective of direct effects (Fabricius et al., 2015; Fabricius et al., 2011; Fabricius et al., 2017). However, our understanding of the physiological mechanisms that drive these changes are poorly understood in most cases, and measurements of direct competition are often lacking in both *in situ* and laboratory research. Individual  $\text{CO}_2$  seeps, while analogues for specific factors like  $\text{CO}_2$ , may not fully represent the broader influences of climate change. Subsequently, in recent years, research has moved towards replicating research across multiple  $\text{CO}_2$  seeps (to encompass multiple environmental drivers alongside ocean acidification) (Cornwall et al., 2017b; Comeau et al., 2022), as well as making use of other natural analogues like semi-enclosed lagoons (e.g., Palau and New Caledonia), which have provided important insights into corals that have adapted to these systems (Kurihara et al., 2021; Tanvet et al., 2023).

#### 4. Conclusions

Ocean acidification has complex biological consequences, affecting calcification, photo-physiology, and pH regulation in various marine taxa. These direct impacts have been quantified in more than one thousand research articles to date. However, more research is required on internal pH regulation of various taxa (see Table 1), and up to date meta-analyses of non-calcareous taxa (see table 2) is also required. Quantitative projections of the tipping points at which  $\text{CO}_2$  will have negative (or positive) impacts is also required for most taxa, where here we generally rely on semi-qualitative assessments for all taxa except corals and coralline algae. These negative and positive impacts will have flow on effects at ecological levels. Natural analogues, like  $\text{CO}_2$  seeps, offer insights into the long-term effects of ocean acidification on marine ecosystems. These natural laboratories show shifts in habitat structure and community composition along  $p\text{CO}_2$  gradients, leading to potential biodiversity loss and ecosystem simplification. Overall, ocean acidification has complex and far-reaching effects on marine life, and predicting specific tipping points for different taxa is challenging due to the multitude of factors involved. Continued research and monitoring efforts are essential to comprehend and address the impacts of ocean acidification on marine ecosystems in the future.

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## Conflicts of interest

The contact author has declared that none of the authors has any competing interests.

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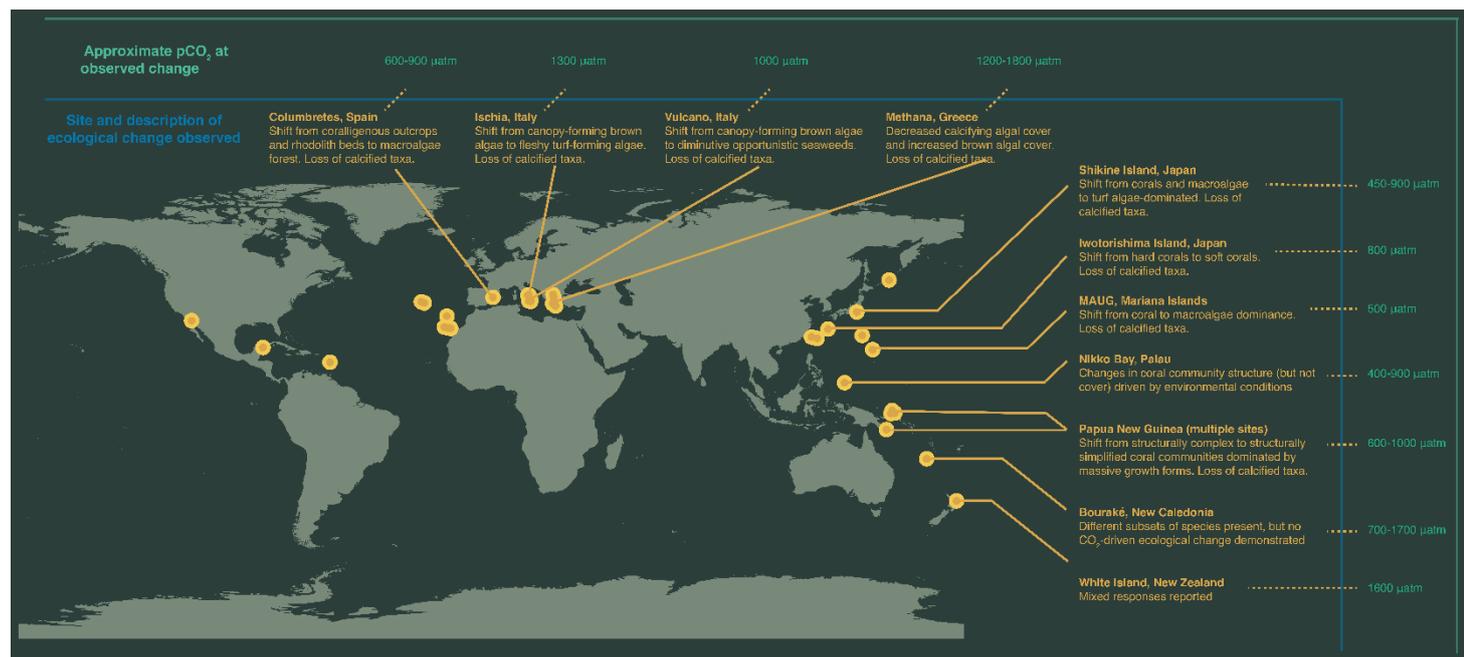
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580 **Figure 1: Map of locations of high CO<sub>2</sub> locations used to make inferences regarding impacts of OA on future marine ecosystems.** Labelled are sites where patterns in ecosystem change is described at Columbretes, Spain (Linares et al., 2015); Ischia, Italy (Hall-Spencer et al., 2008; Porzio et al., 2011); Vulcano, Italy (Cornwall et al., 2017b); Methana, Greece (Baggini et al., 2014); Shikine Island, Japan (Agostini et al., 2018); Iwotorishima Island, Japan (Inoue et al., 2013); Maug, Mariana Islands (Enochs et al., 2015); Papua New Guinea (Fabricius et al., 2011; Comeau et al., 2022); Bouraké, New Caledonia (Maggioni et al., 2021); White Island, New Zealand (Nagelkerken et al., 2016; Blain et al., 2021).



