Comments Reviewer 1

Minor Comments

Abstract: L19-21: Be specific, what changes are seen? In what parameter?

Added "in alkalinity"

L21-22: Not quite sure what that means.

The sentence now reads: Globally, while we see that under RCP2.6 the carbon uptake associated with AOA is only \sim 60% of the total under RCP8.5, the relative changes in temperature are larger, as are the changes in pH (140%) and aragonite saturation state (170%).

L22-23 The change in saturation state is ambiguously describe, refer specifically to changes in omega.

With respect the actual values are listed in the body of paper, and to list all of the values here would make the abstract too long.

L28 It's left a little open ended here, you could be more specific with the regional response. It is one of the more important findings from the experiment.

We would love to but given the length of the abstract we feel that we are somewhat limited in terms of space. But we have tried to be a little clearer, the last sentence of the abstract now states:

Finally, our simulated AOA for 2020-2100 in the RCP2.6 scenario is capable of offsetting warming and ameliorating ocean acidification increases at the global scale, but with highly variable regional responses.

Introduction: Good introduction. Clearly explains why we need CDR and more specifically AOA. Also gives a description of ocean acidification and how AOA works.

Thanks

L50: 'Including through coral bleaching' a little clunky, maybe remove 'through'

We have removed this statement, it now states: While warming represents an imminent global threat which is already significantly impacting the natural environment (Hughes et al., 2017), ocean acidification poses an additional and equally significant threat to the marine environment.

L54: Could you say something about the changing Revelle Factor, and the potential for AOA to impact this?

We have now added the statement to the text: As CO₂ is taken up by the ocean it changes its chemical equilibrium, reducing the carbonate ion concentration and decreasing pH, collectively known as ocean acidification.

We have also added statement later in the introduction to say:

Artificial Ocean Alkalization (AOA), through altering the chemistry of seawater, both enhances ocean carbon uptake (thereby reducing atmospheric CO₂), while at the same time reversing ocean acidification and increasing the buffering capacity of the ocean.

L148-150: simply states "impact" which could be a bit vague. Could go in further and state that they will be investigating the impact on the "carbon cycle, global surface warming (2m surface air temperature), and response and ocean acidification response to the 4 different AOA experiments under the high (RCP8.5) and low (RCP2.6) emissions scenarios." Which is stated in lines 207-210. Regarding my comment above, it is worth exploring the potential experiment space, magnitude of alkalinity addition, lo cation, emission scenario, and the resulting impacts site specific/regional/global/ open ocean/coastal etc. What parts of this picture does your model/this paper deal with, what has already been done by others, and what is left to do? are other models needed?

The goals of this study of this focus on the global response to regional and seasonal AOA, therefore we don't explore are best ways to ameliorate local conditions through AOA. Regarding question of whether to explore experimental space further – the answer is yes there is a lot of work that needs to be done (please see the review by Renforth and Henderson (2017)) and these results are put into this context in the Results and Discussion section. AOA will also have addressed as part of the Carbon Dioxide Removal Model Intercomparison Project (CDR-MIP), which we are involved in.

We have now modified the paragraph to now say:

In this work, we use a fully coupled ESM (CSIRO-Mk3L-COAL), which includes climate and carbon feedbacks, to investigate the impact of AOA on the carbon cycle, global surface warming (2m surface air temperature), and ocean acidification response to the global and regional AOA experiments under the high (RCP8.5) and low (RCP2.6) emissions scenarios

Methods:

Model seems appropriate for the scope of this paper. Clear description of the experimental design which seems appropriate to answer the research question proposed in the introduction. Could explain what the model outputs are? Also should mention the testing for seasonality? (mentioned in lines 538-554)

This model has been assessed in a number of studies already cited here and the outputs are consistent with standard Earth System Model outputs. As stated above the key outputs we consider are surface air and ocean temperature, and changes in the surface pH and aragonite saturation state. We also describe in the methods sections a number of other model prognostic variables (lines: 153-168) and provide references to individual model papers that describes components of the CSIRO Mk3L-COAL earth system model.

L162-164: Do you expect this assumption to hold up under elevated alkalinity? Could the rain ratio change?

Probably, this is already addressed in the lines 445-460, and studies suggest a small feedback.

L204: Fair assumption, but it is worth pointing out that alkalinity manipulation could be from carbonate dissolution or NaOH addition which would not induce and impact from iron and silicate. You are testing the fundamental impact intrinsic to all of these methods of C sequestration.

We agree, the sentence now reads:

We do not consider the biogeochemical response to other minerals and elements that can be associated with the sourcing of alkalinity from the application of finely ground ultra-mafic

rocks such as olivine and forsterite, nor dissolution processes required to increase alkalinity (e.g. Montserrat et al., 2017).

Results and discussion:

L208 - 209: the sentence doesn't make sense, a typo somewhere?

Yes, we have now removed and response

L215: Why have you chose this addition rate? Also, should alkalinity not be expressed in equivalents rather than moles (and throughout)?

As stated, this value is very close to that used by Keller et al (2014) following Kohler (2014), who estimated a value of AOA based on globally shipping. While this allows a comparison of simulated values and a quasi-physical value, our work is more focussed on comparing and contrasting the responses to AOA for low and high emissions to regional and seasonal AOA. Regards units we have followed the convention used by Zeebe and Wolf-Gladrow (2005) and followed Keller et al (2014) and to ensure consistency with previous work.

L218-221: Fascinating, but why was the response different?

This is now addressed in the discussion

L232: "at" is missing

Addressed

L239: I think 'an overall' is missing before 525 ppm in the brackets

Added

L241: could you also give this as a % similar to how you did for RCP8.5

This is a good comment – however. This doesn't really make sense as the atmospheric value at end is less than at the beginning.

L251-254: This is really important...why was there an increase in export?... The 1% is an important outcome because it is the 'efficiency reduction' on the overall engineering system design.

Firstly, this is a really small number (<1%) and as stated occur in the Arabian Sea. In section 3.2.3 it now reads:

The very small changes in export production in RCP2.6 were located in the Arabian Sea (not shown), likely driven by enhanced mixing in this region.

Could tables 1-3 be summarised in one table? I think it would make things a little clearer.

This is a good and we have now combined these three into a single table

Line 331: could be explained more clearly rather than just "due to the Revelle factor" (see previous comment)

We have now rewritten this section to be clearer, it now states:

In the 2020-2100 period, AOA under RCP2.6 led to much larger increases in surface pH and aragonite saturation state, more than 1.3 times, and more than 1.7 times that of RCP8.5 respectively (Table 4). These changes reflect the differences in the mean state associated with high and low emissions, specifically the difference between Alkalinity and Dissolved Inorganic Carbon (ALK-DIC), a proxy for ocean acidification (Lovenduski et al, 2015). As the values of DIC in the upper ocean are larger under RCP8.5 than RCP2.6, the difference between ALK and DIC (ALK-DIC) is smaller and the chemical buffering capacity of CO2 or Revelle Factor (Revelle and Suess, 1957) is less. This means that, for a given addition of ALK the increase in the upper ocean DIC will always be greater under RCP8.5 due to its reduced buffering capacity. Consequently, the changes in ALK-DIC with AOA are greater under RCP2.6 than RCP8.5, which translates to greater increases in pH and aragonite saturation state.

L374: 'This reflects the fact that' should be rewritten 'This is caused by the subduction processes..' or something similar.

We have now added through subduction to this statement

L396: 'Quite low', how low?

We have now removed Quite

L445-447: But could you speculate as you have in the previous sentence? How much would export have to change to make a material difference?

We would not like to speculate to do this as many of the processes are not well understood, instead we reference Matear and Lenton (2014) for a discussion of these processes and feedbacks.

L507: This doesn't quite ring with your abstract, which suggests that ocean acidification would be mitigated. Would it not partially ameliorate the impacts?

The last line of the abstract refers to low emissions, so they are consistent. Yes, it would ameliorate some of the impacts under RCP8.5, as shown, but its impacts would much less than under RCP2.6.

L599: 'Interestingly' is used a bit too often, it gets a bit jarring.

Removed

Figures:

All Figures are clear. Slightly too many for this type of manuscript. Could some be moved to the supporting information? Figures 11-12 are not referred to in the text.

We feel all figures are warranted, and we have added references to figures in the text.

Reviewer 2: Major comments:

Major comment 1: The changes in the land carbon uptake (table 2) in the AOA simulations based on the RCP2.6 are around 4 times higher than those of the simulations based on the RCP8.5. This is an important aspect because the variations in these carbon fluxes determine the final state of the climate. That is why I think that these results should be discussed properly and the cause of this differential behaviour should be explained.

These differences are due to 2 main factors: (i) the temperature differences between RCP2.6 and RCP8.5. The mean SAT cooling over land under RCP2.6 is much larger (2x) this means that the decrease in carbon uptake would be larger RCP2.6 than RCP8.5; and (ii) as seen Zhang et al (2014a), the climate sensitivity of the land carbon-climate feedback under higher emissions is lower than other models' due to nutrient (N&P) limitation. This sensitivity in part explains why the response of the land carbon cycle is about half that reported in Keller et al (2014).

The text now states:

...On the land, in the RCP8.5 simulation there was a smaller reduction in carbon uptake than in RCP2.6 (Table 1), due to larger decreases in surface air temperature (SAT) over land in RCP2.6 than RCP8.5 (~2x; see Section 3.1.2). The land carbon cycle response was also smaller under high than low emissions due to nutrient limitation being reached, thereby limiting the effect of CO₂ fertilization (Zhang et al, 2014a).

Major comment 2: The statements given between the line 285 and 289 are really confusing. On the one hand, it reads as the temperature change in the RCP8.5 experiment is higher than the one associated with the RCP2.6, which is not what I see in the numbers. And on the other hand, making reference to "potentially reflecting feedbacks" in order to explain this cooling signal does not help to understand the signal. Instead, it confuses the reader. Please explain properly how these feedbacks affect the results.

This was a mistake and has now been corrected. We have now clarified the text and removed the reference to feedbacks which was not correct, please see the comment below showing that the disparate responses are primarily due to differences in atmospheric CO₂ growth rate, please the response to Major Comment 3 (next) for more detail.

Major comment 3: The reduction in atmospheric CO2 concentrations by 2100 associated with the AOA scenarios under RCP8.5 emissions (app. 84 ppm) is higher than the one associated with the AOA scenarios conducted under the RCP2.6 (app. 40 ppm). Yet, the mitigated warming in the AOA simulations under RCP2.6 is higher than those conducted under the RCP8.5. This is one of the main findings of this publication, however, there is not any discussion/explanation of this result. Only stating what the model delivers is not enough, since it could be a model artefact, the signal might not be caused by AOA, etc. The RCP8.5 and 2.6 scenarios have atmospheres with quite different levels of CO2, which might lead to differences in the CO2-forcing response to changes in CO2 levels. Not only that, but also the RCP8.5 and 2.6 scenarios differ in the assumed land use and the sea ice extent by the end of this century. This might also cause changes in albedo and therefore in the cooling response due to changes in forcing.

There are a number of mechanisms that may explain the differential response of the cooling which is larger under RCP2.6 than RCP8.5, these including the CO₂ vs outgoing long wave radiation (OLR) log relationship, and land and ocean albedo changes. We find that while these may play a minor role, the major driver of these differences are due to differences in atmospheric CO₂ growth rate between RCP2.6 and RCP8.5.

We have now added the following statement to the text:

...In the period 2081-2100 we see larger mean changes in SAT under RCP2.6 than RCP8.5 primarily due to differences in atmospheric CO2 growth rate. Krasting et al. (2014) showed that the slower rate of emissions, the lower the radiative forcing response. This occurs in response to the timescales associated with the uptake of heat and carbon. Consequently, under RCP8.5 the atmospheric CO2 growth rate is much faster than RCP2.6, leading to a strong radiative forcing response. This explains why, despite a larger reduction in atmospheric CO2 concentration under RCP8.5, the biggest reduction in global mean SAT occur under RCP2.6...

Ref:

Krasting, J. P., Dunne, J. P., Shevliakova, E., and Stouffer, R. J.: Trajectory sensitivity of the transient climate response to cumulative carbon emissions, Geophys Res Lett, 41, 2520-2527, 10.1002/2013gl059141, 2014.

Major comment 4: Between the lines 325 and 334 an explanation to the differential pH and aragonite saturation state responses between simulations is given. This explanation seems confusing and it refers to the other main finding of this publication. Because of this I think that it requires some supporting figures (which could be added into the supplementary information) and some extra work in order to clarify the message. I suggest to look at the buffer factors and the effects of AOA under the two different DIC/ALK regimes associated with the RCP8.5 and 2.6 scenarios. More information can be found in the paper by Egleston et.al. (2010) (http://onlinelibrary.wiley.com/doi/10.1029/2008GB003407/abstract).

We have rewritten this paragraph to better capture our message make it more accessible. It now states:

In the 2020-2100 period, AOA under RCP2.6 led to much larger increases in surface pH and aragonite saturation state, more than 1.3 times, and more than 1.7 times that of RCP8.5 respectively (Table 4). These changes reflect the differences in the mean state associated with high and low emissions, specifically the difference between Alkalinity and Dissolved Inorganic Carbon (ALK-DIC), a proxy for ocean acidification (Lovenduski et al, 2015). As the values of DIC in the upper ocean are larger under RCP8.5 than RCP2.6, the difference between ALK and DIC (ALK-DIC) is smaller and the chemical buffering capacity of CO2 or Revelle Factor (Revelle and Suess, 1957) is less. This means that, for a given addition of ALK the increase in the upper ocean DIC will always be greater under RCP8.5 due to its reduced buffering capacity. Consequently, the changes in ALK-DIC with AOA are greater under RCP2.6 than RCP8.5, which translates to greater increases in pH and aragonite saturation state.

Minor comments:

L16, L27 and L561: "is capable of" gives the impression that AOA has not real big limitations to be implemented which is not the case, please modify the wording

This wording is correct, AOA is capable and is analogous to alkalinity addition that occurs over geological timescales and this has been hypothesised to play a role in glacial-interglacial timescales.

L18, L19: there are acronyms which the reader might have never seen in the abstract, please spell them out or remove

We have now spelt out CO2 and RCP

L25: "lower" and "higher" emissions than what? I think that you meant "low" and "high"

This is correct, we wanted to be more generic than just RCP 2.6 and 8.5, particularly as Shared Socioeconomic Pathways (SSPs) will be used CMIP6.

L26: our simulations show that AOA during the period ...; in any case I do not think that this very last sentence in the abstract is needed

With respect, we think that this is an important statement to make.

L46: ... could help to ...

Corrected

L53-54: CO2 that enters the ocean does not react with seawater to reduce the carbon- ate ion concentration, please reconsider this statement and use correct grammar

This has now been changed to say:

As CO₂ is taken up by the ocean it changes its chemical equilibrium, reducing the carbonate ion concentration and decreasing pH, collectively known as ocean acidification.

L59: ...changes in calcification...

Corrected

L60: are you sure that ocean acidification alters nutrient availability

Yes e.g. Shi et al (2010)

Shi et al (2010) Effect of Ocean Acidification on Iron Availability to Marine Phytoplankton, *Science* 327, 676 (2010);

L62: please change order of publications

They are already in name order

L69: semicolon needed?

Removed

L77: weathering of minerals play a crucial role in modulating the state of the climate in geological timescales, please write an assertive statement

Removed may

L91: reviewed

This should be present tense, I believe

L92-95: way too long sentence, please simplify and split it

This has now been split

L98: Did Kohler used one or several models?

Changed to singular

L110: ocean only without the hyphen

Corrected

L110: and they showed

Corrected

L111: high CO2 emission

Corrected

L114: also concluded that

This should be present tense, I believe L115, L118, L131, ...: impacts of Changed L124: from a high Changed L126: it would be required Changed L127: and it would come Changed L134: 78 ppm between brackets might look better Changed L134: a net atmospheric cooling Changed to surface air temperature

L139: "to be very large" - (very) large in what respect? please clarify $\frac{1}{2}$

The sentence now reads: Capturing these feedbacks is critical as they have the potential to significantly increase atmospheric CO₂ concentrations (Jones et al., 2016).

L141: currently assume (instead of "utilize")

Changed

L141 to L145: way too long sentence, please simplify and split it

Changed to: Furthermore, the feasibility of these approaches which are increasingly questioned due in part to limited land (Smith et al., 2016), whereas the potential CDR capacity of the oceans is orders of magnitude greater (Scott et al., 2015).

L149: and surface warming

L149: questions

L147 to L152: I think that the novelty of this study could be better emphasise. In any case, this last paragraph is crucial and therefore it should be improved since it does not read well.

The above three comments have been addressed in response to Reviewer 1 and the paragraph now reads:

In this work, we use a fully coupled ESM (CSIRO-Mk3L-COAL), which includes climate and carbon feedbacks, to investigate the impact of AOA on the carbon cycle, global surface warming (2m surface air temperature), and ocean acidification response to the global and regional AOA experiments under the high (RCP8.5) and low (RCP2.6) emissions scenarios.

L158: extra dot after citations?

Changed

L160, L163, ...: please remove the brackets in those citations which are subjects of the sentences, this occurs several times in this manuscript

Changed

L164 to L166: does this sentences really add any relevant information? Such a feature of the model is basic to conduct this study

Yes, we agree but it does provide confidence in the tool we are employing in this study – as seen by Review 1's comments

L171 to L172: the land carbon cycle currently has too many uncertainties to state something in such an assertive manner, please consider to modify this or even remove it

We have removed *realistically*

L185: from 2006 onwards, ...

Corrected

L186: corresponding to the Representative ...

Corrected

L218: Subpolar addition

Corrected

L232: is seen in 2100

Corrected

L233 to L235: if you mention this feature of the modelling tool, please explain the associated consequences for the simulations of AOA

We have discussed these results in the manuscript and identify why the sensitivity of the land carbon uptake, particularly under high emissions is less than other studies.

L242 and L256: "more than compensates" and "more than offset" are really confusing ways of describing the obtained values, please clarify

We have removed *more than* in both of these instances

L247: 50% instead of 1.5 maybe?

Corrected

L253: total ocean uptake ...

Corrected

L254 to L258: this explanation reads really confusing, please clarify

This has now been changed to say:

The simulated cooling drove both a reduced net primary production, leading to reduced carbon uptake, and an increase in carbon retention associated with a reduction in heterotrophic respiration. However, overall, the net decrease in land carbon uptake means that in the response to AOA globally the reduced net primary production dominated.

L266: addition studies such as Ilyina ... which demonstrated ...

Corrected

L270: 181 PgC is in ... (instead of was)

Corrected

L277: I think the authors meant "positive denotes enhanced uptake" (instead of "negative")

Yes – this is corrected thank you

L288: "large" twice in the sentence

Corrected

L294: "projected" instead of "anticipated"

Corrected

L295: why is this publication here cited?

Removed

L298: standard deviations with respect to what? what is this (1 - sigma)? Please clarify

The caption has now been improved, and now reads:

Table 1 The differences in global mean surface air temperature in the period 2081-2100

(2090) and associated standard deviation (1- σ) (K; SAT; 2m) for the four different AOA

experiments for each emission scenario, relative to the same emission scenario with no AOA.

. L302 to L304: please consider to reformulate these sentences since "variability" might refer

to many different things (e.g. inter annual, inter model, model internal, ...). In any case I think

that "variability" is not really the term to use since what is described here are differences

between simulations.

The sentence now reads:

Within each of the scenarios, there are some differences in the magnitude of the cooling

within the four different AOA experiments; however, these are smaller than the interannual

variability over the last two decades of the simulations.

L308: mean surface cooling

Corrected

L318: What is the point of this statement and citation? The pH and aragonite saturation state

correlate really well as I can see in the figures.

Yes but the impacts are different and this motivates why we are interested looking at both

aragonite saturation state and pH.

L321: despite the return

Corrected

L344 and L377: the citation here to Groeskamp et.al. seems unfounded

Removed

L348 and L349: Please elaborate on this so that the reader understand the context, e.g.

discuss how this change in pH might (or not) matter, ...

We have modified this sentence to now say:

To put these changes into context, the estimated decrease in pH since the preindustrial period is 0.1 units (Raven et al., 2005), and is responsible for already detectable changes in the marine environment (Albright et al., 2016).

L380: How can one of the experiments (AOA_ST) reflect the timescales of the circulation of the subtropical gyres? Please explain this.

This sentence now reads:

In the case of AOA_ST, this reflects the timescales associated with the longer residence time of upper ocean waters in the subtropical gyres.

L382: ice covered (instead of "non-ice-free")

Changed

L386: by 2100 (instead of "in")

Corrected

L387 and L388: please clarify this, is not understandable

This is now clarified to say:

Specifically, for AOA_G we see 31% remains in the upper ocean and for AOA_T and AOA_ST: 34%, while for AOA_SP: 22-24% remains in the upper ocean which (as anticipated) is lower than in other regions.

L404: ...seen in the...

Corrected

L421 to L425: please work on the grammar of these sentences

Corrected

L442: ... in the ratio ... Corrected L444: Dot missing Added L445: ... remain poorly... Corrected L457 to L459: why do you obtain this result? We have added: *likely driven by enhanced mixing in this region*. L463: remove (SAT) Removed

L468 to L470: why do you obtain this result?

This very much reflects the period over which the mean changes were calculated, and the simulated large variability in SAT in this region, which now stated in the text.

L538 to L554: why no figures are shown in this section on seasonality to support this discussion? Also, only AOA is implemented in the summer season under RCP8.5 emissions, which does not seem to me enough to explore the effects of seasonality.

This is a little confusing as the Reviewer wishes us to remove figures and now requests more. We don't present results here, as it is a sensitivity experiment, rather than a major result.

L560: please remove (COP21)

Removed

L593 to L595: What do you mean? Please clarify this.

It now states:

However, AOA under the RCP2.6 emissions scenario changes the roles played by the ocean and land in carbon uptake as compared with the scenario of RCP2.6 with no AOA, resulting in a reduced uptake in the terrestrial biosphere and increased uptake in the ocean.

L605: double "that"

Corrected

L620: mention "preindustrial period" and remove (1850) reads better

Changed

L621: cases (subject?) leads ...

Corrected

L633: for the role

Corrected

L638: ...therefore it needs ...

Added

L642 and L645: Earth system (instead of earth system)

Corrected

L649: please put "e.g. mesocosm experiments" between brackets

Added

Please keep an eye on the format in which the references are given and be consistent with it.

Done

Comments Reviewer 3

The Lenton et al., study investigates the impacts from adding artificial alkalinity to the oceans using the model CSIRO under 2 different emission pathways - RCP2.6 and RCP8.5. It was a really well done and interesting study to read technically, however my main comment is that the way the paper is currently structured makes it confusing to read. For example, each paragraph jumps back and forth between RCP26 and RCP85 making the story line hard to follow. I suggest setting up the story for one of the emissions pathways and then comparing to that one for the other pathway. It would also be useful to set up the chemistry in a little more detail or reiterate the paragraph in the intro. This would be useful when explaining why adding alk under a 2.6 scenario is more effective.

At this stage, we do not feel that major rewrite or reordering of the paper is warranted. It is clear from other studies is that AOA will reduce OA and global warming; what is more interesting is whether the response to the same amount of AOA differs between emissions scenarios. This is the main focus of the study hence it does not make sense to restructure the paper as suggested. Instead we have gone through the paper to ensure that it is clearer and easier to follow.

It would also be useful to set up the chemistry in a little more detail or reiterate the paragraph in the intro. This would be useful when explaining why adding alk under a 2.6 scenario is more effective.

Please see the response to Reviewer 2.

Lastly, section 3.1 was confusing (you may want to expand on the methods section to make this section clearer). For each run you added 0.25Pmol/yr of alkalinity but then I read in ln216 that the magnitude of the increase in alkalinity is dependent on where it was added. Is the 0.25Pmol/yr added to all the boxes? or is it divided up between the boxes for a total of 0.25Pmol/yr? Can you put everything in the same units to be constant?

We apologise for any ambiguity and have now clarified this section, it now states:

For each emissions scenario, we simulated four different AOA experiments, which all had the same 0.25 Pmol/yr of alkalinity added. In the case of the regional experiments the per surface values were larger than the case of global addition.

Minor comments: ln50: "including through coral bleaching" - not clear what this means

We have now removed this statement.

ln79-80: This sentence seems out of place.

We have now removed this sentence.

ln149: what do you mean by impact?

We have now been more explicit and the sentence now says:

In this work, we use a fully coupled ESM (CSIRO-Mk3L-COAL), which includes climate and carbon feedbacks, to investigate the impact of AOA on the carbon cycle, global surface warming (2m surface air temperature), and ocean acidification response to the global and regional AOA experiments under the high (RCP8.5) and low (RCP2.6) emissions scenarios.

ln158: extra period between feedbacks and references

Corrected

ln230: the first sentence does not make sense.

Rewritten it now states:

The large atmospheric CO₂ concentration at 2100 under RCP8.5 reflects the large projected increase in emissions during this century, while under RCP2.6 a similar atmospheric concentration of CO₂ is seen in 2100 as at the beginning of the simulation (2020) (Figure 2a).

ln250: why is there a difference in export?

The text now reads:

...Consistent with Keller et al. (2014) and Hauck et al. (2016) the simulated changes in ocean export production were very small (~0.2 PgC) under RCP8.5 and due to small changes in

ocean state, e.g. stratification. Under RCP2.6, it was slightly larger at 1.2 PgC, but still less

than 1% percent of the total ocean uptake increase simulated under AOA, due to small

changes in ocean state in a more stratified ocean...

Section 3.1.2: I don't understand how soil-moisture feedbacks are influencing temperature in

this case. Temperature is more variable on land than over the ocean, could it be simply that?

We apologise for the confusion we have removed this section and attribute these changes to

the differences different in atmospheric CO2 growth rate.

The section now states:

...In the period 2081-2100 we see larger mean changes in SAT under RCP2.6 than RCP8.5

primarily due to differences in atmospheric CO2 growth rate. Krasting et al. (2014) showed

that the slower rate of emissions, the lower the radiative forcing response. This occurs in

response to the timescales associated with the uptake of heat and carbon. Consequently,

under RCP8.5 the atmospheric CO2 growth rate is much faster than RCP2.6, leading to a

strong radiative forcing response. This explains why, despite a larger reduction in

atmospheric CO₂ concentration under RCP8.5, the biggest reduction in global mean SAT

occur under RCP2.6...

Ref:

Krasting, J. P., Dunne, J. P., Shevliakova, E., and Stouffer, R. J.: Trajectory sensitivity of the

transient climate response to cumulative carbon emissions, Geophys Res Lett, 41, 2520-2527,

10.1002/2013gl059141, 2014.

Table4: relative to what?

It now reads:

Table 4 The differences in surface value of aragonite saturation state and pH between the

AOA experiments for each emission scenarios in 2100 relative to the emissions scenario with

no AOA.

Figures: relative to what?

We have added text to each of the captions to clarify

Figures: There are a lot of figures which I'm not sure add much to main text. You could simply the story in the text and only look at a few and toss the rest into the supplement. Or could you combine the 2.6 and 8.5 figures into 1? Could you do a difference between them? That would visually show the reader where the differences between scenarios lie....

While this seems attractive, we think that there is value in keeping these figures. Furthermore, we do not see a simply way of combining these into 8 panel figures, nor does doing the differences make much sense, as differences of differences is quite confusing.

Section 3.2.5: I think the figures referenced are not the correct figures. Plus I think a description of figures 11 and 12 are missing.

Thank you for this – we have now ensured that the figures are referenced correctly, and switched the order to better reflect the order they are appear in the text.

I see that the previous reviewers picked up a bunch of typos

Corrected.

1 Assessing Carbon Dioxide Removal Through Global and Regional Ocean Alkalization 2 under High and Low Emission Pathways. 3 Lenton A.^{1,2}, Matear R.J.¹, Keller D.P.³, Scott V.⁴, and Vaughan N.E.⁵ 4 5 ¹ CSIRO Oceans and Atmosphere, Hobart, Australia ² Antarctic Climate and Ecosystems Co-operative Research Centre, Hobart, Australia 6 ³ GEOMAR Helmholtz Centre for Ocean Research, Kiel, Germany 7 8 ⁴ School of Geosciences, University of Edinburgh, Edinburgh, United Kingdom 9 ⁵Tyndall Centre for Climate Change Research, School of Environmental Sciences, University 10 of East Anglia, Norwich, UK. 11 1. Abstract 12 Atmospheric Carbon Dioxide (CO₂) levels continue to rise, increasing the risk of severe 13 14 impacts on the Earth system, and on the ecosystem services that it provides. Artificial 15 Ocean Alkalization (AOA) is capable of reducing atmospheric CO₂ concentrations, and surface warming and addressing ocean acidification. Here, we simulate global and 16 17 regional responses to alkalinity (ALK) addition (0.25 PmolAlkPmolALK/year) over the 18 period 2020-2100 using the CSIRO-Mk3L-COAL Earth System Model in the period 19 2020-2100, under high (Representative Concentration Pathway 8.5; RCP8.5) and low 20 (RCP2.6) emissions. While regionally there are large changes in alkalinity associated 21 with locations of AOA, globally we see only a very weak dependence on where and when 22 AOA is applied. WeGlobally, while we see that under RCP2.6, while the carbon uptake 23 associated with AOA is only ~60% of the total under RCP8.5, the relative changes in temperature are larger, as are the changes in pH $(\frac{1.4x}{140})$ and aragonite saturation 24 25 (1.7x). state (170%). The results of this modelling study are significant as they 26 demonstrate that simulations reveal AOA is more effective under lower emissions, 27 and therefore the higher the emissions the more AOA is required to achieve the same 28 reduction in global warming and ocean acidification. -Finally, our simulations 29 showsimulated AOA in the period for 2020-2100 in the RCP2.6 scenario is capable of 30 offsetting global warming and ameliorating ocean acidification increases due to low 31 emissions at the global scale, but regionally the response is more with highly variable. 32 regional responses. 33 34

35

37	Atmospheric carbon dioxide (CO ₂) levels continue to rise primarily as a result of human
38	activities. Recent studies have suggested that even deep cuts in emissions may not be
39	sufficient to avoid severe impacts on the Earth system, and the ecosystem services that it
40	provides (Gasser et al., 2015). Recent international negotiations (UNFCCC, 2015) agreed to
41	limit global warming to well below 2°-, °C. The application of Carbon Dioxide Removal
42	(CDR), sometimes referred to as "Negative Emissions", appears to be required to achieve this
43	goal, as emission reductions alone are likely to be insufficient (Rogelj et al., 2016). In this
44	context, there is an urgent need to assess how Carbon Dioxide Removal (CDR)CDR could
45	help either mitigate climate change or even reverse it, and to understand the potential risks
46	and benefits of different options.
47	
48	While warming represents a majoran imminent global threat, including through coral
49	bleaching which is already significantly impacting the natural environment (Hughes et al.,
50	2017), ocean acidification poses an additional and equally significant threat to the marine
51	environment. Ocean acidification occurs as CO2 taken up by the ocean reacts with the
52	seawater to reduce the carbonate ion concentration and decrease the pH. AnnuallyAt present
53	the oceans take up about 28% of anthropogenic CO_2 emitted annually (Le Quéré et al., 2015).
54	As CO ₂ is taken up by the ocean it changes its chemical equilibrium, reducing the carbonate
55	ion concentration and decreasing pH, collectively known as ocean acidification. Furthermore,
56	as the ocean continues to take up carbon the buffering capacity or Revelle Factor (Revelle
57	and Suess, 1957) of the seawater decreases, thereby accelerating the rate of ocean
58	acidification.
59	
60	Ocean acidification is the unavoidable consequence of rising atmospheric CO2 levels and will
61	impact the entire marine ecosystem — from plankton at the base, through to fishhigher-
62	trophic species at the top. Potential -impacts include changes toin calcification, fecundity,
63	organism growth and physiology, species composition and distributions, food web structure
64	and nutrient availability (Doney et al., 2012; Dore et al., 2009; Fabry et al., 2008; Iglesias-
65	Rodriguez et al., 2008; Munday et al., 2010; Munday et al., 2009); Fabry et al., 2008; Iglesias-
66	Rodriguez et al., 2008; Munday et al., 2010; Munday et al., 2009). Within this century, the
67	impacts of ocean acidification will increase in proportion to emissions (Gattuso et al., 2015).

36

1. Introduction

68 Furthermore, these changes will be long-lasting, persisting for centuries or longer even if 69 emissions were are halted (Frolicher and Joos, 2010). 70 71 To date, many different CDR techniques have been proposed both on the land and in the 72 ocean (Royal Society, 2009; National Research Council, 2015). Their primary purpose is to 73 reduce atmospheric CO₂ levels; and thus most CDR methods will amelioratealso reduce the 74 impacts of ocean acidification, although some proposed techniques such as ocean pipes 75 (Lovelock and Rapley, 2007) and micro-nutrient addition (Keller et al., 2014) may actually 76 lead to ana regional acceleration of ocean acidification in surface waters. 77 78 Artificial Ocean Alkalization (AOA), through altering the chemistry of seawater, both 79 enhances ocean carbon uptake, (thereby reducing atmospheric CO_{27}), while at the same time 80 directly reversing ocean acidification, and increasing the buffering capacity of the ocean. 81 AOA can be thought of as a massive acceleration of the natural processes of chemical 82 weathering of minerals that may have played a role in modulating the climate on geological 83 timescales (Zeebe, 2012; Colbourn et al., 2015); Sigman and Boyle, 2000). Alkalinity 84 changes may also have played an important role in controlling glacial-interglacial cycles of 85 atmospheric CO₂ e.g. Sigman and Boyle (2000). 86 87 Specifically, as alkalinity enters the ocean, the pH increases leading to an elevated carbonate 88 ion concentration, a reduction in the hydrogen ion concentration and a decrease in the concentration of aqueous CO2 (or pCO2). This in turn enhances the disequilibrium of CO2 89 between the ocean and atmosphere (or $\Delta pCO_2 = pCO_2^{\text{ocean}} - pCO_2^{\text{atmosphere}} pCO_2^{\text{atmosphere}}$) leading 90 91 to increased ocean carbon uptake, and <u>a</u> reduction in <u>the</u> atmospheric CO₂ concentration. 92 These increases in pH and carbonate ion concentration thus reverse the ocean acidification 93 due to uptake of anthropogenic CO₂. 94 95 Kheshgi (1995) first proposed AOA as a method of CDR. Renforth and Henderson (2017) 96 review the early experimental, engineering and modelling work undertaken to investigate 97 AOA. From the observational perspective, we draw particular attention to the experimental 98 work of Albright et al. (2016) which provided an in situ demonstration of localised AOA to 99 offset the observed changes in ocean acidification on the Great Barrier Reef that have 100 occurred since the pre-industrial period.