585 Table 1: Summary of the impacts of ocean acidification on calcification/growth, photosynthetic rates and internal pH regulation, with tipping points from literature identified where possible. Top to bottom rows, taxa are listed from first to be negatively impacted at the top to most positively impacted at the bottom. Red = significant decrease in values from zero, yellow = no detectable difference, green = significant increase in values from zero, grey = not enough data to determine. Where possible, quantitative or qualitative tipping points given in  $\mu\text{atm CO}_2$ . Empty cells = not applicable. See Table 2 for further details on references used to generate tipping points.

Taxon	Calcification/growth	Photosynthetic rates	Internal pH regulation	References
Coralline algae	~538		~538 to 930	(Cornwall et al., 2021; Cornwall et al., 2022; Leung et al., 2022)
Corals	~572	~538 to 930	~538 to 930	(Leung et al., 2022; Cornwall et al., 2021)
Molluscs	~781			(Leung et al., 2022; Wittmann and Pörtner, 2013)
Sea urchins	~870			(Leung et al., 2022; Wittmann and Pörtner, 2013)
Polychaetes	~538 to 930			(Leung et al., 2022)
Foraminifera	~538 to 930			(Leung et al., 2022)
Coccolithophores	~538 to 930	~538 to 930		(Leung et al., 2022; Meyer and Riebesell, 2015)
Sponges				(Leung et al., 2022)



Echinoderms (non-sea urchin)			~538 to 930	(Leung et al., 2022)
<i>Halimeda</i> spp.				(Schubert et al., 2023)
Bryozoans				(Leung et al., 2022)
Crustaceans				(Leung et al., 2022)
Seagrasses		~538 to 930		(Kroeker et al., 2013b; Harvey et al., 2013)
Cephalopods	~538 to 930			(Leung et al., 2022)
Fish	~538 to 930			(Kroeker et al., 2013b; Harvey et al., 2013)
Diatoms	~538 to 930			(Kroeker et al., 2013b)
Non-calcareous seaweeds	~538 to 930			(Kroeker et al., 2013b; Harvey et al., 2013)

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**Table 2: Meta-analyses or pseudo-meta-analyses that support the results of table 1.** Note: some studies excluded here due to merging of different taxa, inclusion of too few studies, or assessing only specific processes in combination with ocean acidification (e.g., Rivest et al., 2017; Ramajo et al., 2016). Studies listed chronologically and greyed out if their results not used due to complete overlap with more recent research. Note: fish behaviour excluded from these analyses. Meta-analyses checked by searching “ocean acidification” AND “meta-analysis” on July 2023. All non-regional assessments including photosynthesis, calcification or internal pH included below.

Study	Limited to specific taxa?	Year literature search conducted	Number of studies used	Taxa used to create table 1
Leung (2022)	Calcifying taxa	2020	985	All calcifying taxa except <i>Halimeda</i> spp.
Cornwall et al. (2022)	Coralline algae	2021	64	Coralline algae
Schubert et al. (2023)	<i>Halimeda</i> spp.	Unknown	31	<i>Halimeda</i> spp.
Cornwall et al. (2021)	Coralline algae, corals, bioerosion, sediments	2021	98	Coralline algae and corals
Kornder et al. (2018)	Corals	2016	62	None – though this study was used extensively in Cornwall et al. (2021)
Meyer and Riebesell (2015)	Coccolithophores	Unknown but 2015 or before	33	Coccolithophores (however, they often compared against 280 ppm CO <sub>2</sub> )
Harvey et al. (2013)	Calcifying algae, corals, crustaceans, echinoderms, molluscs, phytoplankton, fish, non-calcareous algae, seagrass	2012	107	All non-calcareous taxa except diatoms
Kroeker et al. (2013b)	Calcifying algae, corals, coccolithophores,	2012	155	All non-calcareous taxa



	molluscs, echinoderms, crustaceans, fish, fleshy algae, seagrasses, diatoms			
Wittmann and Pörtner (2013)	Crustaceans, echinoderms, corals, fish	2012	167	Echinoderms and molluscs
Chan and Connolly (2013)	Corals	2011	25	None
Kroecker et al. (2010)	Calcifying algae, corals, coccolithophores, molluscs, echinoderms, crustaceans, fish, fleshy algae, seagrasses, diatoms	2010	139	None
Hendriks et al. (2010)	Bivalves, corals, coccolithophores, phytoplankton, cyanobacteria	2010	59	None