101 102 Several modelling studies have explored the impacts of AOA both on carbon sequestration 103 and ocean acidification. Using ocean—only biogeochemical models, Kohler et al. (2013) 104 explored AOA via olivine addition. Olivine, in addition to increasing alkalinity also adds iron 105 and silicic acid, both of which can enhance ocean productivity (Jickells et al., 2005; 106 Ragueneau et al., 2000). Kohler et al. (2013) estimated the response of atmospheric CO₂ 107 levels and pH to different levels of olivine addition over the period 2000-2010, and later-this 108 was <u>later</u> extended to 2100 by Hauck et al. (2016). These studies demonstrate a global 109 impact, that appeared appears to scale with the amount of olivine added. Importantly, Kohler 110 et al. (2013) showed that the global effect of alkalinity added along shipping routes (as an 111 analogue for practical implementation) was not significantly different from that of alkalinity 112 added in a highly idealized uniform manner. 113 114 Ilyina et al. (2013) explored the potential of AOA to mitigate rising atmospheric CO₂ levels 115 and ocean acidification in ocean-only biogeochemical simulations, and they showed that 116 AOA has the potential to ameliorate future changes due to high <u>CO</u> emissions. They did not 117 limit the amount of AOA, as their goal was to offset the projected future changes; and 118 showed that the amount of AOA required to do this would drive the carbonate system to 119 levels well above pre industrial preindustrial levels. Ilyina et al. (2013) also conclude that 120 local AOA could potentially be used to offset the impactimpacts of ocean acidification, with 121 enhanced CO₂ uptake being only a side benefit. This regional approach was explored further 122 by Feng et al. (2016) who suggested that local AOA in the tropical ocean, in areas of high 123 coral calcification, has the potential to offset the impactimpacts of future rising atmospheric 124 CO₂ levels under a high emissions scenario (RCP8.5). This study also revealed strong 125 regional sensitivities in the response of ocean acidification related to the locations in which it 126 was applied. 127 128 To date several Several other studies estimate have estimated the response of the Earth system 129 to AOA. Gonzalez and Ilyina (2016) used an Earth System Model (ESM) to estimate the 130 AOA required to reduce atmospheric concentrations from a high emissions scenario 131 (RCP8.5) to the medium emissions scenario (RCP4.5). They estimated that to mitigate the 132 associated 1.5K warming difference, via reducing atmospheric CO₂ concentrations by ~400 133 ppm, would require an addition of 114 Pmol of alkalinity (between 2018-2100), would be

134	required, and it would come at the cost of very large (unprecedented) changes in ocean
135	chemistry.
136	
137	Keller et al. (2014) used an Earth System Model of Intermediate Complexity (EMIC) to
138	explore the impactimpacts of AOA over the period 2020-2100, to arising from a globally
139	uniform addition of alkalinity (0.25 PmolALK/yr), an amount based on the estimated
140	carrying capacity of global shipping following Kohler et al. (2013). Keller et al. (2014)
141	showed that AOA led to <u>a</u> reduction in atmospheric CO_2 of 166 PgC (or \sim 78 ppm), a
142	net surface air temperature cooling of 0.26K and a global increase in ocean pH of 0.06 in the
143	period 2020-2100.
144	
145	To date, not all modelling studies have been emissions driven, and this is important as
146	potential climate and carbon cycle feedbacks may not have been accounted for. Capturing
147	these feedbacks is critical as they have the potential to be very large significantly increase
148	atmospheric CO ₂ concentrations (Jones et al., 2016). Further, no studies have explored the
149	impact of AOA under low emissions scenarios such as RCP2.6. This is important because
150	scenarios that limit warming to 2° C or less, currently utilizeassume considerable land-based
151	CDR via afforestation and/or Bio-Biomass Energy with Carbon Capture and Storage
152	(BECCS). Furthermore, the feasibility of which are these approaches is increasingly
153	questioned due in part to limited land (Smith et al., 2016), whereas the potential CDR
154	capacity of the oceans is orders of magnitude greater (Scott et al., 2015).
155	
156	In this work, using we use a fully coupled Earth System Model ESM (CSIRO-Mk3L-COAL),
157	which includes climate and carbon feedbacks, weto investigate the impact of AOA on the
158	carbon cycle, global surface warming (2m surface air temperature), and the ocean
159	acidification, land and ocean carbon uptake and warming. Specifically, the question this
160	study tackles is: What is the impact of global response to the global and regional AOA on
161	the Earth System, and how efficient it is at mitigating global warming and ocean
162	acidification experiments under the high (RCP8.5) and low (RCP2.6) emissions
163	trajectories?scenarios.
164	
165	2. Methods
166	2.1 <u>2.1 Model Description</u>

167	The model simulations were performed using the CSIRO-Mk3L-COAL (Carbon, Ocean,
168	Atmosphere, Land) Earth System Model ESM which includes climate-carbon interactions and
169	feedbacks- (Matear and Lenton, 2014; Zhang et al., 2014a) The ocean component of the
170	Earth System ModelESM has a resolution of 2.8° by 1.6° with 21 vertical levels. The ocean
171	biogeochemistry is based on (Lenton and Matear, (2007;) and Matear and Hirst, (2003)
172	simulating the distributions of phosphate, oxygen, dissolved inorganic carbon and alkalinity
173	in the ocean. The-model simulates particulate inorganic carbon (PIC) production as <u>a</u>
174	function of particulate organic carbon (POC) production- via the rain ratio (9%) following
175	(Yamanaka and Tajika-, (1996) This ocean biogeochemical model was shown to simulate
176	the observed distributions of total carbon and alkalinity in the ocean (Matear and Lenton,
177	2014) and phosphate (Duteil et al., 2012).
178	
179	The atmosphere resolution is 5.6° x 3.2° with 18 vertical layers. The land surface scheme
180	uses CABLE (Best et al., 2015) coupled to CASA-CNP (Wang et al., 2010;; Mao et al.,
181	2011) which simulates biogeochemical cycles of carbon, nitrogen and phosphorus in plants
182	and soils. The response of the land carbon cycle was shown to realistically simulate the
183	observed biogeochemical fluxes and pools on the land surface (Wang et al., 2010).
184	
185	To quantify the changes in ocean acidification, we calculate pH changes on the total scale
186	following the recommendation of Riebesell et al. (2010). To calculate the changes of
187	carbonate saturation state, we use the equation of Mucci (1983).
188	
189	2.2 Model Experimental Design
190	Our ESM was spun-up under a preindustrial atmospheric CO_2 concentration of 284.7 ppm,
191	until the simulated climate was stable (> 2000 years) (Phipps et al., 2012). From the spun-up
192	initial climate state, the historical simulation (1850 - 2005) was performed using the
193	historical atmospheric CO ₂ concentrations as prescribed by the CMIP5 simulation protocol
194	(Taylor et al., 2012).
195	
196	Following the historical concentration pathway from 2006 onward, two different future
197	projections to 2100 were made using the atmospheric CO_2 emissions corresponding to the
198	Representative Concentration Pathways of low emissions (RCP2.6) and high emissions
199	(RCP8.5 or 'business as usual') (Taylor et al., 2012). All simulations include the forcing due

200 to non-CO₂ greenhouse gas concentrations (Taylor et al., 2012). We define RCP8.5 and 201 RCP2.6 as our control cases for the corresponding experiments below. 202 203 In the period 2020-2100, we undertook a number of AOA experiments using a fixed quantity 204 of 0.25 Pmol/yr of alkalinity, the same a similar amount used by Keller et al. (2014). 205 Consistent with this study, we applied AOA in the surface ocean all year-round in ice-free 206 regions, set to be between 60°S and 70°N. (note that this ignores the presence of seasonal 207 sea-ice in some small regions). For each of the two emissions scenarios, we considered four 208 different regional applications of AOA, shown in Figure 1. These are: (i) AOA globally 209 (AOA_G) between 60S60°S and 70N70°N; (ii) the higher latitudes comprising the subpolar 210 northern hemisphere oceans (40N -70N40-70°N) and the (ice-free) Southern Ocean (40S-211 60S40-60°S) (AOA_SP); (iii) the subtropical oceans (15-40N40°N and 15S and 40S; 15-212 40°S) (AOA_ST); and (iv) in the equatorial regions (15N 15S; 15°N-15°S) (AOA_T). In this 213 study, we only look at the response of the Earth system to alkalinity injection. We do not 214 consider the biogeochemical response to other minerals and elements that can be associated 215 with the proposed sourcing of alkalinity from the application of finely ground ultra-mafic 216 rocks such as olivine and fosterite forsterite, nor dissolution processes required to increase 217 alkalinity (e.g. Montserrat et al. (... 2017). 218 219 3. Results and Discussion 220 To aid in presenting our results and to compare these with previous studies, we first discuss 221 the carbon cycle, global surface warming (2m surface air temperature), and response and 222 ocean acidification response to the 4four different AOA experiments under the high (RCP8.5) 223 and low (RCP2.6) emissions scenarios. We then look at the regional behaviour of the 224 simulations in the different AOA experiments. 225 226 3.1 Global Response 227 For each emissions scenario, we simulated four different AOA experiments, which 228 all had the same 0.25 Pmol/yr of alkalinity added. In the case of the regional experiments the 229 per surface values were larger than the case of global addition. As anticipated, by 2100, AOA 230 increased the global mean surface ocean alkalinity relative to the corresponding scenario 231 control case, with the magnitude of the increase in alkalinity being dependent on where it was

added (Table 1). PolarSub-polar addition (AOA_SP) led to the smallest net increase in

232

surface alkalinity, while tropical addition (AOA_T) produced the greatest increase. As expected, the global mean changes in surface alkalinity between emissions scenarios are very small (less than 3 μ mol/kg difference). The slightly greater increase in surface values in alkalinity under RCP8.5 likely reflects enhanced ocean stratification under higher emissions (Yool et al., 2015).

2	3	S
~	J	·

		RCP8.5	RCP2.6		
	,	AOA_G- RCP	<u>AOA_SP</u> 108.3	<u>AOA_ST</u> 105.	<u>AOA_T</u>
(a) Relativ	ve increase in	global mean oce	ean surface alkalir	nity (µmol/kg) in	<u>n 2100</u>
RCP8.5A	OA_SP	108.3	79.7	<u>115.1</u> 74.4	<u>129.8</u>
RCP2.6A	OA_ST -	115 105.1	<u>74.4</u>	112.9	127.1
(b) Total i	ntegrated add	ditional carbon u	ptake (in PgC) in	the period 2020	<u>0-2100</u>
	<u>Total</u>	<u>178.6</u>	183.3	180.7	<u>174.5</u>
RCP8.5	<u>Ocean</u>	<u>184.4</u>	<u>188.1</u>	<u>185.1</u>	<u>177.2</u>
	Land	<u>-5.8</u>	<u>-4.8</u>	<u>-4.4</u>	<u>-2.7</u>
DCD2 (<u>Total</u>	<u>121.1</u>	<u>122.1</u>	<u>122.0</u>	<u>116.0</u>
<u>RCP2.6</u>	<u>Ocean</u>	<u>143.1</u>	<u>145.2</u>	<u>143.1</u>	<u>139.2</u>
AOA_T- RCP Land 129.8		127 -22.1	<u>-24.1</u>	<u>-21.2</u>	<u>-23.1</u>
			ir temperature in	the period 2081	<u>'-2100 (2090)</u>
and ass			-σ) (K; SAT; 2m)		
	<u>Total</u>	<u>-0.16±0.08</u>	<u>-0.13±0.10</u>	<u>-0.08±0.05</u>	<u>-0.14±0.06</u>
<u>RCP8.5</u>	<u>Ocean</u>	<u>-0.14±0.07</u>	<u>-0.11±0.07</u>	<u>-0.06+0.03</u>	<u>-0.12±0.05</u>
	Land	-0.22±0.15	-0.18±0.20	-0.13±0.14	<u>-0.19±0.11</u>
	<u>Total</u>	-0.25±0.08	-0.23±0.08	-0.20±0.09	<u>-0.16±0.06</u>
<u>RCP2.6</u>	<u>Ocean</u>	-0.19±0.05	-0.18±0.05	<u>-0.15±0.06</u>	<u>-0.13±0.05</u>
	Land	-0.39±0.22	-0.35±0.22	-0.30±0.20	-0.24±0.16

Table 1 For the two RCP scenarios, (a) the relative increase in global mean ocean surface alkalinity (μ mol/kg) between each AOA experiment and control experiment in 2100. (b) The total integrated additional carbon uptake (in PgC) in the period 2020-2100 in different experiment and emissions scenarios, positive denotes enhanced uptake. (c) The differences in global mean surface air temperature in the period 2081-2100 (2090) and associated standard deviation (1- σ) (K; SAT; 2m) for the four different AOA experiments for each emission scenario, relative to the same emission scenario with no AOA.

247	
248 249	3.1.1 Carbon Cycle
250	The large atmospheric CO ₂ concentration at the end of the century 2100 under RCP8.5
251	reflects the large projected increase in emissions projected (under RCP8.5),during this
252	century, while under RCP2.6 a similar atmospheric concentration of CO ₂ is seen in 2100 as a
253	the beginning of the simulation (2020) ($\frac{\text{Figure 2a}}{\text{Figure 2a}}$). We note that atmospheric CO ₂
254	levels in our CSIRO-MK3L-COAL for the control cases are greater than for their respective
255	concentration driven RCPs due to nutrient limitation in the land, leading to reduced carbon
256	uptake (Zhang et al., 2014a)—.
257	
258	Under all emissions scenarios and experiments, AOA leads to reduced atmospheric CO ₂
259	levels (Figure 2a). Under RCP8.5, AOA reduces atmospheric concentration by 82-86 ppm-,
260	this represents; representing a ~16% decrease in atmospheric concentration (525 ppm
261	increase over the period 2020-2100) In contrast to RCP8.5, AOA under RCP2.6 leads to a
262	smaller reduction in atmospheric concentration (53-58 ppm)Figure 2a shows that, by the
263	end of the century, AOA more than compensates for the projected increase in atmospheric
264	CO ₂ due to RCP2.6.
265	
266	Over the 2020-2100 period, the reduction in atmospheric CO ₂ levels associated with AOA is
267	primarily due to increased ocean carbon uptake, offset by small decreases in the land surface
268	carbon uptake (Table 21). In the ocean, RCP8.5 has leads to much greater net uptake than
269	RCP2.6, about 1.5 times 50% more, due to the larger (and growing) disequilibrium between
270	the atmosphere and ocean.
271	
272	In the ocean, the relative increase in carbon uptake in response to AOA is primarily abiotic in
273	nature. Consistent with Keller et al. (2014) and Hauck et al. (2016) the simulated changes in
274	ocean export production were very small (~0.2 PgC) under RCP8.5. While under and due to
275	small changes in ocean state, e.g. stratification. Under RCP2.6, it was slightly larger at 1.2
276	PgC, but still less than 1% percent of the total ocean <u>uptake</u> increase simulated under AOA.
277	and due to small changes in ocean state in a more stratified ocean. In contrast, the relative
278	decreases in land carbon uptake were biotic in nature. The simulated cooling over land drove
279	both a reduction in reduced net primary production that more than offset the decrease in .
280	leading to reduced carbon flux due in uptake, and an increase in carbon retention associated

with a reduction in heterotrophic respiration. However, overall, the net decrease in land carbon uptake means that in the response to AOA globally the reduced net primary production dominated. On the land, in the RCP8.5 simulated simulation there was a smaller reduction in carbon uptake than in RCP2.6, potentially due to a (Table 1), due to larger decreases in surface air temperature (SAT) over land in RCP2.6 than RCP8.5 (~2x; see Section 3.1.2). The land carbon cycle response was also smaller relative cooling (Table 3 under high than low emissions due to nutrient limitation being reached, thereby limiting the effect of CO₂ fertilization (Zhang et al, 2014a).

-For both emissions scenarios, the 4<u>four</u> AOA experiments all produced similar reductions in atmospheric CO₂ concentrations (Figure 2) with less than a 5% difference in the total land and ocean carbon uptake. The global changes in land and ocean carbon uptake <u>doare</u> not appear to be very sensitive to where we add the alkalinity to the surface ocean. This is consistent with Kohler et al. (2013) who saw little difference in adding olivine along existing shipping tracks, versus uniformly adding it to the surface ocean. It is also consistent with regional addition studies <u>of</u> Ilyina et al. (2013) <u>and</u>, Feng et al. (2016) <u>who and Feng et al</u> (2017) which demonstrated a global impact.

Our simulated total increased carbon uptake under AOA_G with RCP8.5 (179 PgC) is comparable to the 166 PgC reported by (Keller et al., (2014). Their cumulative increase in ocean carbon uptake by 2100 of 181 PgC wasis in very good agreement with our value of 184 PgC. However, they simulated a reduction in land uptake nearly twice the -5.8 PgC reduction in our AOA_G simulation. These differences may reflect both the lower sensitivity of the simulated climate feedbacks in our ESM, and differences in land surface models.

	Total	Ocean	Land	Total	Ocean	Land
	RCP8.5	RCP8.5	RCP8.5	RCP 2.6	RCP 2.6	RCP 2.6
AOA_G RCP	178.6	184.4	-5.8	121.1	143.1	-22.1
AOA_SP-	183.3	188.1	-4.8	122.1	145.2	-24.1
RCP						
AOA_ST	180.7	185.1	4.4	122.0	143.1	21.2
RCP						
AOA_T RCP	174.5	177.2	2.7	116.0	139.2	-23.1

306 307 Table 2 The total integrated additional carbon uptake (in PgC) in the period 2020-2100 in 308 different experiment and emissions scenarios, negative denotes enhanced uptake. 309 310 3.1.2 Surface Air Temperature 311 In the control simulations, the global mean surface air temperature (SAT; 2m) increased in 312 the period 2020-2100 with RCP2.6 simulating a net warming of 0.4±0.1K while RCP8.5 313 warmed by 2.7±0.1K (2081-2100). All AOA experiments simulated a reduction in global 314 mean SAT relative to their corresponding control simulation (Figure 2b). Within each 315 emissions scenario the global mean SAT decline associated with AOA is always greater and 316 more variable over the land than ocean (Table 3). While the mean cooling, in the period 317 2081-2100, is also greater over the land under RCP8.5 than RCP2.6, potentially reflecting 318 feedbacks such as soil moisture (Seneviratne et al., 2010) snow and ice cover changes. 319 However, these changes are associated with large interannual large variability, and therefore 320 not significantly differentAOA experiments simulated a reduction in global mean SAT 321 relative to their corresponding control simulation (Figure 2b). Within each emissions scenario 322 the global mean SAT decline associated with AOA is always greater and more variable over 323 the land than ocean (Table 1). In the period 2081-2100 we see larger mean changes in SAT 324 under RCP2.6 than RCP8.5 primarily due to differences in atmospheric CO₂ growth rate. 325 Krasting et al. (2014) showed that the slower rate of emissions, the lower the radiative 326 forcing response. This occurs in response to the timescales associated with the uptake of heat 327 and carbon. Consequently, under RCP8.5 the atmospheric CO₂ growth rate is much faster 328 than RCP2.6, leading to a strong radiative forcing response. This explains why, despite a 329 larger reduction in atmospheric CO₂ concentration under RCP8.5, the biggest reduction in 330 global mean SAT occur under RCP2.6. These mean changes are also associated with large 331 interannual variability. 332 333 Under RCP2.6, all the AOA experiments keep warming levels much close to values in 2020 334 than RCP2.6 by the end of this century (2100; Figure 2b). In contrast, under the RCP8.5 335 scenario, none of the AOA experiments have a significant impact on the projected warming 336 by the end of this century (less than 10%) reflecting the large warming anticipated under high 337 emissions (Rogeli et al., 2012). 338

	Total	Ocean	Land	Total	Ocean	Land
	RCP8.5	RCP8.5	RCP8.5	RCP 2.6	RCP 2.6	RCP 2.6
AOA_G-	-0.16±0.08	-0.14±0.07	-0.22±0.15	-0.25±0.08	-0.19±0.05	-0.39±0.22
RCP						
AOA_SP	-0.13±0.10	-0.11±0.07	-0.18±0.20	-0.23±0.08	-0.18±0.05	-0.35±0.22
RCP						
AOA_ST	-0.08±0.05	-0.06+0.03	-0.13±0.14	-0.20±0.09	-0.15±0.06	-0.30±0.20
RCP						
AOA_T	-0.14±0.06	-0.12±0.05	-0.19±0.11	-0.16±0.06	-0.13±0.05	-0.24±0.16
RCP						

Table 3 The differences in global mean surface air temperature and their standard deviations $(1-\sigma)$ (K; SAT; 2m) for 2090 (in the period 2081-2100) for different AOA experiments and emissions scenarios relative to the emissions scenarios with no AOA.

Under RCP2.6, all the AOA experiments keep global warming levels much closer to values in 2020 than RCP2.6 by the end of this century (2100; Figure 2b). In contrast, under the RCP8.5 scenario, none of the AOA experiments have a significant impact on the projected warming by the end of this century (less than 10%) reflecting the large warming projected under high emissions.

Within each of the scenarios, there <u>isare</u> some <u>variabilitydifferences</u> in the magnitude of the cooling within the <u>4four</u> different AOA experiments; however, these <u>differences</u> are smaller than the interannual variability over the last <u>2two</u> decades of the simulations. Therefore, it appears that the global mean SAT decline with AOA is not very sensitive to where the alkalinity is added under either emission scenario.

The global mean cooling associated with AOA_G under RCP8.5 (-0.16±0.08K; 2081;-2100) is close to the mean surface air temperature decreasecooling of -0.26K reported by (Keller et al., (2014) for the same similar levels of AOA. These differences may reflect the simplified atmospheric representation of the UVIC Intermediate Complexity Model and different climate sensitivities.

362 3.1.3 Ocean Acidification 363 364 Here, we quantify changes in ocean acidification in terms of pH and aragonite saturation state 365 changes. –We consider these two diagnostics because they are associated with different 366 biological impacts and are not necessarily well correlated (Lenton et al., 2016). In the future, 367 the global mean changes in pH and aragonite saturation state will be proportional to the 368 emissions trajectories following Gattuso et al. (2015), with the largest changes associated 369 with the higher emissions (RCP8.5) (Figure 2c-d). By 2100, despite ather return to 2020 370 values of atmospheric CO₂ concentration under RCP2.6 (Figure 2), neither pH ornor 371 aragonite saturation state return to 2020 values, consistent with Mathesius et al. (2015). 372 373 In the 2020-2100 period, AOA under RCP2.6 led to much larger increases in surface pH and 374 aragonite saturation state, more than 1.3 times, and more than 1.7 times that of RCP8.5 375 respectively (Table 42). These changes reflect the differences in the mean state associated 376 with high and low emissions, specifically the difference between Alkalinity and DIC. 377 The Dissolved Inorganic Carbon (ALK-DIC), a proxy for ocean acidification (Lovenduski et 378 al, 2015). As the values of DIC in the upper ocean are larger under RCP8.5 than RCP2.6, and 379 therefore the difference between ALK- and DIC (ALK-DIC) is higher. For smaller and the 380 chemical buffering capacity of CO₂ or Revelle Factor (Revelle and Suess, 1957) is less. This 381 means that, for a given addition of alkalinity, ALK the increase in the upper ocean DIC will 382 <u>always</u> be greater in the high emission case than the low emission case under RCP8.5 due to 383 the Revelle Factor (Revelle and Suess, 1957). its reduced buffering capacity. Consequently, 384 the difference between Alkalinity and changes in ALK-DIC with AOA increases less in the 385 high emission scenario than the low scenario are greater under RCP2.6 than RCP8.5, which 386 translates into smaller to greater increases in pH and aragonite saturation state in the high 387 scenario. 388 389 While there was a significant difference in pH and aragonite saturation state changes with 390 AOA between high and low emissions cases, the global mean changes for different AOA 391 experiments within each scenario isare quite similar (Table 4). The 2), the exception being 392 the AOA_SP experiment, where its the pH and aragonite saturation state changes are only 393 ~75% of the change in the other AOA experiments. This reduced change in the polar region 394 is consistent with the smaller changes in the surface ocean alkalinity values associated with 395 AOA SP (Table 1). These differences at higher latitudes reflect the enhanced subduction of

alkalinity away from the surface ocean into the ocean interior that occurs in the high latitude oceans (Groeskamp et al., 2016).

AOA_G under RCP8.5 leads to a relative increase in pH of 0.06, which is consistent with (Keller et al., ...(2014), while our the relative increase in aragonite saturation state (0.28) is also very close to their simulated value (0.31). To put these changes into context, the estimated decrease in pH since the preindustrial period is 0.1 units (Raven et al., 2005), and is responsible for already detectable changes in the marine environment (Albright et al., 2016).

	Aragonite	pН	Aragonite	pН
	RCP8.5	RCP8.5	RCP2.6	RCP
				<u>2RCP2</u> .6
AOA_G -RCP	0.28	0.06	0.50	0.07
AOA_SP-	0.20	0.05	0.39	0.07
RCP				
AOA_ST-	0.30	0.06	0.54	0.08
RCP				
AOA_T -RCP	0.28	0.06	0.5	0.07

Table 42 The relative differences in surface value of aragonite saturation state and pH between the AOA experiments and the high and low emissions for each emission scenarios in 2100 relative to the emissions scenario with no AOA.

3.2 Regional Responses

For both RCP scenarios, there are large regional differences in the relative surface changes in alkalinity, temperature, and ocean acidification associated with the different AOA experiments. -The regional nature of these changes is closely associated with where alkalinity addition is applied, and the two different emissions scenarios considered here do not differ significantly in their behaviour. This implies that any differences in stratification and overturning circulation between the two scenarios are insufficient todo not significantly modulatealter the response to AOA.

420 3.2.1. Surface Alkalinity 421 422 For both scenarios, the greatest surface alkalinity changes occur where the alkalinity is added 423 (Figure 3). Spatially, under either emission scenario, the relative differences in 2090 are very 424 similar; consequently, we only show the changes under RCP2.6 (Figure 3). The only 425 significant differences occur in the Arctic, reflecting larger longer-term changes in alkalinity 426 projected under higher emissions (Yamamoto et al., 2012). 427 428 Overall, the greatest increases are seen in the tropical ocean (AOA_T) suggesting this is the 429 most efficient region in retaining the added alkalinity in the upper ocean. -This reflects the 430 fact that subduction processes in the tropical ocean are less efficient than in other regions 431 such as the higher latitudes. In the The (ice-free) subpolar oceans (AOA_SP) produced the 432 smallest relative increase in alkalinity, and this reflects the strong and efficient surface to 433 interior connections through subduction occurring at higher latitudes (Groeskamp et al., 434 2016). The global mean relative increase associated with AOA in the subtropical gyres 435 (AOA_ST) and globally (AOA_G) fall between the tropical (AOA_T) and higher latitude 436 (AOA_SP) values. In the case of AOA_ST, this reflects the time scalestimescales associated 437 with <u>eirculation</u>the <u>longer residence time</u> of <u>upper ocean waters in</u> the subtropical gyres. 438 439 The most modest relative increase in alkalinity occurs in the non-ice-freecovered regions 440 where alkalinity is not explicitly added. Interestingly, even when alkalinity is added in the 441 very high latitude Southern Ocean, it is carried northward by the Ekman current 442 explainingwhich explains the very modest increase in the region where AOA occurs between 443 50S to 60S. In terms of the total alkalinity added to the surface ocean, about one-third 444 remains in the upper 200m inby 2100. (Figure 4). Specifically, for AOA G we see 31% 445 remains, in the upper ocean, and for AOA_T and AOA_ST: 34%,% remains in the upper 446 ocean, while for AOA_SP: the figure is 22-24% remains, which (as anticipated) is lower than 447 in other regions. 448 449 Spatially, AOA in the higher latitude regions (AOA_SP) leads to very large relative increases 450 in alkalinity (> 1000 \(\mu\)mol/kg; 2090) occurring along the northern most boundary of the 451 Northern Subpolar Gyres, particularly the North Pacific. Clearly, in this region the rate of 452 AOA exceeds the rate of subduction allowing alkalinity to build up. Large relative increases 453 in alkalinity also occur in the Southern Ocean under AOA_SP, particularly along Western

454 Boundary Currents. However, in contrast to northern high latitudes the values still remain 455 quite low suggesting that the rate of addition does not exceed the rate of subduction even 456 under the highest emission scenario. 457 458 AOA_ST shows a large relative increase of ~300 µmol/kg (2081-2100) in the 459 subpolarsubtropical gyre regions. Overall, we find that these relative increases are quite 460 homogenous across the entire subtropical gyres, with strong mixing with tropical waters 461 leading to significant relative increases in tropical Atlantic, Western Pacific and Indian 462 Oceans. -Within the tropical ocean, under AOA T the largest relative changes are found 463 across the entire tropical Indian Ocean (~400 µmol/kg) with large relative increases also 464 seen in the Indonesian seas (~280 µmol/kg; 2081-2100). Away from the tropical Indian 465 Ocean, we find that relatively homogenous increases occur in the Western Pacific and the 466 Atlantic, with much more modest relative increases in the Eastern Pacific reflecting the 467 dominant East to West upper ocean circulation. Consistent with the response of AOA_ST, 468 AOA T leads to relative increases in surface alkalinity that are consistent with the response 469 to AOA ST – in the AOA_ST region of \sim 130 μ mol/kg (2081-2100). 470 471 In the case of AOA_G₂ a relatively uniform net increase in alkalinity occurs in all regions 472 with the exception of the upwelling regions such as the tropical Pacific, which showed a 473 more modest relative increase. In AOA_G there is little evidence of any of the very large 474 increases in alkalinity seen in the more regional AOA experiments. This spatial pattern of 475 relative increase is broadly consistent with the pattern of global alkalinity increase simulated 476 by Ilyina et al. (2013) and Keller et al. (2014) for AOA in the (ice-free) global ocean. 477 478 3.2.2 Changes in the interior distribution of alkalinity in the global ocean 479 480 As only about 30% of the total AOA remains in the upper 200m, we explore the fate of this 481 alkalinity in the interior ocean in the zonal sections of alkalinity (Figure 4). As the pattern is 482 very similar, between RCP2.6 and RCP8.5, we only show RCP2.6, noting that in the North 483 Atlantic where the projected ocean stratification is stronger under higher emissions (not 484 shown) leading to slightly decreased subsurface values. This increased stratification is 485 consistent with other studies (e.g. (Yool et al., 2015) Yool et al., 2015). 486

487 Unlike the surface plots of AOA, the relative increases in <u>subsurface</u> alkalinity due to AOA 488 are very similar across all experiments. This heterogeneous <u>spatial</u> pattern of alkalinity 489 increase is associated with water entering the interior ocean along specific surface to interior 490 pathways (Groeskamp et al., 2016). Specifically, we see alkalinity moving. Alkalinity also 491 moves into the interior ocean along the poleward boundaries of the subtropical gyres, 492 associated with the formation and subduction of mode waters, and an increase in the 493 subtropical gyres associated with large-scale downwelling, and deep mixing in the North 494 Atlantic. The changes in alkalinity are mainly found in the upper ocean (<\frac{1000 m}{1000m}) 495 which reflects the relatively short period of alkalinity addition. Given the short period, this is 496 analogous to present-day observed distributions of anthropogenic carbon (Sabine et al., 497 2004). 498 499 As the changes in export production are very small, the large changes in the interior alkalinity 500 concentrations primarily reflect the physical transport, rather than the sinking and 501 remineralization of calcium carbonate. -Clearly other biological processes, not represented in 502 our model, have the potential to impact the surface and interior values of alkalinity (Matear 503 and Lenton, 2014). One such process is the reduction in the (rain) ratio of PIC:POC under 504 higher emissions (Riebesell et al., 2000) however. However, it has been shown that even a 505 very large reduction in PIC production (50%) would not significantly impact our results 506 (Heinze, 2004). Unfortunately, at present, the magnitude and sign of many of these other 507 feedbacks remainsremain poorly known (Matear and Lenton, 2014); consequently, 508 quantifying their impact on our results is very difficult, and beyond the scope of this study. 509 510 3.2.3 Ocean Carbon Cycle Response 511 512 The similarity in global ocean carbon uptake associated with all AOA experiments for a 513 given emission scenario hides the large spatial differences between simulations. Given that 514 the largest carbon cycle response occurs in the ocean (Table 21), we focus on this response 515 for RCP8.5 and RCP2.6 (Figures 5 and 6). -As expected, ocean carbon uptake is strongly 516 enhanced in the regions of AOA. Away from regions of AOA, there is a reduction in carbon 517 uptake, associated with the weakening of the gradient in CO₂ between the atmosphere and 518 ocean due to AOA. Interestingly, the largest increase spatially occurs in the Southern Ocean 519 under AOA_SP for RCP2.6, while in contrast the largest changes under RCP8.5 occursoccur

in the tropical ocean under AOA_SPT. The very small changes in export production in

521 RCP2.6 were located in the Arabian Sea (not shown), and while likely driven by enhanced mixing in this region. While these changes are <-1% of the total change in carbon uptake, 522 523 they may nevertheless they maybe important regionally. 524 525 3.2.4 Temperature (SAT) 526 527 The decrease in global mean SAT associated with all AOA experiments for a given emission 528 scenario again hides the large spatial differences between the simulations. -The response of 529 surface temperature is spatially very heterogeneous (Figures 7 and 8), and while the regional 530 surface temperature changes are very similar between the two emissions scenarios. The 531 exception to this is the Arctic which did not show a consistent response across the different 532 AOA experiments-, reflecting the period over which the mean changes were calculated, and 533 the simulated large variability in SAT in this region. Under both emission scenarios, the 534 largest cooling associated with AOA occurs over Northern Russia and Canada, and 535 Antarctica (greater than a -1.5K cooling) with a larger cooling in these regions under RCP2.6. 536 537 In the surface ocean, AOA in the RCP2.6 scenario shows brings about a net cooling overof 538 the surface ocean, with the exception of the North Atlantic, east of New Zealand, and off the 539 southern coast of Alaska, which show a very a-modest warming. A similar pattern is evident 540 in RCP8.5; however, there is a greater cooling in the high latitudes, and less cooling in the 541 lower latitudes than under RCP2.6. 542 543 3.2.5 Ocean Acidification Response 544 545 Globally, the response of pH and aragonite saturation state associated with AOA are similar. 546 however, large spatial and regional differences are present. (Figures 9-14). To aid in the 547 interpretation of changes in aragonite saturation state, overlain on the aragonite saturation 548 state maps (Figures 9 and 10) are the contours corresponding to the value of 3, the 549 approximate threshold for suitable coral habitat (Hoegh-Guldberg et al., 2007). On these 550 surface maps and subsequent section plots (Figures 13 and 14) we plot the saturation 551 horizon, i.e. the contour corresponding to the transition from chemically stable to unstable (or 552 corrosive), i.e. aragonite saturation state is equal to 1 (Orr et al., 2005).

554 The largest relative changes in pH and aragonite saturation state were associated with regions 555 of AOA₇ (Figures 9-12), reflecting increases in the surface values of alkalinity- (Figure 3). 556 All simulations increase pH and aragonite saturation state in the Arctic despite no direct 557 addition in this region, with the largest changes here associated with AOA_G and AOA_SP. 558 Interestingly, all simulations show little to no increase in the high latitude Southern Ocean, 559 consistent with more efficient transport of the added alkalinity into the ocean interior. 560 561 The changes in pH associated with AOA experiments under RCP8.5, while spatially very 562 different, particularly when added in the subpolar ocean, are still much less than the 563 decreases associated with RCP8.5 with no AOA- (Figure 9). In terms of aragonite saturation 564 state (Figure 10), the conditions for coral growth in the tropical ocean remain very 565 unfavorable unfavourable by the end of century (i.e. aragonite saturation state <-3) under all 566 regional and global experiments, with the exception of AOA_T, where only a very small 567 region in the Central Pacific Ocean exhibits suitable conditions. 568 569 Consistent with Feng et al. (2016), we find that this level of AOA under RCP8.5 is 570 insufficient to ameliorate or significantly alter the large-scale changes in ocean acidification. 571 More positively, at the higher latitudes the saturation horizon is moved poleward with the 572 largest shift associated with AOA_SP, and the smallest shift at the high latitudes occurring 573 under AOA_T. Consistent with these changes, we see a deepening of the saturation horizon 574 everywhere (Figure 13), and little difference spatially between AOA experiments, consistent 575 with zonal mean changes in alkalinity for the 4four AOA experiments. (Figure 11). 576 577 The spatial pattern of changes associated with AOA under RCP2.6 are broadly consistent 578 with those that seen under higher emissions; however, the magnitude of the response is much 579 larger <u>—</u> again, due to the larger differences between Alkalinity and DIC with AOA under 580 RCP2.6- (Figures 12 and 13). In terms of aragonite saturation state, the area of tropical ocean 581 favourable for corals is considerably expanded suggesting that conditions for tropical coral 582 growth are improved under AOA.. As anticipated the largest changes in the area favourable 583 for tropical corals is associated with AOA_T, closely followed by AOA_ST. -As the 584 saturation horizon does not reach the surface under RCP2.6, we can only look at the changes 585 in the interior ocean. Here, there is a deepening in the saturation horizon in all experiments of 586 a very similar magnitude in all experiments (Figure 14), with the exception of the Arctic.

587 Here, the response of the saturation horizon is more sensitive to the location of the AOA, 588 varying between ~100m under AOA_T and ~280 m under AOA_SP- (Figure 14). 589 590 Spatially, the large changes in ocean acidification in response to AOA under RCP2.6 more 591 than compensate for the changes in ocean chemistry due to low emissions in the period 2020-592 2100. Globally, in the changes in the period 2020-2100 are sufficient to reversed reverse or 593 compensate for the changes since the preindustrial period (1850). -However, spatially in some 594 regions such as equatorial upwelling, an important area of global fisheries (Chavez et al., 595 2003), AOA in fact leads to higher values of aragonite saturation state and pH than the ocean experienced in the preindustrial period (Feely et al., 2009). -We can only speculate on the 596 597 potential impact on marine biota throughof a reduction in aqueous CO₂ and elevated pH 598 levels in these regions. For- a recent review of the potential impact of rising pH and 599 Aragonite aragonite saturation state on marine organisms, we direct the reader to Renforth and 600 Henderson (2017). 601 602 3.2.6 Importance of Seasonality 603 604 In this paper, while we have focused on year-round AOA, as a sensitivity experiment we also explored whether AOA added in summer or winter was more efficient. To do this, we 605 606 focused on the higher latitudes regions where the largest seasonal changes in mixing are 607 found (de Boyer Montegut et al., 2004; Trull et al., 2001). -Here, we tested whether AOA in 608 either summer or winter was more effective than year-round addition. To test this for 609 RCP8.5, we add alkalinity only during the summer at half of the annual rate (or 610 0.125PmolALK/year) in the AOA_SP region. 611 612 Our results showed that the response to AOA in summer was very close to 50% of the 613 response of the year-round addition associated with AOA_SP (or 0.25PmolALK/year). This 614 suggests that the response of AOA appears invariant with regard to when the alkalinity is 615 added. This also suggests, consistent with published studies (e.g. (Keller et al., 2014; Feng et 616 al., 2016; Kohler et al., 2013), that the response of the ocean to different quantities of AOA 617 is scalable under the same emissions scenario. Whether this is true under very much larger 618 additions of alkalinity, as simulated by (Gonzalez and Ilyina, (2016), is less clear. 619

620	4. Summary and Concluding Remarks
621	
622	Integrated Assessment Modelling for the Intergovernmental Panel on Climate Change shows
623	that CO ₂ removal (CDR) may be required to achieve the goal of limiting warming to well
624	below 2° (COP21)C (Fuss et al., 2014). Of the many schemes that have been proposed to
625	limit warming, only Artificial Ocean Alkalization (AOA) is capable of both reducing the rate
626	and magnitude of global warming through reducing atmospheric CO2 concentrations, while
627	simultaneously directly addressing ocean acidification. Ocean acidification, while often
628	receiving-often less attention, is likely to have very long lasting and damaging impacts on the
629	entire marine ecosystem, and the ecosystem services it provides.
630	
631	Here, for the first time, we investigate the response of a fully coupled climate ESM, (i.e. one
632	that accounts for climate-carbon feedbacks,) to a fixed addition of alkalinity
633	(0.25PmolALK/year) under high (RCP8.5) and low (RCP2.6) emissions scenarios to a fixed
634	addition of alkalinity (0.25PmolALK/year) We explore the effect of global and regional
635	application of AOA focusing on the subpolar gyres, the subtropical gyres and the tropical
636	ocean. To assess AOA, we look at changes in surface air temperature, carbon cycling and
637	ocean acidification (aragonite saturation state and pH) in the period 2020-2100.
638	
639	Consistent with other published studies, we see that AOA leads to reduced atmospheric CO_2
640	concentrations, cooler global mean surface temperatures, and reduced levels of ocean
641	acidificationGlobally, for these metrics we observed that they do not vary significantly
642	between the various AOA experiments under each emissions scenario. This implies that at
643	thisthe global scale there is little sensitivity of the global responses to the region where AOA
644	is applied. We also investigate as a sensitivity experiment adding alkalinity in different
645	seasons and see little difference in response to when AOA was undertaken.
646	
647	We see under AOA that the increased carbon uptake is dominated by the oceanUnder
648	RCP8.5, the changes due to AOA are only capable of reducing atmospheric concentrations by
649	a maximum of 86 ppm versus the projected change of 560ppm, 16 % and, as such, the
650	response of the climate system remains strongly dominated by warming. This is consistent
651	with published studies of the response of the climate system under RCP8.5, and studies that
652	have estimated the amount of AOA required to counteract a high emissions trajectory.

653 In contrast, AOA under RCP2.6 — while only capable of reducing atmospheric CO₂ levels by 654 655 58 ppm₅ – is sufficient to reduce atmospheric CO₂ concentrations and warming to close to 656 2020 levels at the end of the century. This is significant as it suggests that, in combination 657 with a rapid reduction in emissions, AOA could make an important contribution to the goal to 658 keepof keeping the rise in global mean temperatures below 2°. However, AOA under the 659 RCP2.6 does not ameliorate spatialemissions scenario changes of the roles played by the 660 ocean and land in carbon uptake associated as compared with the scenario of RCP2.6 with no 661 AOA, resulting in a reduced uptake in the terrestrial biosphere and increased uptake in the 662 ocean. This highlights that, while the atmospheric CO₂ and warming may be reversible, the 663 response of individual components of the Earth Systemsystem to different CDR may not be 664 (Lenton et al., 2017). 665 666 Interestingly despite Despite the impact of AOA on the atmospheric CO₂ concentration under 667 RCP2.6 being only ~60% of the value theimpact under RCP8.5, we see much larger changes 668 in ocean acidification associated with RCP2.6 than RCP8.5₇ more than 1.3 times in pH and 669 more than 1.7 times in aragonite saturation state. This reflect reflects the larger reductions of in 670 the difference between ALK and DIC that occurs under RCP2.6. -We also see larger relative 671 decreases in global temperature associated with RCP2.6. These results are very important as 672 they demonstrate that that AOA is more effective under lower emissions in reducing ocean 673 acidification and global warming under lower emissions. 674 675 While there is little sensitivity in the global responses to the region in which AOA is applied, 676 spatially the largest changes in ocean acidification (and ocean carbon uptake) were seen in 677 the regions where AOA was applied. -Despite large changes regionally, these cannot 678 compensate for the large changes associated with RCP8.5. Even targeted AOA in the tropical 679 ocean can preserve only a tiny area of the ocean conducive to healthy coral growth; and even 680 then the concomitant large warming is likely to be a stronger influence on coral growth than 681 ocean chemistry (D'Olivo and McCulloch, 2017). 682 683 In contrast, AOA under RCP2.6 is more than capable of ameliorating the projected ocean 684 acidification changes in the period 2020-2100. We see that, in all cases, the area of the 685 tropical ocean suitable for healthy coral growth expands, with the largest changes are

686 associated with tropical addition (AOA_T).- In some areas, such as the equatorial Pacific, the 687 changes that have occurred since the preindustrial (1850) period are also completely 688 compensated reversed, and in some cases, leads to higher values that are higher of aragonite 689 saturation state and pH than were experienced in the preindustrial period. 690 691 While the amount of alkalinity added in this study is small in comparison to other published 692 studies, the challenge of achieving even this level of AOA should not be underestimated. 693 Indeed, it is not clear whether such an effort is even feasible given the cost, and the logistical, 694 political and engineering challenges of producing and distributing such large quantities of 695 alkaline material (Renforth and Henderson, 2017). In the case of RCP8.5, it is unlikely that 696 this level of AOA could be justified given our results. If emissions can be reduced along an 697 RCP2.6 type trajectory, this study suggests that AOA is much more effective and may 698 provide a method to remove atmospheric CO₂ to complement mitigation, albeit with some 699 side-effects, and may be an alternative to reliance on land--based CDR. 700 701 In this work, and other published studies to date, we have not accounted for the role of the 702 mesoscale in AOA. In the real ocean (mesoscale), eddies are ubiquitous, and associated with 703 strong convergent and divergent flows, and mixing that plays an important role in ocean 704 transport (Zhang et al., 2014b). It is plausible that the mesoscale, and indeed fine-scale 705 circulation in the coastal environment (e.g. (Mongin et al., 2016a; Mongin et al., 2016b) 706 would, may modulate the <u>local</u> response to AOA and this therefore needs to be considered in 707 future studies. 708 709 Furthermore, this is a single model study, and the results of this work need to be tested and 710 compared in other models. The Carbon Dioxide Removal Model Intercomparison Project 711 (CDR-MIP) was created to coordinate and advance the understanding of CDR in the 712 earthEarth system (Lenton et al., 2017). CDR-MIP brings together Earth Systemsystem 713 models of varying complexity in a series of coordinated multi-model experiments, one of 714 which is a global AOA experiment (C4) (Keller et al., 2017 accepted). This will allow the 715 response of the earth Earth system to AOA to be further explored and quantified in a robust 716 multi-model framework, and will examine important further questions such as including 717 cessation effects of alkalinity addition, and the long-term fate of additional alkalinity in the

718 ocean. -In parallel, more process and observational studies (e.g. mesocosm experiments,) are 719 needed to better understand the implications of AOA. 720 721 5. Acknowledgments 722 D. P. Keller acknowledges funding received from the German Research Foundation's Priority 723 Program 1689 "Climate Engineering" (project CDR-MIA; KE 2149/2-1). The authors declare 724 that there are no conflicts of interest. The model code, simulations and scripts used in this 725 study are available by contacting Andrew Lenton (andrew.lenton@csiro.au). -We also wish 726 to also to thank Tom W. Trull for his helpful comments on this manuscript.

728 **6.** BiblographyBibliography

- Albright, R., Caldeira, L., Hosfelt, J., Kwiatkowski, L., Maclaren, J. K., Mason, B. M.,
- Nebuchina, Y., Ninokawa, A., Pongratz, J., Ricke, K. L., Rivlin, T., Schneider, K., Sesboue,
- M., Shamberger, K., Silverman, J., Wolfe, K., Zhu, K., and Caldeira, K.: Reversal of ocean
- acidification enhances net coral reef calcification, Nature, 531, 362—, 10.1038/nature17155,
- 733 2016.
- Best, M. J., Abramowitz, G., Johnson, H. R., Pitman, A. J., Balsamo, G., Boone, A., Cuntz,
- 735 M., Decharme, B., Dirmeyer, P. A., Dong, J., Ek, M., Guo, Z., Haverd, V., Van den Hurk, B.
- J. J., Nearing, G. S., Pak, B., Peters-Lidard, C., Santanello, J. A., Stevens, L., and Vuichard,
- 737 N.: The Plumbing of Land Surface Models: Benchmarking Model Performance, J
- 738 Hydrometeorol, 16, 1425-1442, 10.1175/JHM-D-14-0158.1, 2015.
- Chavez, F. P., Ryan, J., Lluch-Cota, S. E., and Niquen, M.: From anchovies to sardines and
- back: Multidecadal change in the Pacific Ocean, Science, 299, 217-221, Doi
- 741 10.1126/Science.1075880, 2003.
- Colbourn, G., Ridgwell, A., and Lenton, T. M.: The time scale of the silicate weathering
- negative feedback on atmospheric CO2, Global Biogeochemical Cycles, 29, 583-596,
- 744 10.1002/2014GB005054, 2015.
- 745 D'Olivo, J. P., and McCulloch, M. T.: Response of coral calcification and calcifying fluid
- composition to thermally induced bleaching stress, Scientific reports, 7, Artn 2207
- 747 10.1038/S41598-017-02306-X, 2017.
- de Boyer Montegut, C., Madec, G., Fischer, A. S., Lazar, A., and Iudicone, D.: Mixed layer
- depth over the global ocean: An examination of profile data and a profile-based climatology,
- J Geophys Res-Oceans, 109, Artn C12003,10.1029/2004jc002378, 2004.
- 751 10.1029/2004je002378, 2004.
- 752 Doney, S. C., Ruckelshaus, M., Duffy, J. E., Barry, J. P., Chan, F., English, C. A., Galindo,
- H. M., Grebmeier, J. M., Hollowed, A. B., Knowlton, N., Polovina, J., Rabalais, N. N.,
- 754 Sydeman, W. J., and Talley, L. D.: Climate Change Impacts on Marine Ecosystems, Annu
- 755 Rev Mar Sci, 4, 11-37, Doi 10.1146/Annurev-Marine-041911-111611, 2012.
- 756 Dore, J. E., Lukas, R., Sadler, D. W., Church, M. J., and Karl, D. M.: Physical and
- biogeochemical modulation of ocean acidification in the central North Pacific, P Natl Acad
- 758 Sci USA, 106, 12235-12240, Doi 10.1073/Pnas.0906044106, 2009.
- Duteil, O., Koeve, W., Oschlies, A., Aumont, O., Bianchi, D., Bopp, L., Galbraith, E.,
- Matear, R., Moore, J. K., Sarmiento, J. L., and Segschneider, J.: Preformed and regenerated
- phosphate in ocean general circulation models: can right total concentrations be wrong?,
- 762 Biogeosciences, 9, 1797-1807, 10.5194/bg-9-1797-2012, 2012.
- Fabry, V. J., Seibel, B. A., Feely, R. A., and Orr, J. C.: Impacts of ocean acidification on
- marine fauna and ecosystem processes, Ices J Mar Sci, 65, 414-432, Doi
- 765 10.1093/Icesjms/Fsn048, 2008.
- Feely, R. A., Doney, S. C., and Cooley, S. R.: Ocean Acidification: Present Conditions and
- Future Changes in a High-CO2 World, Oceanography, 22, 36-47, Doi
- 768 10.5670/Oceanog.2009.95, 2009.
- Feng, E. Y., Keller, D. P., Koeve, W., and Oschlies, A.: Could artificial ocean alkalinization
- protect tropical coral ecosystems from ocean acidification?, Environ Res Lett, 11, Artn
- 771 074008, doi:10.1088/1748-9326/11/7/074008, 2016.
- Feng, E. Y., Koeve, W., Keller, D. P. and Oschlies, A, Model-Based Assessment of the CO₂
- 773 Sequestration Potential of Coastal Ocean Alkalinization. Earth's Future, 5, 1252–1266,
- 774 doi:10.1002/2017EF000659, 2017

- Frolicher, T. L., and Joos, F.: Reversible and irreversible impacts of greenhouse gas
- emissions in multi-century projections with the NCAR global coupled carbon cycle-climate
- 778 model, Clim Dynam, 35, 1439-1459, Doi 10.1007/S00382-009-0727-0, 2010.
- Fuss, S., Canadell, J. G., Peters, G. P., Tavoni, M., Andrew, R. M., Ciais, P., Jackson, R. B.,
- Jones, C. D., Kraxner, F., Nakicenovic, N., Le Quere, C., Raupach, M. R., Sharifi, A., Smith,
- P., and Yamagata, Y.: Commentary: Betting on Negative Emissions, Nat Clim Change, 4,
- 782 850-853, 2014.
- Gasser, T., Guivarch, C., Tachiiri, K., Jones, C. D., and Ciais, P.: Negative emissions
- physically needed to keep global warming below 2 degrees C, Nat Commun, 6, Artn 7958
- 785 10.1038/Ncomms8958, 2015.
- Gattuso, J.-P., Magnan, A., Billé, R., Cheung, W. W. L., Howes, E. L., Joos, F., Allemand,
- D., Bopp, L., Cooley, S. R., Eakin, C. M., Hoegh-Guldberg, O., Kelly, R. P., Pörtner, H.-O.,
- Rogers, A. D., Baxter, J. M., Laffoley, D., Osborn, D., Rankovic, A., Rochette, J., Sumaila,
- 789 U. R., Treyer, S., and Turley, C.: Contrasting futures for ocean and society from different
- anthropogenic CO2 emissions scenarios, Science, 349, 10.1126/science.aac4722, 2015.
- Gonzalez, M. F., and Ilyina, T.: Impacts of artificial ocean alkalinization on the carbon cycle
- and climate in Earth system simulations, Geophys Res Lett, 43, 6493-6502,
- 793 10.1002/2016GL068576, 2016.
- Groeskamp, S., Lenton, A., Matear, R., Sloyan, B. M., and Langlais, C.: Anthropogenic
- carbon in the oceanSurface to interior connections, Global Biogeochemical Cycles, 30, 1682-
- 796 1698, 10.1002/2016GB005476, 2016.
- Hauck, J., Kohler, P., Wolf-Gladrow, D., and Volker, C.: Iron fertilisation and century-scale
- reflects of open ocean dissolution of olivine in a simulated CO2 removal experiment, Environ
- 799 Res Lett, 11, Artn 024007, doi:10.1088/1748-9326/11/2/024007, 2016.
- Heinze, C.: Simulating oceanic CaCO3 export production in the greenhouse, Geophys. Res.
- 801 Let., 31, 2004.
- Hoegh-Guldberg, O., Mumby, P. J., Hooten, A. J., Steneck, R. S., Greenfield, P., Gomez, E.,
- Harvell, C. D., Sale, P. F., Edwards, A. J., Caldeira, K., Knowlton, N., Eakin, C. M., Iglesias-
- Prieto, R., Muthiga, N., Bradbury, R. H., Dubi, A., and Hatziolos, M. E.: Coral reefs under
- rapid climate change and ocean acidification, Science, 318, 1737-1742,
- 806 10.1126/science.1152509, 2007.
- Hughes, T. P., Kerry, J. T., Alvarez-Noriega, M., Alvarez-Romero, J. G., Anderson, K. D.,
- Baird, A. H., Babcock, R. C., Beger, M., Bellwood, D. R., Berkelmans, R., Bridge, T. C.,
- Butler, I. R., Byrne, M., Cantin, N. E., Comeau, S., Connolly, S. R., Cumming, G. S., Dalton,
- 810 S. J., Diaz-Pulido, G., Eakin, C. M., Figueira, W. F., Gilmour, J. P., Harrison, H. B., Heron,
- 811 S. F., Hoey, A. S., Hobbs, J. P. A., Hoogenboom, M. O., Kennedy, E. V., Kuo, C. Y., Lough,
- J. M., Lowe, R. J., Liu, G., Cculloch, M. T. M., Malcolm, H. A., Mcwilliam, M. J., Pandolfi,
- J. M., Pears, R. J., Pratchett, M. S., Schoepf, V., Simpson, T., Skirving, W. J., Sommer, B.,
- Torda, G., Wachenfeld, D. R., Willis, B. L., and Wilson, S. K.: Global warming and recurrent
- mass bleaching of corals, Nature, 543, 373-+, 10.1038/nature21707, 2017.
- Iglesias-Rodriguez, M. D., Halloran, P. R., Rickaby, R. E. M., Hall, I. R., Colmenero-
- Hidalgo, E., Gittins, J. R., Green, D. R. H., Tyrrell, T., Gibbs, S. J., von Dassow, P., Rehm,
- 818 E., Armbrust, E. V., and Boessenkool, K. P.: Phytoplankton calcification in a high-CO2
- world, Science, 320, 336-340, Doi 10.1126/Science.1154122, 2008.
- 820 Ilyina, T., Wolf-Gladrow, D., Munhoven, G., and Heinze, C.: Assessing the potential of
- calcium-based artificial ocean alkalinization to mitigate rising atmospheric CO2 and ocean
- 822 acidification, Geophys Res Lett, 40, 5909-5914, 10.1002/2013GL057981, 2013.
- Jickells, T. D., An, Z. S., Andersen, K. K., Baker, A. R., Bergametti, G., Brooks, N., Cao, J.
- J., Boyd, P. W., Duce, R. A., Hunter, K. A., Kawahata, H., Kubilay, N., laRoche, J., Liss, P.
- 825 S., Mahowald, N., Prospero, J. M., Ridgwell, A. J., Tegen, I., and Torres, R.: Global iron

- connections between desert dust, ocean biogeochemistry, and climate, Science, 308, 67-71,
- 827 Doi 10.1126/Science.1105959, 2005.
- Jones, C. D., Arora, V., Friedlingstein, P., Bopp, L., Brovkin, V., Dunne, J., Graven, H.,
- Hoffman, F., Ilyina, T., John, J. G., Jung, M., Kawamiya, M., Koven, C., Pongratz, J.,
- Raddatz, T., Randerson, J. T., and Zaehle, S.: C4MIP-The Coupled Climate-Carbon Cycle
- Model Intercomparison Project: experimental protocol for CMIP6, Geosci Model Dev, 9,
- 832 2853-2880, 10.5194/gmd-9-2853-2016, 2016.
- Keller, D. P., Feng, E. Y., and Oschlies, A.: Potential climate engineering effectiveness and
- side effects during a high carbon dioxide-emission scenario, Nat Commun, 5, Artn 3304
- 835 10.1038/Ncomms4304, 2014.
- Keller, D. P., Lenton, A., Scott, V., Vaughan, N. E., Bauer, N., Ji, D., Jones, C., Kravitz, B.,
- Muri, H., and Zickfeld, K.: The Carbon Dioxide Removal Model Intercomparison Project
- 838 (CDR-MIP): Rationale and experimental design, Geoscientific Model Development
- 839 Discussions, 2017, accepted.
- 840 Kheshgi, H. S.: Sequestering Atmospheric Carbon-Dioxide by Increasing Ocean Alkalinity,
- 841 Energy, 20, 915-922, Doi 10.1016/0360-5442(95)00035-F, 1995.
- Kohler, P., Abrams, J. F., Volker, C., Hauck, J., and Wolf-Gladrow, D. A.: Geoengineering
- impact of open ocean dissolution of oliving on atmospheric CO2, surface ocean pH and
- marine biology, Environ Res Lett, 8, doi:10.1088/1748-9326/8/1/014009, 2013.
- Krasting, J. P., Dunne, J. P., Shevliakova, E., and Stouffer, R. J.: Trajectory sensitivity of the
- transient climate response to cumulative carbon emissions, Geophys Res Lett, 41, 2520-2527,
- 847 10.1002/2013gl059141, 2014.
- Le Quéré, C., Moriarty, R., Andrew, R. M., Peters, G. P., Ciais, P., Friedlingstein, P., Jones,
- 849 S. D., Sitch, S., Tans, P., Arneth, A., Boden, T. A., Bopp, L., Bozec, Y., Canadell, J. G.,
- 850 Chini, L. P., Chevallier, F., Cosca, C. E., Harris, I., Hoppema, M., Houghton, R. A., House, J.
- 851 I., Jain, A. K., Johannessen, T., Kato, E., Keeling, R. F., Kitidis, V., Klein Goldewijk, K.,
- Koven, C., Landa, C. S., Landschützer, P., Lenton, A., Lima, I. D., Marland, G., Mathis, J.
- 853 T., Metzl, N., Nojiri, Y., Olsen, A., Ono, T., Peng, S., Peters, W., Pfeil, B., Poulter, B.,
- Raupach, M. R., Regnier, P., Rödenbeck, C., Saito, S., Salisbury, J. E., Schuster, U.,
- Schwinger, J., Séférian, R., Segschneider, J., Steinhoff, T., Stocker, B. D., Sutton, A. J.,
- Takahashi, T., Tilbrook, B., van der Werf, G. R., Viovy, N., Wang, Y. P., Wanninkhof, R.,
- Wiltshire, A., and Zeng, N.: Global carbon budget 2014, 7, 47-85, 2015.
- Lenton, A., and Matear, R. J.: The role of the Southern Annular Mode (SAM) in Southern
- Ocean CO2 uptake, Global Biogeochemical Cycles, 21, doi: 10:1029/2006GB002714, 2007.
- Lenton, A., Tilbrook, B., Matear, R. J., Sasse, T. P., and Nojiri, Y.: Historical reconstruction
- of ocean acidification in the Australian region, Biogeosciences, 13, 1753-1765, 10.5194/bg-
- 862 13-1753-2016, 2016.
- Lenton, A., Keller, D. P., and Pfister, P.: How Will Earth Respond to Plans for Carbon
- 864 Dioxide Removal?, EOS, 98, 10.1029/2017EO068385, 2017.
- Lovelock, J. E., and Rapley, C. G.: Ocean pipes could help the Earth to cure itself, Nature,
- 866 449, 403-403, 10.1038/449403a, 2007.
- Lovenduski, N. S., Long, M. C., and Lindsay, K.: Natural variability in the surface ocean
- 868 <u>carbonate ion concentration, Biogeosciences, 12, 6321-6335, 10.5194/bg-12-6321-2015,</u>
- 869 2015.
- Mao, J., Phipps, S. J., Pitman, A. J., Wang, Y. P., Abramowitz, G., and Pak, B.: The CSIRO
- Mk3L climate system model v1.0 coupled to the CABLE land surface scheme v1.4b:
- evaluation of the control climatology, Geosci Model Dev, 4, 1115-1131, 10.5194/gmd-4-
- 873 1115-2011, 2011.
- Matear, R. J., and Hirst, A. C.: Long term changes in dissolved oxygen concentrations in the
- ocean caused by protracted global warming, Global Biogeochemical Cycles, 17, 1125, 2003.

- Matear, R. J., and Lenton, A.: Quantifying the impact of ocean acidification on our future
- 877 climate, Biogeosciences, 11, 3965-3983, Doi 10.5194/Bg-11-3965-2014, 2014.
- Mathesius, S., Hofmann, M., Caldeira, K., and Schellnhuber, H. J.: Long-term response of
- oceans to CO2 removal from the atmosphere, Nat Clim Change, 5, 1107-+,
- 880 10.1038/NCLIMATE2729, 2015.
- Mongin, M., Baird, M. E., Hadley, S., and Lenton, A.: Optimising reef-scale CO₂ removal by
- seaweed to buffer ocean acidification, Environ Res Lett, 11, Artn 034023
- 883 10.1088/1748-9326/11/3/034023, 2016a.
- Mongin, M., Baird, M. E., Tilbrook, B., Matear, R. J., Lenton, A., Herzfeld, M., Wild-Allen,
- K., Skerratt, J., Margvelashvili, N., Robson, B. J., Duarte, C. M., Gustafsson, M. S. M.,
- Ralph, P. J., and Steven, A. D. L.: The exposure of the Great Barrier Reef to ocean
- acidification, Nat Commun, 7, Artn 10732, 10.1038/Ncomms10732, 2016b.
- 888 10.1038/Ncomms10732, 2016b.
- Montserrat, F., Renforth, P., Hartmann, J., Leermakers, M., Knops, P., and Meysman, F. J.
- 890 R.: Olivine Dissolution in Seawater: Implications for CO2 Sequestration through Enhanced
- Weathering in Coastal Environments, Environmental science & technology, 51, 3960-3972,
- 892 10.1021/acs.est.605942, 2017.
- Mucci, A.: The Solubility of Calcite and Aragonite in Seawater at Various Salinities,
- Temperatures, and One Atmosphere Total Pressure, Am J Sci, 283, 780-799, 1983.
- Munday, P. L., Donelson, J. M., Dixson, D. L., and Endo, G. G. K.: Effects of ocean
- acidification on the early life history of a tropical marine fish, P Roy Soc B-Biol Sci, 276,
- 897 3275-3283, Doi 10.1098/Rspb.2009.0784, 2009.
- Munday, P. L., Dixson, D. L., McCormick, M. I., Meekan, M., Ferrari, M. C. O., and
- 899 Chivers, D. P.: Replenishment of fish populations is threatened by ocean acidification, P Natl
- 900 Acad Sci USA, 107, 12930-12934, Doi 10.1073/Pnas.1004519107, 2010.
- National Research Council: Climate Intervention: Carbon Dioxide and the Reliable
- 902 Sequestration of Carbon Washington D.C., US, 2015.
- 903 Orr, J. C., Fabry, V. J., Aumont, O., Bopp, L., Doney, S. C., Feely, R. A., Gnanadesikan, A.,
- Gruber, N., Ishida, A., Joos, F., Key, R. M., Lindsay, K., Maier-Reimer, E., Matear, R.,
- Monfray, P., Mouchet, A., Najjar, R. G., Plattner, G. K., Rodgers, K. B., Sabine, C. L.,
- Sarmiento, J. L., Schlitzer, R., Slater, R. D., Totterdell, I. J., Weirig, M. F., Yamanaka, Y.,
- and Yool, A.: Anthropogenic ocean acidification over the twenty-first century and its impact
- 908 on calcifying organisms, Nature, 437, 681-686, 2005.
- Phipps, S. J., Rotstayn, L. D., Gordon, H. B., Roberts, J. L., Hirst, A. C., and Budd, W. F.:
- The CSIRO Mk3L climate system model version 1.0-Part 2: Response to external forcings,
- 911 Geosci Model Dev, 5, 649-682, 10.5194/gmd-5-649-2012, 2012.
- Ragueneau, O., Treguer, P., Leynaert, A., Anderson, R. F., Brzezinski, M. A., DeMaster, D.
- J., Dugdale, R. C., Dymond, J., Fischer, G., Francois, R., Heinze, C., Maier-Reimer, E.,
- Martin-Jezequel, V., Nelson, D. M., and Queguiner, B.: A review of the Si cycle in the
- modem ocean: recent progress and missing gaps in the application of biogenic opal as a
- paleoproductivity proxy, Global Planet Change, 26, 317-365, Doi 10.1016/S0921-
- 917 8181(00)00052-7, 2000.
- Raven, J., Caldeira, K., Elderfield, H., Hoegh-Guldberg, O., Liss, P., and Riebesell, U.:
- Ocean acidification due to increasing atmospheric carbon dioxide. The Royal Society, Policy
- Document, London, UK, 2005.
- Renforth, P., and Henderson, G.: Assessing ocean alkalinity for carbon sequestration, Review
- 922 in Geophysics, 55, 10.1002/2016RG000533, 2017.
- Revelle, R., and Suess, H. E.: Carbon dioxide exchange between atmosphere and ocean and
- the question of an increase of atmospheric CO2 during the past decades, Tellus, 9, 18-27,
- 925 1957.

- Riebesell, U., Zondervan, I., Rost, B., Tortell, P. D., Zeebe, R. E., and Morel, F. M. M.:
- P27 Reduced calcification of marine plankton in response to increased atmospheric CO₂, Nature,
- 928 407, 364-367, 2000.
- Riebesell, U., Fabry, V. J., Hansson, L., and Gattuso, J.-P.: Guide to best practices for ocean
- 930 acidification research and data reporting, Publications Office of the European Union.,
- 931 Luxembourg, 260, 2010.
- P32 Rogelj, J., Meinshausen, M., and Knutti, R.: Global warming under old an new scenarios
- 933 using IPCC climate sensitivity range estimates, Nat Clim Change, 2, 248-253,
- 934 10.1038/NCLIMATE1385, 2012.
- P35 Rogeli, J., den Elzen, M., Hohne, N., Fransen, T., Fekete, H., Winkler, H., Chaeffer, R. S.,
- Ha, F., Riahi, K., and Meinshausen, M.: Paris Agreement climate proposals need a boost to
- 937 keep warming well below 2 degrees C, Nature, 534, 631-639, 10.1038/nature18307, 2016.
- 938 Sabine, C. L., Feely, R. A., Gruber, N., Bullister, J. L., Wanninkhof, R., Wong, C. S.,
- Wallace, D. W. R., Tilbrook, B., Millero, F. J., Peng, T.-H., Kozyr, A., Ono, T., and Rios, A.
- 940 F.: The Oceanic Sink for Anthropogenic CO₂, Science, 305, 367-371, 2004.
- 941 Scott, V., Haszeldine, R. S., Tett, S. F. B., and Oschlies, A.: Fossil fuels in a trillion tonne
- 942 world, Nat Clim Change, 5, 419-423, 10.1038/NCLIMATE2578, 2015.
- 943 Seneviratne, S. I., Corti, T., Davin, E. L., Hirschi, M., Jaeger, E. B., Lehner, I., Orlowsky, B.,
- 944 and Teuling, A. J.: Investigating soil moisture climate interactions in a changing climate: A
- 945 review, Earth-Science Reviews, 99, 125-161, 10.1016/j.earscirev.2010.02.004, 2010.
- 946 Sigman, D. M., and Boyle, E. A.: Glacial/interglacial variations in atmospheric carbon
- 947 dioxide, Nature, 407, 859-869, Doi 10.1038/35038000, 2000.
- 948 Smith, P., Davis, S. J., Creutzig, F., Fuss, S., Minx, J., Gabrielle, B., Kato, E., Jackson, R. B.,
- Cowie, A., Kriegler, E., van Vuuren, D. P., Rogelj, J., Ciais, P., Milne, J., Canadell, J. G.,
- McCollum, D., Peters, G., Andrew, R., Krey, V., Shrestha, G., Friedlingstein, P., Gasser, T.,
- 951 Grubler, A., Heidug, W. K., Jonas, M., Jones, C. D., Kraxner, F., Littleton, E., Lowe, J.,
- Moreira, J. R., Nakicenovic, N., Obersteiner, M., Patwardhan, A., Rogner, M., Rubin, E.,
- Sharifi, A., Torvanger, A., Yamagata, Y., Edmonds, J., and Cho, Y.: Biophysical and
- economic limits to negative CO2 emissions, Nat Clim Change, 6, 42-50,
- 955 10.1038/NCLIMATE2870, 2016.
- Society, T. R.: Geoengineering the Climate System: Sceonce Goverence and Uncertainity,
- 957 London, UK, 2009.
- P58 Taylor, K. E., Stouffer, R. J., and Meehl, G. A.: An Overview of Cmip5CMIP5 and the
- Experiment Design, B Am Meteorol Soc, 93, 485-498, 10.1175/BAMS-D-11-00094.1, 2012.
- 960 Trull, T. W., Rintoul, S. R., Hadfield, M., and Abraham, E. R.: Circulation and seasonal
- evolution of polar waters south of Australia: Implications for iron fertilizations for the
- 962 Southern Ocean, Deep-Sea Research II, 48, 2439-2466, 2001.
- United Nations Framework on Climate Change: Adoption of the Paris Agreement, 21st
- Conference of the Parties. 2015.
- Wang, Y. P., Law, R. M., and Pak, B.: A global model of carbon, nitrogen and phosphorus
- cycles for the terrestrial biosphere, Biogeosciences, 7, 2261-2282, 10.5194/bg-7-2261-2010,
- 967 2010.
- Yamamoto, A., Kawamiya, M., Ishida, A., Yamanaka, Y., and Watanabe, S.: Impact of rapid
- sea-ice reduction in the Arctic Ocean on the rate of ocean acidification, Biogeosciences, 9,
- 970 2365-2375, 10.5194/bg-9-2365-2012, 2012.
- Yamanaka, Y., and Tajika, E.: The role of vertical fluxes of particulate organic material and
- 972 calcite in the oceanic carbon cycle: Studies using a ocean biogeochemical general; circulation
- model, Global Biogeochemical Cycles, 10, 361-382, 1996.

Yool, A., Popova, E. E., and Coward, A. C.: Future change in ocean productivity: Is the

975 Arctic the new Atlantic?, J Geophys Res-Oceans, 120, 7771-7790, 10.1002/2015JC011167,

976 2015.

P77 Zeebe, R. E.: History of Seawater Carbonate Chemistry, Atmospheric CO₂, and Ocean

978 Acidification, Annu Rev Earth Pl Sc, 40, 141-165, 10.1146/annurev-earth-042711-105521,

979 2012.

P80 Zhang, Q., Wang, Y. P., Matear, R. J., Pitman, A. J., and Dai, Y. J.: Nitrogen and

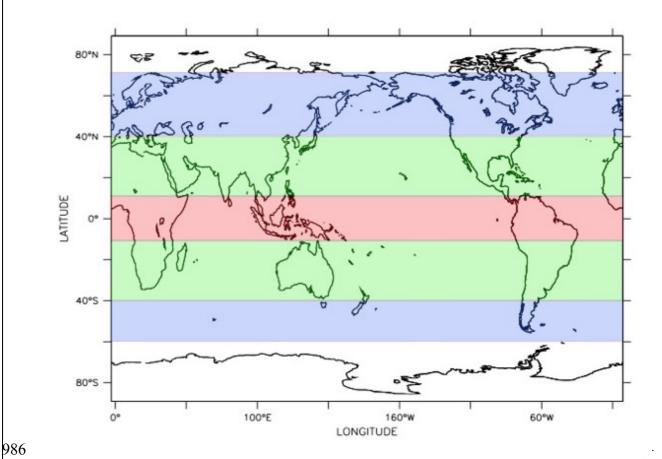
phosphorous limitations significantly reduce future allowable CO2 emissions, Geophys Res

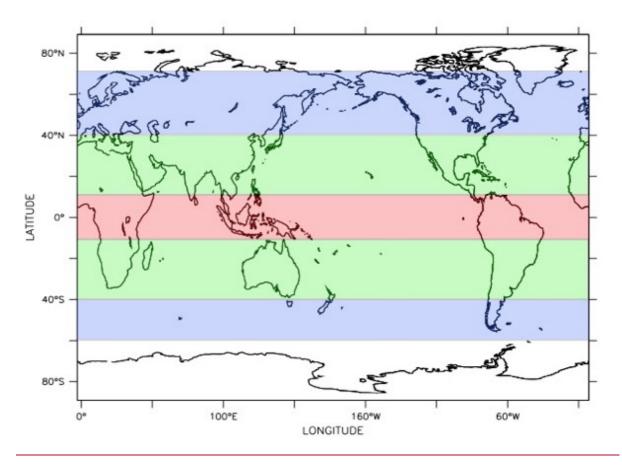
982 Lett, 41, 632-637, 10.1002/2013GL058352, 2014a.

P83 Zhang, Z. G., Wang, W., and Qiu, B.: Oceanic mass transport by mesoscale eddies, Science,

984 345, 322-324, 10.1126/science.1252418, 2014b-

985





993

994 Figure 1 Ocean regions used for Alkalinity Injection in the period 2020-2100, the blue

995 denotes the subpolar regions (AOA SP), the green regions represent the subtropical gy

denotes the subpolar regions (AOA_SP), the green regions represent the subtropical gyres (AOA_ST), red the tropical ocean (AOA_T), and all <u>colored_coloured</u> regions combined the global alkalinity injection (AOA_G). Note that the ocean regions not <u>colored_coloured</u> represent the seasonal sea-ice, where no alkalinity was added in the simulation.

Page 1 of 40

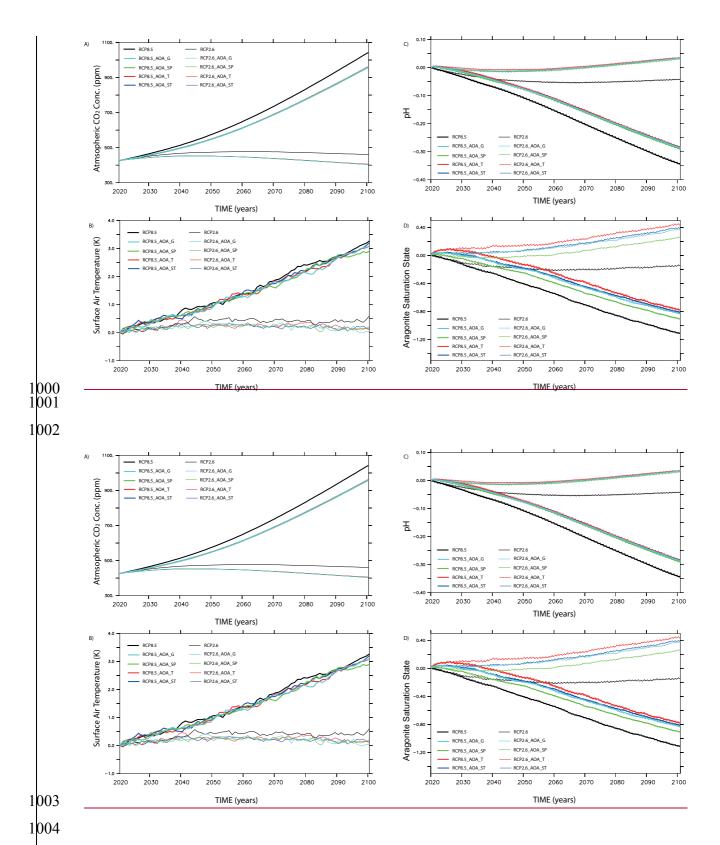
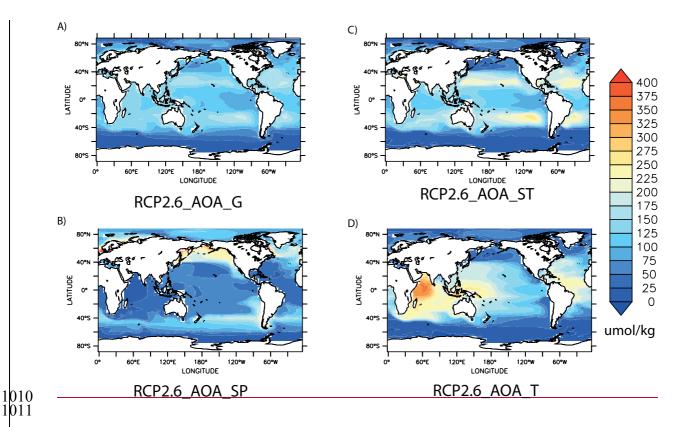


Figure 2 The global mean changes in: Atmospheric CO2 concentration (Aa), Surface Air Temperature (SAT; Bb), surface ocean pH (Cc) and Aragonite Saturation State (Dd) for high (RCP8.5) and low emissions (RCP2.6) with global and regional AOA in the period 2020-2100.



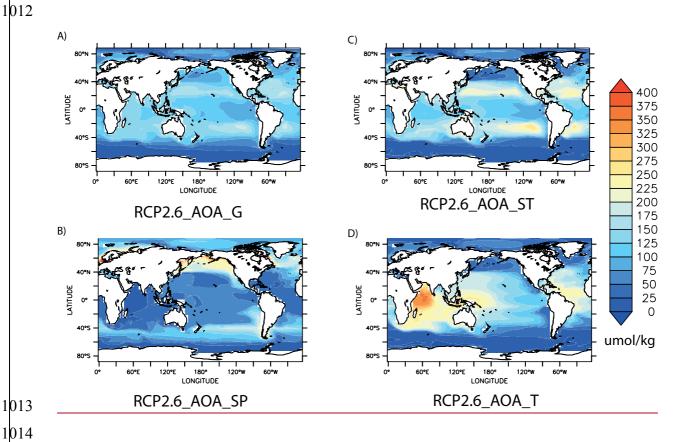


Figure 3 The spatial map of <u>relativethe</u> increase in surface alkalinity in 2090 (mean; 2081-1016 2100) associated with global and regional AOA under RCP2.6-<u>relative to RCP2.6 with no</u> 1017 <u>AOA.</u> Units are μmol/kg.

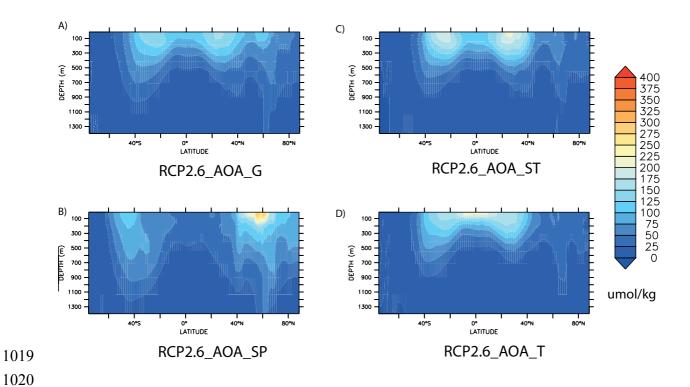


Figure 4 The zonal mean-relative changes in alkalinity in the interior ocean associated with global and regional AOA under RCP8.5 in 2090 (mean÷; 2081-2100)-) relative to RCP8.5 with no AOA. Units are μmol/kg.

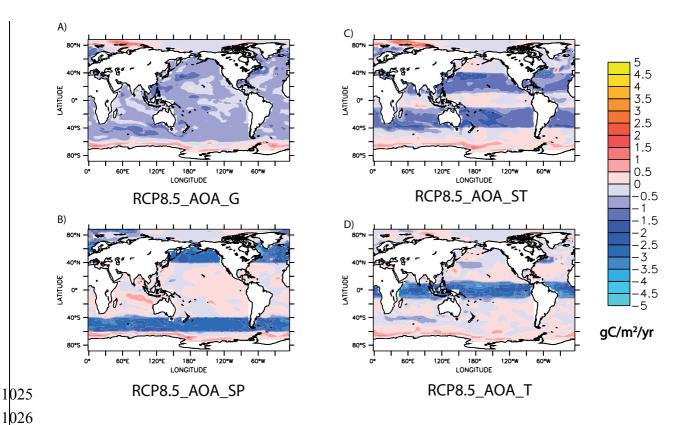


Figure 5 The spatial map of relative changes in ocean carbon uptake in 2090 (mean; 2081-2100) associated with global and regional AOA under RCP8.5. Units are gC/m²/yr

1028



1034

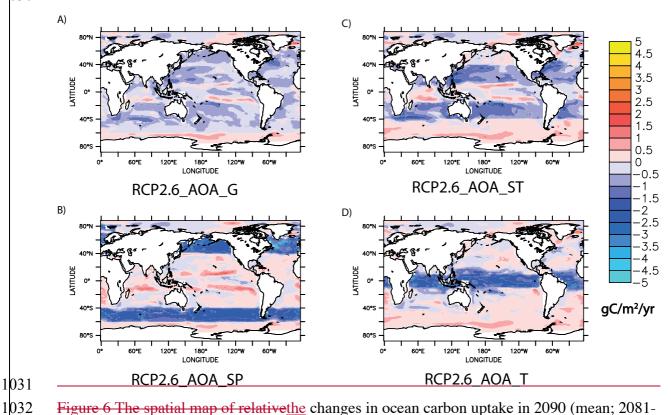


Figure 6 The spatial map of relativethe changes in ocean carbon uptake in 2090 (mean; 2081-2100) associated with global and regional AOA under RCP2.6.RCP8.5, relative to RCP8.5 with no AOA. Units are gC/m²/yr.

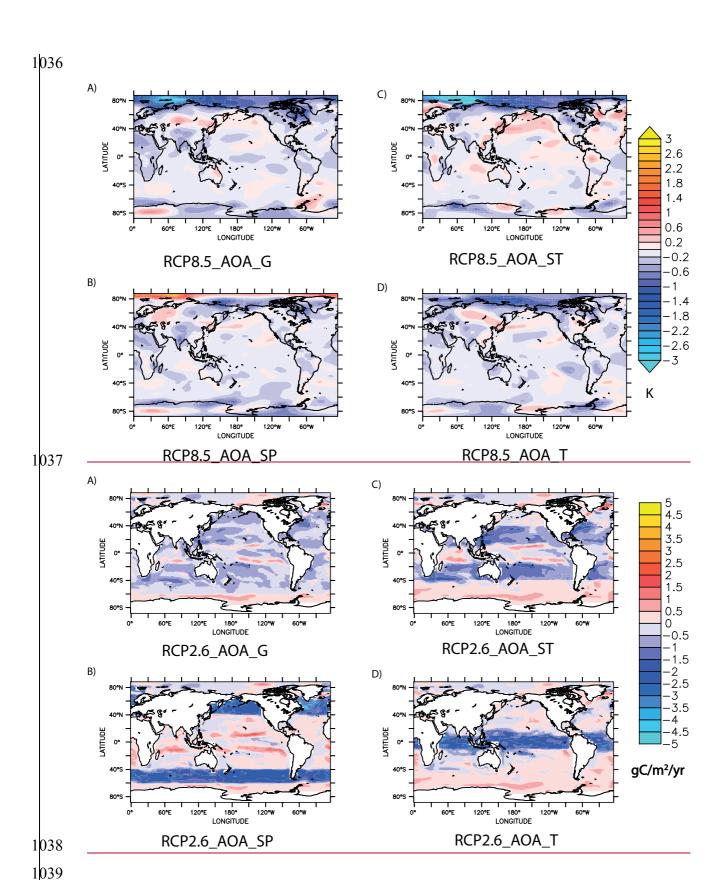
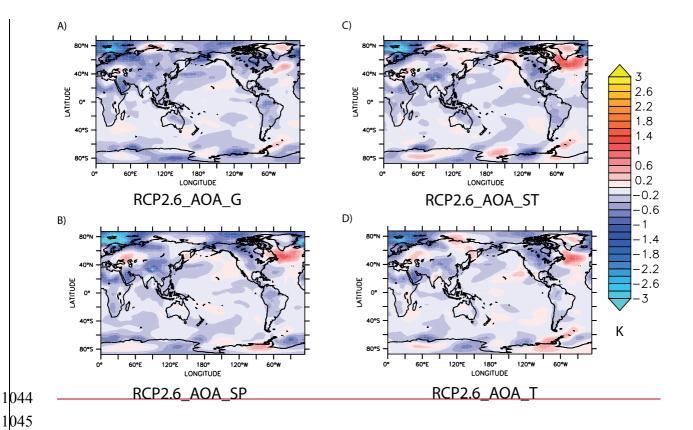


Figure 76 The spatial map of relativethe changes in surface air temperatureocean carbon uptake in 2090 (mean; 2081-2100) associated with global and regional AOA under RCP8.5.RCP2.6, relative to the RCP2.6 with no AOA. Units are KgC/m²/yr.

1041



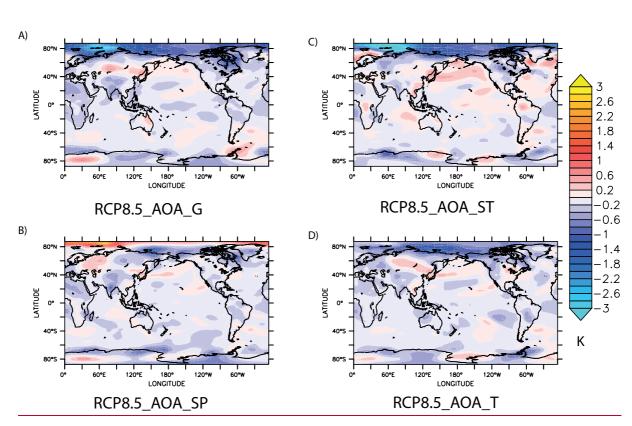


Figure <u>87</u> The spatial map of <u>relativethe</u> changes in surface air temperature 2090 (mean; 2081-2100) associated with global and regional AOA under <u>RCP2.6.RCP8.5</u>, <u>relative to RCP8.5</u> with no AOA. Units are K.

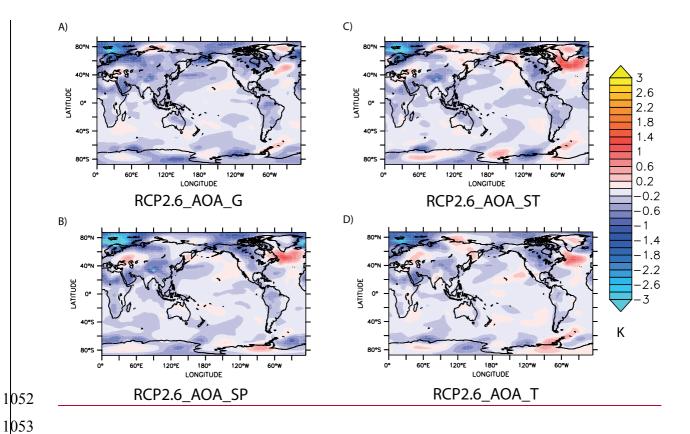


Figure 8 The spatial map of the changes in surface air temperature 2090 (mean; 2081-2100) associated with global and regional AOA under RCP2.6, relative to the RCP2.6 with no AOA. Units are K.

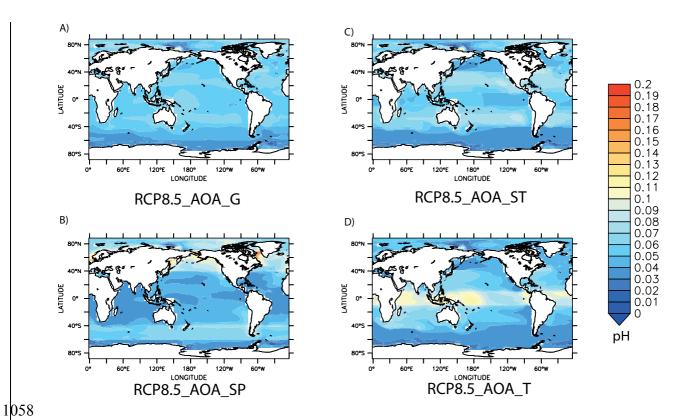
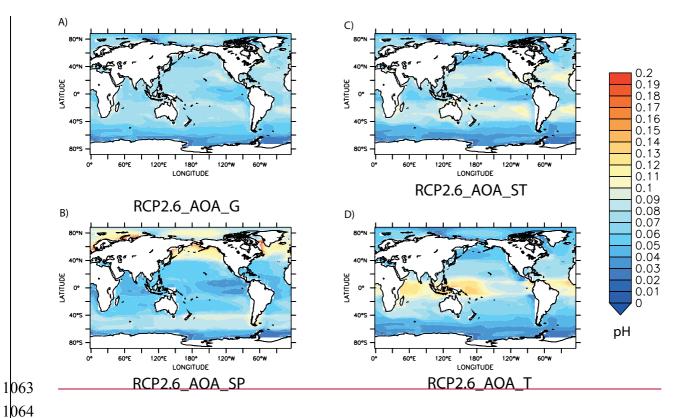


Figure 9 The spatial map of the relative changes in pH in 2090 (mean; 2081-2100) associated with global and regional AOA under RCP8.5, relative to RCP8.5 with no AOA.



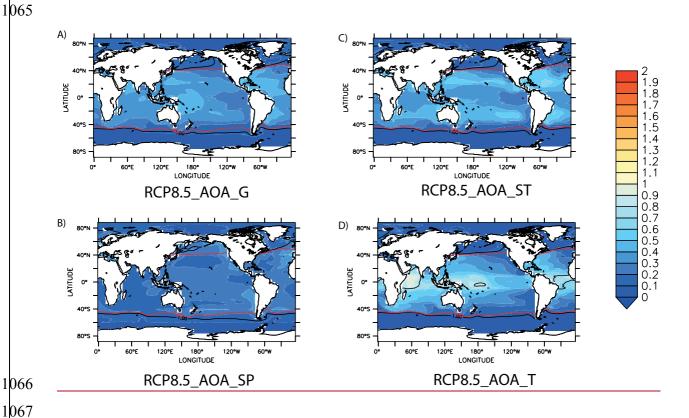


Figure 10 The spatial map of the relative changes in pH in 2090 (mean; 2081-2100) associated with global and regional AOA under RCP2.6.

1069

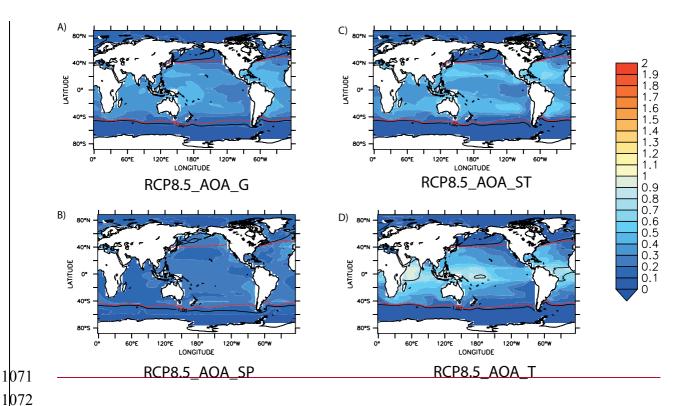


Figure 11 The spatial map of the relative differences in surface aragonite saturation state in 2090 (mean 2081-2100), associated with global and regional AOA under RCP8.5. Contoured on each map are the values of aragonite saturation state of 1 and 3, please see the text for more explanation. The red contours represent RCP8.5 and black AOA for each experiment

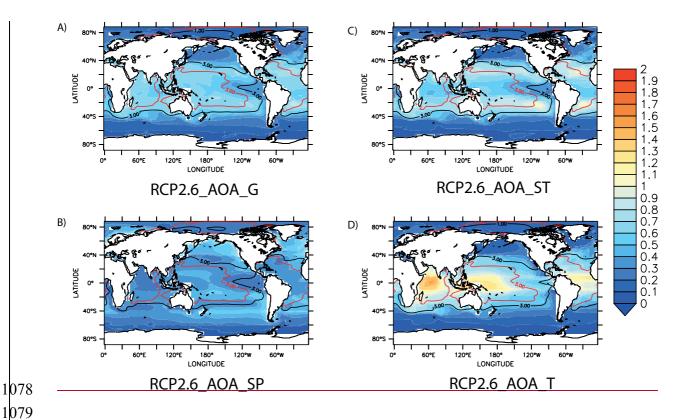


Figure 12 The spatial map of the relative differences in surface aragonite saturation state in 2090 (mean 2081-2100), associated with global and regional AOA under RCP2.6. Contoured on each map are the values of aragonite saturation state of 1 and 3, please see the text for more explanation. The red contours represent RCP2.6 and black AOA for each experiment

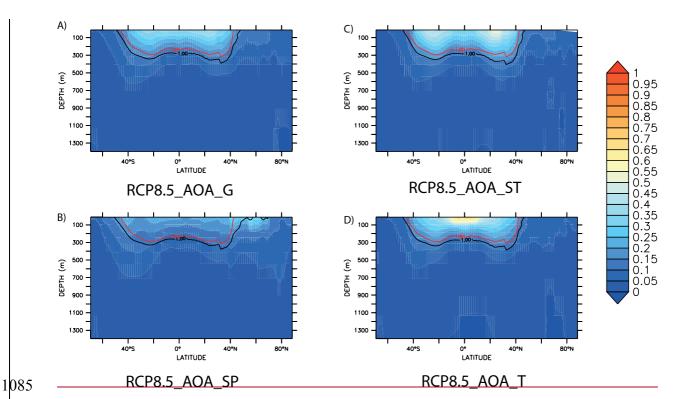


Figure 13 The relative zonal mean differences in aragonite saturation state in 2090 (mean; 2081-2100), associated with global and regional AOA under RCP8.5-, relative to RCP8.5 with no AOA. Contoured on each map are the values of aragonite saturation state of 1 and 3; please see the text for more explanation. The red contours represent RCP8.5 without AOA and the black contours represent RCP8.5 with AOA for each experiment.

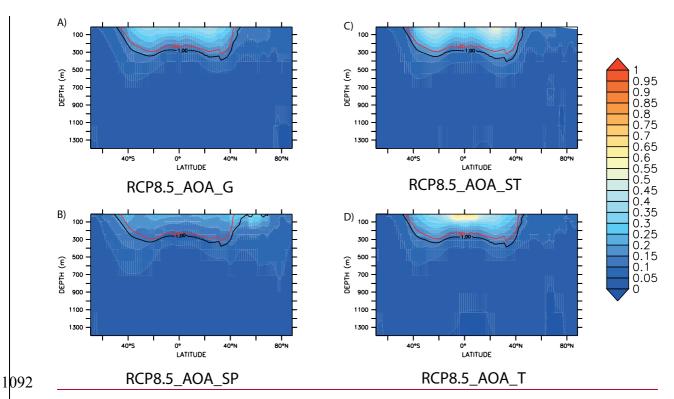
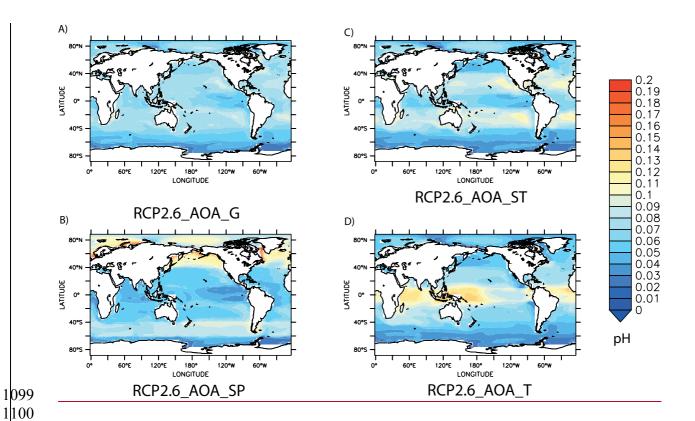


Figure 11 The zonal mean differences in aragonite saturation state in 2090 (mean; 2081-2100), associated with global and regional AOA under RCP8.5, relative to RCP8.5 with no AOA. Contoured on each map are the values of aragonite saturation state of 1; please see the text for more explanation. The red contours represent RCP8.5 without AOA and the black contours represent RCP8.5 with AOA for each experiment-



<u>Figure 12 The spatial map of the changes in pH in 2090 (mean; 2081-2100) associated with global and regional AOA under RCP2.6, relative to RCP2.6 with no AOA.</u>

1 102

1103

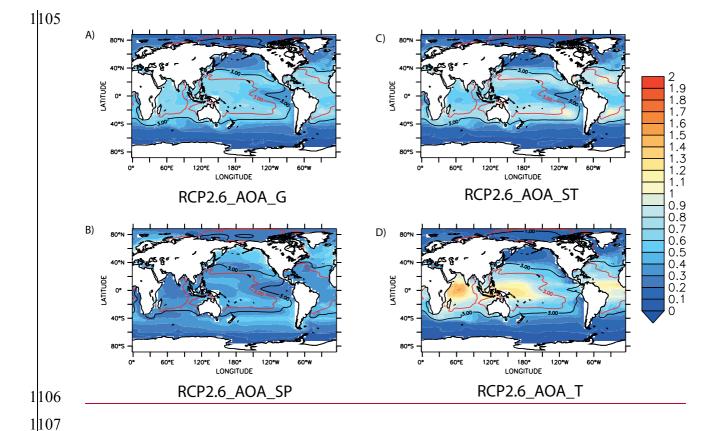


Figure 13 The spatial map of the differences in surface aragonite saturation state in 2090 (mean; 2081-2100), associated with global and regional AOA under RCP2.6, relative to RCP2.6 with no AOA. Contoured on each map are the values of aragonite saturation state of 1 and 3; please see the text for more explanation. The red contours represent RCP2.6 without AOA and the black contours represent RCP2.6 with AOA for each experiment

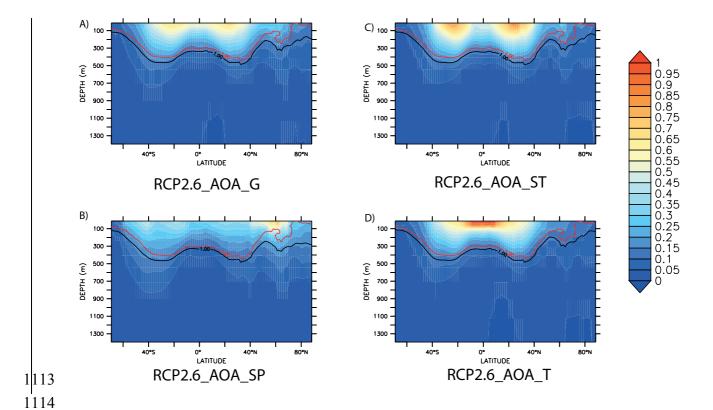


Figure 14 The relative zonal mean differences in aragonite saturation state in 2090 (mean; 2081-2100), associated with global and regional AOA under RCP2.6, relative to RCP2.6 with no AOA. Contoured on each map are the values of aragonite saturation state of 1; please see the text for more explanation. The red contours represent RCP2.6 without AOA and the black contours represent RCP2.6 with AOA for each experiment.

1|120